

SAND CULTURES —POPPIES GROWING IN STERILE SAND

A Watered with normal culture solution B Watered with a solution incl. nitrates (See Chapter viii) (From a photograph)

PRACTICAL PLANT PHYSIOLOGY

BY

F. FREDERICK KEEBLE, Sc D

PROFESSOR OF BOTANY AND DEAN OF THE FACULTY OF SCIENCE
UNIVERSITY COLLEGE, READING
AUTHOR OF "PLANT ANIMALS"

ASSISTED BY

M C RAYNER, BSc

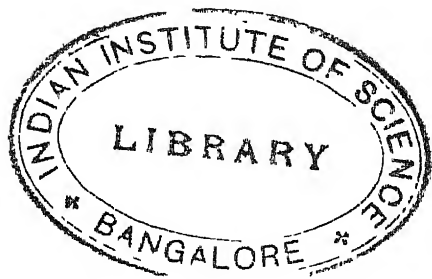
LECTURER IN BOTANY IN UNIVERSITY COLLEGE, READING



LONDON

G BELL AND SONS, LTD

1919



2829

1911

1912

GLASGOW PRINTED AT THE UNIVERSITY PRESS
BY ROBERT MACLEHOSE AND CO. LTD.

PREFACE

THE purpose of this book is to provide students and teachers with an outline of the experimental investigations on which our knowledge of the physiology of plants is based.

It would have been easier to present the most attractive of the facts and theories of this branch of science than to make the attempt, of which this book is the outcome, to provide a practical text-book which should serve as an educational instrument and as a stimulus to independent observations.

The student, however, cannot grasp the facts of Natural Science by reading books on the subject. Though the man of mature mind, trained in the art of sifting and grouping facts, and in the habit of seizing essential truths, may gain a useful acquaintance with Natural Science by the perusal of text-books, the student cannot. He needs a method of education which serves, not only to furnish, but also to train the mind. In other words, the student requires both information and discipline. He should be told less, and find out more. It cannot be disputed that the element of intellectual discipline is ignored too much by modern educational methods, nor that, as a result, students suffer from lack of training.

In science, in particular, the teacher is too much concerned with rendering the subject attractive and too little with making it to serve the end of developing the reasoning and imaginative faculties of his pupils.

The method of teaching Natural Science by means of lectures and a ritual of practical work associated therewith is fundamentally wrong. It appeals to the memory but

sterilizes the imagination. It preaches comfortable to the student, who ought, instead of being submit scientific sermons, to be experiencing the process, at pleasurable and painful, of teaching himself.

In the course of the lecture hour, the teacher covers large area of ground which is a terra incognita to student. From the fulness of his experience, the teacher clears the ground of all difficulties which beset the of the scholar, with the result that the latter, though may, or may not, have enjoyed the personally-constructed tour, has learned nothing of the art of exploration. The student would be unable, indeed, to find his way over the same ground if left to take the journey alone.

Subjected to this lecture-method of instruction, the student becomes the repository of the second-hand. When he is told to look for, he sees with the teacher's eye; the method is successful and the student docile, the student passes his examination with distinction, but remains ignorant of the importance of initiative. The less the student takes and expresses—often by idleness—a distaste to science, preferring, instinctively, some study more calculated to train the mind or less certain to overtax the memory.

In either case, the lecture-method, which dates from time when books were scarce, or dear, or bad, accomplishes little in comparison with what might be achieved if the ground were traversed during the lecture-excursion at the student, instead of being told things, was encouraged to observe and think for himself.

Until the present, deplorable method is replaced by a better, Natural Science, which must come in time to an essential part of all educational curricula, must stand to the reproach of being of less value, as a supplemental training, than Mathematics or Classics. The reproach is not deserved. Without a knowledge of Natural Science, men and women remain blind to half the beauty and meaning of life; without it, they are ignorant of modes whereby all manner of pressing problems, social, political and other, ought to be attacked if they are to be solved.

If reform is to be effected in the teaching of Natural

Science, it must be by making the subject less an affair of information and more a matter for thought. The logical argument which runs through the Natural Sciences, cohering the innumerable facts to a whole, must be displayed, and the student must find, in his scientific studies, means of developing the higher functions of his mind, namely, the powers of thinking rightly and imagining freely.

The proper method of teaching Natural Science is as follows. Very young children should be rendered familiar with large numbers of natural objects, including, of course, plants and animals. The purpose of this preliminary course is to make the child take cognisance of natural objects and phenomena, of flowers and their names, of fruits and seeds, of buds and leaves, of herbs and trees. In short, confining our remarks to the biological side of this course, the child should learn by means of it to see and watch and care for some of the myriad living things of field, hedgerow and garden.

This is the Nature Study course, in which no attempt should be made to force explanations or to develop the reasoning faculty of the child, the object being to furnish the mind.

Having achieved this end, Nature Study has exhausted its educational service. It is as useless in later stages as it is valuable in the early stage of education.

The Nature Study course should be succeeded in the school by one on elementary physics and chemistry. Such a course is, as a fact, followed in many schools. In it, the properties of matter are first investigated experimentally, and, later in the course, the relations of facts or phenomena, one with another, are studied. During this course, the child begins to reason and to imagine scientifically.

Then should follow the third part of the course in Natural Science, which should be of a biological nature.

It is a great misfortune that this third part but rarely finds a place in the school curriculum. The scholars continue too long with the more mechanical Natural Sciences, Physics and Chemistry, and leave school only too often with no knowledge of Biology.

This is the case generally with boys, though, not infre-

quently, girls follow a course of morphological Botany instead of one on Physics and Chemistry.

In either case, the results are bad. Neither the boys nor the girls know anything of the modes of life of plants or animals, the boys, because they have not studied the subject, the girls, because, without a knowledge of the elements of Physics and Chemistry, they cannot understand it. The remedy is to require the scholars, boys and girls alike, to follow the elementary course in Physics and Chemistry, and then to proceed to one on the elements of Biology.

The object of this book is to provide such a biologic course, that is, one suitable for the higher classes in schools and for the first year class in the University.

It is true that the book deals only with the physiology of plants, but it is true also that anyone who is familiar with the facts of this science knows not a little of the essentials of animal-physiology.

By following the course, the training in reasoning and in imagination, begun during the study of Physics and Chemistry, is extended. The scholar who has passed through the three courses outlined above possesses confident knowledge of the most important facts of Natural Science. He has also a trained mind.

This book is useless as a reading-book, but, if its intention has been fulfilled, it should prove a serviceable tool wherewith the student may dig out for himself the fundamental truths of the science.

To do this means, of course, hard work on his part; but once the student has learned that one of the chief pleasures of life lies in hard, intellectual work—just another great pleasure lies in the hard, physical work which he bestows on his games—he will, it is to be hoped, prefer this course to the easier, sitting-down method wherein he plays the part of a recipient of knowledge which he knows neither the origin nor the mode of getting.

If the book is to serve the end for which it has been designed, it must be used, like any other tool, in a proper way.

The teacher should decide on the number of chapters which are to be dealt with in any special course. I

should then divide the chapters into lessons, each of which should be accomplished in the hours which may be devoted to the subject in the course of the week. Having explained that the need for making one experiment follows from the conclusions drawn from its predecessor, the teacher should point out to the students that they should pay no attention to the results of any experiment recorded in the text-book until they themselves have obtained records.

Before starting an experiment, the student should read the instructions both in the text and in the appendix, so that he may know all the details necessary for the setting-up and use of the apparatus. Having done this, the student should make, in each case in which the experiment admits of it, a working drawing of the apparatus which he proposes to construct, and he should write a brief prediction of the results which he expects to obtain by the performance of the experiment. When the apparatus is made, and before he begins to use it, the student should examine it critically with the object of making sure in advance that the apparatus can do the work required of it. Should the apparatus fail to work properly, the student must submit it again to critical examination in order to discover the cause of the failure.

Nothing is more valuable for the education of students than learning to construct apparatus, making it work, discovering the reasons of its failure and devising means for overcoming its defects.

Instead of expending his energies in lecturing, the teacher should devote himself to obtaining material suitable for the experiments, assisting students in overcoming difficulties and securing permanent records of the experiments. These records should be deposited in the physiological museum, to which frequent reference is made in the text.

When a lesson is completed, the teacher should discuss the results and conclusions with the students, and point out the bearing of these results on other parts of the subject.

In conclusion, though the author must be held responsible for any errors which are contained in the book, and for the opinions expressed in the preface, he acknowledges

with pleasure, what is acknowledged already in the page, that, in writing this book, he has had the advantage of the assistance of his colleague, Miss M. C. Ray.

Nearly all of the illustrations have been made expressly for this volume. Some are reproductions from photographs, others are due to the skill and kindness of Dorothea Richardson, from whose original drawings have been reproduced.

FREDERICK KEI

University College, Reading,
December, 1910



CONTENTS

CHAPTER I		PAGE
INTRODUCTORY		I
The problems of plant physiology and the method by which they are to be solved The scientific method Classification of physiological problems		
CHAPTER II		
GERMINATION		7
The mode of germination of seeds the parts of the seed and seedling the resting and active states of seeds the resisting powers of resting seeds germination capacity the visible order of events in germination The nature and function of cotyledons and of endosperm adaptation in plants large seeds and small seeds		
CHAPTER III		
THE FOOD-MATERIALS OF SEEDS		29
The nature and chemical properties of the food substances contained in the cotyledons and endosperm of seeds		
CHAPTER IV		
CHANGES DURING GERMINATION		44
The changes undergone by the reserve food materials of the seed during germination the mode of passage of food-		

materials from the place of storage (endosperm or cotyledons) to the place of consumption (the growing embryo).

CHAPTER V

NUTRITION

The meaning of the term Nutrition the use which the plant makes of food substances The germinating seed considered as a machine The source of the power which drives the machine and the conditions under which it works

CHAPTER VI

THE WORK OF ROOTS

The seedling as an independent plant the lowest forms of plants and animals and the lines followed in the evolution of the higher plants and animals The distinguishing characters of root and shoot systems The mode of growth of the root the functions of its parts the root hairs, the absorbent organs of the root

CHAPTER VII

OSMOSIS AND OSMOTIC PRESSURE

The way in which water is absorbed by root hairs and other cells Osmosis and osmotic pressure The plant cell as an osmotic apparatus.

CHAPTER VIII

THE SOIL IN RELATION TO PLANT-LIFE

The substances taken up by the roots of plants The composition of plant ash Water and sand cultures The soil in relation to plant life The origin of soils their physical, chemical, and biological properties

CONTENTS

xv

CHAPTER IX

TRANSPIRATION	PAGE 143
-----------------------	-------------

The absorption and loss of water by the plant. The water-requirements of various types of plants —hygrophytes and xerophytes. The process of the transpiration of water by the leaves. the structure of the leaf in relation to this process. the part played by stomata. the opening and closing of stomata and the conditions under which these movements occur. Apparatus for measuring rate of transpiration—(Potometer)

CHAPTER X

THE TRANSPIRATION CURRENT	162
-----------------------------------	-----

The passage of water from root to leaves. the channels followed by the transpiration current. water conducting wood and skeletal wood. The causes of the ascent of water. Phenomena connected with the absorption of water. root pressure. bleeding. excretion of water. water-pores (hydathodes)

CHAPTER XI

PHOTOSYNTHESIS	176
------------------------	-----

The origin of the carbon compounds contained in plants. The raw materials from which the plant constructs these compounds. The part played by chlorophyll grains (chloroplasts) in the manufacturing process. the energy by which the process is carried on. The passage of carbohydrates from the leaves to other parts of the plant. The synthesis of organic nitrogen compounds in the plant.

CHAPTER XII

PLANT SENSITIVENESS	197
-----------------------------	-----

The modes of response of plants to stimulation. Irritability. The reflex actions of plants. Tropisms (Geotropism, Photo-

tropism, etc.) Morphogenetic responses The component parts of a reflex action — perception, excitation, transmission of nervous impulses, excitation and response of the motor region

APPENDIX

BIBLIOGRAPHY

INDEX

FRONTISPIECE WITH TEXT-FIGURES 1-32



CHAPTER I

THE problems of plant physiology and the method by which they
to be solved The scientific method Classification of physiolo-
problems

A SEED sown under suitable conditions germinates, gives rise to a seedling. The seedling grows, puts forth leaves and branches, drives its roots further and further into the soil, and becomes a mature plant. Presently and in due season the flowers appear, endure in their delicate beauty for a while, and, their work being done, fade away. The stalks which bore the flowers now support the swell of fruits within which the seeds are ripening. When the seeds are set, the fruit bursts open, scattering them wide and wide, or, falling to the ground and rotting, sows seeds near the parent plant. Such are the more striking episodes in the life of a flowering plant.

This regular sequence of events—the germination of a sown seed, the formation of roots and branches, the growth of flowers, the setting of the seeds and the ripening of the fruits—seems so natural that we are apt to accept it as needing no explanation. But when we begin to observe the several processes more closely, to reflect upon them, to compare one plant with another with respect to them, we find ourselves asking all sorts of questions. Why is it that the chickweed of the hedgerow runs its course from seedling to fruiting stage in less than one brief season, while the foxglove does not reach the flowering stage till its second year? By what gymnastic exercises does the seedling extricate itself so neatly from its seed coat? How is it that the plantain on the lawn hugs the ground so closely as to escape almost uninjured the knives of the cutting machine?

Whence come the power and the material by means of which the giant oak with gnarled trunk and spreading branches forms itself from an acorn? How is it that matter which way up we plant a bean seed in the soil, root of the seedling turns downward and burrows in earth, whilst the stem twists itself so that it comes to grow upward into the air? From what source does the plant obtain the sugar to which it owes its sweetness or the perfume with which it scents the air?

When once we begin to interest ourselves in plants we find that the problems which they suggest are as varied as they are numerous, and we realise that in every park, hedgerow, field, or garden, all sorts of strange events are happening.

Could we but find answers to the questions which plants suggest to us, we should be in possession of a great body of knowledge concerning their life and work. In other words, we should have taught ourselves not a little of the science of plant physiology.

Hence the most pressing of our problems is, how are we to set about obtaining answers to any of the questions which arise in our minds when we observe living plants? Curiosity suggests the problems, how does science solve them?

In order to find out the method of scientific discovery, let us fix our attention on some particular phenomenon exhibited by a plant, and consider how we may assess its significance.

The phenomenon which we will choose for investigation is the origin of the drops of water which appear on the leaves of certain plants, such, for example, as the *Impatiens*.

I. We sow oats in ordinary soil in two pots, and, when the leaves of the seedlings are four or five inches in height, we may find, on examining them in the early morning, that near the apex of each leaf is a shining drop of water, looking like a dew-drop (Fig. 1).

We want to discover whence the water-drops come. Though we were to sit up all night watching the plants, we should obtain no solution of the problem: all we should

† The numerals, in heavy type, which occur throughout the book in the experiments which are to be performed, see Preface, p. vi.

see is that the drops, when first formed, are small, and that they may increase in size very rapidly. Observation, therefore, though it provides us with scientific puzzles, does not

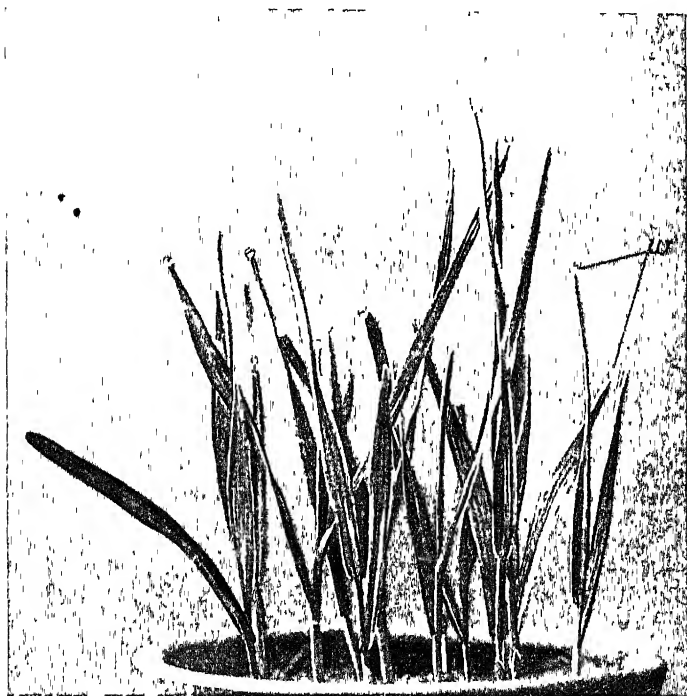


FIG. 1. OAT SEEDLINGS

Drop of water (w), excreted from water pore (hydathode) in the tips of the leaves
From a Photograph

necessarily help us to solve them. Observation, careful and continued observation of the living plant, is essential for the study of plant physiology, but something besides observation is wanted now.

Evidently all we can do is to make a guess as to the origin of the water-drops. Confronted with the problem,

we call imagination to our assistance, and by its aid give us the answer. As the result of guessing, we suggest that the water may be dew formed from water-vapour in the air. Next we ask ourselves, suppose the guess is wrong, what then? At once common-sense makes answer, if the water does not come from the air, it must come from the plant itself. Having exhausted our guessing powers, we proceed to look coldly at the alternative suggestions, and to revolve them in the light of common-sense. In this case, common sense admits that either guess may be right. But guesses cannot both be right. Therefore we must discover some way of deciding between them. If the problem were one of a kind with which we are more familiar, for example as to the height of a friend, and if two people guessed differently with respect to this, we should not hesitate to our method of verification. We should stand the friend against the wall and measure him. That is, we should put the guesses to the test of experiment. Whereas the amount of discussion would determine the correctness of the guesses, a yard measure properly used would settle the matter in a minute.

In like manner, to solve the problem of the origin of the water-drops, we submit it to the test of experiment. But how is this to be done? Once again we must appeal to imagination and common-sense. We must use the faculties conjointly in order to invent an experimental method. In our particular problem, it is easy enough to devise a method. We know that plants take up water from the soil, and so we argue thus: if the drops of water on leaves come from the air, they may make their appearance as readily on the unwatered as on the watered plants, if the drops come from the plants themselves, it probably matters fundamentally whether the plants can get much or little water. Thus we arrive at our method of experiment. Water one pot thoroughly, withhold water from the other, and examine the plants on successive mornings. When we do this, we find that the water-drops are plentiful on the leaves of the watered seedlings, are absent from, or, at all events, fewer on the others.

To complete the proof we devise a further experiment.

2 For example, give water to the previously

watered pot, stand it under a bell jar, and observe that in five or ten minutes drops of water appear on the tips of the leaves. Therefore we conclude that the water which appears on the leaves comes from the plant and not from the air.

But in solving the particular problem, we have discovered also the answer to the general problem—how to set about getting replies to special questions? The answer is—by using exactly the same method as that which we have just employed. There is indeed no other way. It is called the scientific method, and involves, as we have learned, processes of guessing, reasoning, and trying.

Thus the processes involved in the use of the method are as follows:

1. The guessing process in which the imagination is invited to suggest possible answers to the problem under investigation. Our guesses may be as wild as we like to make them. The more the imagination is allowed to run riot, the more likely are we to open up new paths for investigation. Indeed, it is no exaggeration to say that the greatest discoveries are the outcome of the wildest guesses.

2. The judging process, in which common-sense assumes the part of advisor, recommending this or that guess as more likely to prove true, and rejecting any guess which runs counter to established truth.

3. The testing process, which consists in the devising and execution of experiments calculated to demonstrate the truth or falsity of the guess, or, as it may be called, the hypothesis, which gains the approval of common-sense.

4. The summing-up process, by which we decide, whether the evidence provided by the results of the experiments is convincing or not.

If the evidence is absolutely conclusive in favour of our hypothesis, we speak of that hypothesis as a fact, if the evidence is inconclusive, we may yet continue, for want of a better, to hold the hypothesis and to use it in our arguments, though in doing this we have to be extremely cautious, and to remember that our hypothesis is 'not proven'.

Hence to study a science aright is not to become a narrow

specialist, but to develop all the highest faculties of the mind. This scientific method is not peculiar to plant physiology—it is the method employed in all the sciences and by its use all the knowledge of nature which we possess has been obtained.

All that remains to be done in this introductory chapter is to classify our problems, that is, to arrange those of like nature in groups, and the groups in convenient order. In doing this we will make an assumption which may not, at first sight, seem very probable, but which will be of great service to us. Whether the assumption is true or false we shall discover as we proceed with our investigations. We assume that the life of a plant is not different in essentials from that of an animal or from that of man himself. Unless the assumption is wholly false, and, in that case, we shall soon discover our mistake, it will be of great assistance to us in the otherwise puzzling problem of the arrangement of our questions. For we know, without the aid of science and from our common experience, a good deal about our own life-processes. We know, for example, that we feed and that without food of certain kinds we cannot live. We know that we breathe, and that we cannot exist for more than a few minutes without air. We know also that we like animals in general, move, and that some movements, as, for example, getting up in the morning, depend on an effort of will, whilst other movements, for example, the beating of the heart, are independent of consciousness. We know also that animals and plants grow, give birth to young, and ultimately die.

Hence we arrive at the following classification of the problems of plant physiology:

- (1) Feeding processes (nutrition)
- (2) Breathing processes (respiration)
- (3) Growth processes
- (4) Phenomena of movement and of sensitiveness (or irritability)

CHAPTER II

THE mode of germination of seeds the parts of the seed and seedling the resting and active states of seeds the resisting powers of resting seeds, germination capacity the visible order of events in germination The nature and function of cotyledons and of endosperm adaptation in plants large seeds and small seeds

A FULLY grown plant is by no means a convenient subject for experimental purposes Not only is it bulky, but its roots are hidden in the ground and cannot be disturbed without damage to the plant On the other hand, a handful of pea or bean seeds may be obtained for a penny, and, when planted, the seeds produce seedlings in the course of a week or two Moreover, inasmuch as the seedlings grow rapidly, we may assume, from analogy with young children, that they are likely to feed hungrily Hence seeds and seedlings should prove very useful to us in our studies in plant-nutrition We will therefore commence our work by an examination of seeds and seedlings

Since we shall require seeds for all sorts of experiments, we must take every opportunity of getting together a large and varied collection At the proper times of the year, ripe seeds of garden plants, weeds, and common trees should be gathered, dried, and stored in corked or stoppered bottles The bottles should be labelled, and on each label should be written the name of the seed (or fruit), the locality whence it was obtained, and the date of gathering If it is not possible to collect a sufficiently varied assortment of seeds, some may be purchased from seedsmen and stored in labelled bottles Samples of the seeds and fruits should be affixed to cards with the name, natural order, and other interesting details, such, for example,

as locality and weight, and the cards placed in the physiological museum, in which records of experiments, specimens, and photographs, etc., should be kept. In case the beginner does not know how to distinguish seeds from fruits—and some fruits look exactly like seeds—he should refer to an elementary text-book (Bibliography, 3, 5), which deals more particularly with the morphology of plants, that is, with the characters and peculiarities of their form and structure. For, though we are studying the work of plants—that is their functions—we shall have to take notice of their form and structure, and have not space to deal fully with the morphological branch of botany.

Having become familiar with the shapes, sizes, and peculiarities of the seeds and fruits of the common plants, we proceed to germinate some peas. At once the question arises: since the seeds in our collection do not germinate whilst in the bottles, what is to be done in order to make them begin to grow? Now everyone who has access to a garden or to the country knows how quick weeds and other plants spring up in showery weather, and hence we make the sure guess that a supply of water is necessary for germination. Even though we know this we prove it, for, by so doing, we shall extend our knowledge and make it more precise.

3 To this end, prepare three pots of garden soil, dry one thoroughly in a kitchen oven, and, in order to prevent the soil from getting moist again, set it to cool under an inverted marmalade jar, or similar vessel. See that the soil in the other two pots is thoroughly moist. Determine the average weight of ordinary, dry pea seeds by weighing several samples of twelve each. Put a couple of dozen seeds to soak in tepid water, and, after twenty-four hours, dry their surfaces by means of a cloth, weigh and compare them with respect to weight, size, and shape with the dry seeds. Calculate the percentage of water taken up. Now plant four or six peas in each of the three pots, putting dry seeds in the pot with the dry soil, dry seeds in one of the pots with moist soil, and soaked seeds in the remaining pot. Label the pots 1, 2, 3, and note in a rough note-book the details of time of planting, and state the

seed and soil. Cover each of the pots with a glass plate or piece of cardboard or stiff brown paper, and see that the soil in pots 2 and 3 does not get dry. Record the dates of appearance of the seedlings in each of the pots. Copy out the results neatly in a note-book kept for the purpose, and add any remarks that seem interesting. Records should be made of the results of every experiment that is performed, and, whenever it is useful, sketches should accompany the records, which should be arranged in brief, tabulated form.

4. Take the remaining soaked seeds, wipe them, put them in a dry place—for instance, on a shelf in a living room—and weigh them at daily intervals, and thus determine the rate at which they lose water. When they seem fairly dry, put them in a thin paper bag in a desiccator (See Appendix A). At the same time, weigh and place in a paper bag a dozen dry, unsoaked peas, and put this bag, properly labelled, in another desiccator. After an interval of about a week, weigh the two lots and determine how much each has lost in weight. Leave them exposed to the air of a room for some hours, weigh them again, and compare these weights with those of the same seeds when taken from the desiccator. From the experiments, it is evident that seeds are hygroscopic, that is, they take up water from moist air and give up a certain amount of water when the air to which they are exposed is dry. The bearing which these facts have on such matters as the following should be considered—the importance of storing seeds out of contact with moist air—the difference that the weather at the time of harvesting is likely to make to the viability of the seed—the advantage and possible disadvantage of soaking bean or pea seeds before sowing in the garden—the fact that, in wet autumns, seeds of various plants may be found germinating whilst still attached to the parent plant.

In cases where students work in groups, some should use one kind of seed for the above experiments and others another, *e.g.* barley grains (which are strictly fruits), horse chestnuts, onion seeds, etc. The results obtained with these different seeds should be compared with one another.

We have now confirmed our knowledge that seeds,

in order to germinate, require water, we have found that the amount of water which seeds, such as peas, can absorb is surprisingly large, and we have learnt also that seeds are hygroscopic. We recognise that a knowledge of these facts helps us to store our seeds properly and shows us how we may hasten their germination. We will next determine whether seeds dried as thoroughly as possible in a desiccator are absolutely dry, or whether they still contain water.

5 To this end weigh a dozen peas which have been in the desiccator for a week, soak them till they are soft, weigh them with a cloth, and pound them in a mortar, transfer the whole of the mash to a weighed, dry porcelain dish and dry it thoroughly in a drying oven at about 100°C . After two days, take the dish out of the oven, stand it in a desiccator to cool, and then weigh it. Replace the dish in the oven and continue the weighing at daily intervals till no further loss of weight is recorded. We thus obtain the *dry weight* of the substance of the seeds, and a comparison of this weight with that of the desiccator-dried seeds tells us how much water the latter, apparently dry seeds really contained. The result of the experiment proves that even the driest seed contains a considerable percentage of water. The above experiment will be the more instructive if, at the same time, other vegetable tissues, e.g. carrot, turnips, and also fresh leaves (grass or spinach, etc.) are weighed, dried in a desiccator, weighed again, then chopped up (there will be no need to soak them first, as they are not so flinty hard as the seeds), pounded in a mortar, dried in the drying oven, and then dry weights determined.

6 A ready way of proving that ordinary air-dry seeds contain a considerable amount of water is as follows. Half fill a wide-mouthed glass bottle with peas. Stopper the bottle, and place it in the drying oven at about $90\text{--}100^{\circ}\text{C}$. After 1-2 hours remove the bottle and observe that, as it cools, water, given off by the peas, condenses to form drops on the sides. By using different kinds of vegetable tissue it will be discovered that they contain a certain amount—some a very large amount—of water, and that, of vegetable structures, seeds contain far less water than any others. That it is to this la

that seeds owe their resistant powers we demonstrate in the following way:

7 Prepare a saucepan of boiling water, place a few ordinary dry peas and equal numbers of soaked and of desiccator-dry peas in small canvas or muslin bags, plunge them in the boiling water for a few seconds, plant the three lots (after soaking the dry seeds) in pots, and record their germination. Whereas the thoroughly dry seeds have not been injured by their short immersion in boiling water, the soaked seeds show by their failure to germinate that they have been killed. It is noteworthy that advantage is taken of the resistance of dry seeds to high temperatures in treating grains of oats, the surfaces of which are suspected to be contaminated with the spores of a disease-producing fungus called smut. The grains are plunged for five minutes in water at a temperature of 55°C ., and subsequently sown. The effect of the hot water is to destroy the smut spores without injuring the oats.

8 Next, the effects of low temperatures on very dry and on soaked seeds should be determined. The most convenient way to do this is to pound up ice and salt and to put the freezing mixture into a small pail, in the middle of which a glass flask is placed. The several small lots of peas, each lot in a muslin bag, are put into the glass vessel and left there for some hours. The bags of seeds are then withdrawn, and the germination capacities of the three lots of seeds tested.

From the result of this and similar experiments it is learned that dry seeds are more resistant to adverse conditions than soaked seeds.

During the summer we make a comparison between unripe and ripe peas. Definite experiment is not necessary to convince us that the unripe seeds in their young pods contain far more water than the ripe seeds, and we may take it that, during ripening, one process which goes on is the gradual loss of water by the seeds. When, on the one hand, we call to mind the extremes of temperature to which the seeds of plants are exposed during their long winter's sojourn in the ground, and when, on the other hand, we realise the great resistant power of dry seeds, we cannot doubt that this natural drying process, which

takes place during the ripening of seeds, is of advantage to the plant, making undoubtedly in many cases the difference between destruction and survival. That the resting state is due, in large measure, to the natural dryness during ripening may be inferred from the experiments we have made, and also from the fact already noted that, in wet autumns, various kinds of plants may be met with the seeds of which are already beginning to germinate in the parent plants. Specimens illustrating this phenomenon should be collected and added to the museum.

Experiments recently made have proved that certain seeds may retain their capacity for germination for a great number of years, in one instance, among seeds known to have been kept for 87 years, some were found to be capable of germination, and it is interesting to know that experiments are now in progress to determine for how long seeds, which have been dried as thoroughly as possible, can retain their vitality. Though, as we have just learned, dry seeds may survive for many years, there is no evidence to prove the truth of the statements which are often made that wheat grains and seeds of other plants deposited thousands of years ago with mummies in tombs in Egypt have retained till the present day their powers of germination. Indeed, there is good reason to believe that such "mummy wheat" has long ago lost its vitality.

Our experiments demonstrate that a seed is a structure which, by reason of its dryness, is capable of passing through a long resting stage. Whilst in the dry state it is far more resistant than is the growing plant. Providing it with water, the seed may be caused to pass from its resting or latent state into one of activity. What is true of seeds is also true of the simplest reproductive bodies of many of the lower plants. For instance, it has been shown that the resting spores of some bacteria are not destroyed by exposure to such high temperatures as 100° - 120° C., at which temperatures the bacteria in their active, growing state are killed. That this great power of heat-resistance is due to the dryness of the spores is proved by the fact that, if the spores are brought under such conditions of moisture and warmth that they begin to germinate,

they lose their resistant powers. Since the group of plants known as bacteria includes many disease-producing forms, and since some of the latter produce resting-spores, the bacteriologist and the doctor have to take the resistant powers of spores into account in their efforts to exterminate disease-producing germs.

Let us now return to our study of seeds, and set ourselves to find out what is the first visible sign of germination.

9 In order to do this sow samples of soaked seeds, some in earth, others in germinators. Germinators of various patterns may be obtained ready-made (Appendix B), but one of the simplest and most useful may be made from a couple of ordinary saucers. Several layers of thick white blotting-paper are moistened thoroughly and fitted neatly into one saucer. A few soaked seeds are distributed on the blotting-paper, and the other saucer, into which moist blotting-paper may also be fitted, is inverted over them. The only precautions necessary are that the blotting-paper should not be too wet to begin with nor become too dry. The saucers may be covered with a large jar or box and stood in a warm place. Every second day, two or three soaked peas are put into the germinator in order that we may obtain all the various stages of germination at one and the same time. In the course of a day or two, a small white conical structure is to be seen projecting from the first sown seeds. It elongates to form a cylindrical body with a rounded end. By comparing older with younger stages, it will be seen that this body is, at all events, the greater part of it is the young root or radicle. This method of making comparisons backward, that is, of comparing an older with a younger structure, is very useful, and will often enable us to determine the nature of doubtful structures. Whilst the pea seeds are germinating, drawings should be made of each stage up to the time when the various parts of the seedling are recognisable. Germinate also a number of other seeds, e.g. mustard or cabbage, onion, etc., in order to demonstrate that what is true of the pea is true of other seeds, namely, that the root is the first member to make its appearance. We

might indeed have guessed that this would prove the case, for it is by the root that the plant lives in the soil, and it is an essential condition for the majority of flowering plants that they should be "rooted" in the soil.

The next stage in germination may be seen in seedlings, the radicles of which are about an inch length. In such seedlings, a curiously looped structure makes its appearance. One end of this structure is continuous with the radicle, the other end is still in the soil. Gradually the loop lengthens and ultimately its free end becomes visible. This free end, when looked at thru a pocket lens, presents a somewhat plum-like appearance, the parts corresponding to the feathers being small, g flat structures, the smallest of which are very minute wrapped round the tip of the looped body. By comparing this with later stages, it is evident that the looped body which soon straightens out and points upward, is the stem, and the small green structures, which arise as outgrowths from it, are the leaves. This seedling stem, with its small leaves, is called the plumule. Following it downward towards the radicle we find that, at a certain point, are attached to it, on either side, two stalks. If we remove the seed-coat from the seed of a germinated pea, we discover that the part of the seed which was enclosed separates into two halves, and that each part is continuous with one of these stalks. The two, massive, more or less hemispherical bodies, which are connected by stalks to the stem of the seedling, are called the cotyledons. Presently we shall have to find out what they are and why they have this peculiar shape.

The different parts of the seedling have received different names. Not is this unnecessary, for it facilitates comparison between different seedlings and enables us to appreciate the fact that, despite their great difference in shape and size, large numbers of seeds and seedlings built on the same fundamental lines. That this is so may be seen by comparing the seeds and seedlings of the radish, and sycamore, illustrated in Figs. 2, 3, and 4.

Each of these seedlings consists of a shoot and a radicle. The shoot is made up of a leaf-bearing

which terminates in a bud (the plumule). That part of the shoot which lies above the insertion of the cotyledons is called the epicotyl, and that part which lies below the cotyledons and merges insensibly into the root is called the hypocotyl. In some plants, e.g. pea, the epicotyl makes up nearly all of the mature shoot-system, the hypocotyl remaining short; in others, e.g. radish, the epicotyl grows but little. Throughout its first year, it remains very short, and produces a rosette of leaves arising

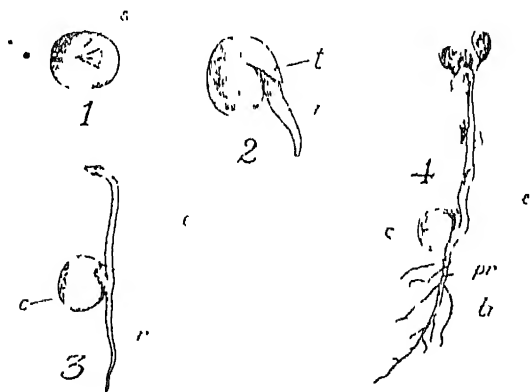


FIG. 1. GARDEN PEA (*Pisum sativum*). SEED AND SEEDLINGS.

1, Seed; 2, 3, 4, Seedlings in three successive stages of germination. *t*, Seed, *c*, testicle, *e*, cotyledon, *p*, radicle, *e*, epicotyl, *p*, primary root, *l*, lateral root.

ing from about the ground level. The hypocotyl, on the contrary, increases considerably in length, and, with the root, forms the edible part of the radish.

By comparing such very different seedlings as those of the pea, mustard or turnip, the castor-oil plant, etc., we reach the conclusion that similar parts are present in each, though they are developed to different extents in the different plants.

We have now to investigate the nature of the two large, lobed bodies which, as we have seen (p. 14), are attached each by a stalk to the stem of the seedling. By re-

moving the coat of a soaked, ungerminated pea or bean seed we find that, on pressing gently on the seed along the edge opposite the radicle, the white fleshy mass shows signs of separating into two halves. If we insert the point of a pen-knife in the split we

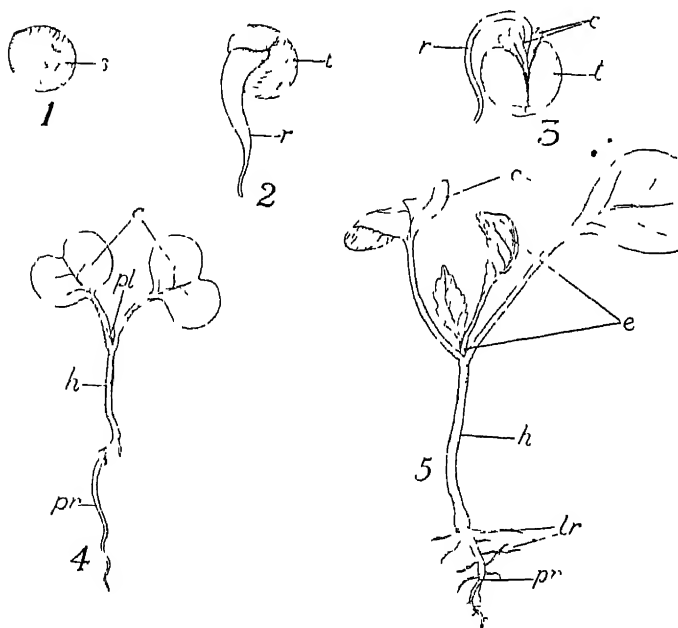


FIG. 3 —RADISH (*RAPHANUS RAPHANISTRUM*) SEED AND SEEDLINGS

1. Seed. 2, 3, 4, 5. Seedlings in four successive stages of germination. *s*, seed, *r*, root, *t*, radicle, *e*, cotyledon, *pl*, plumule, *h*, hypocotyl, *pr*, epicotyl, *pr*, primary root, *lr*, lateral roots.

can, without tearing anything, force the two halves apart and see, by the aid of a lens, that the epicotyl, though small, is already formed, and lies pressed close to the inner surface of one of the two fleshy bodies, which we recognise as the cotyledons. Hence the seed of

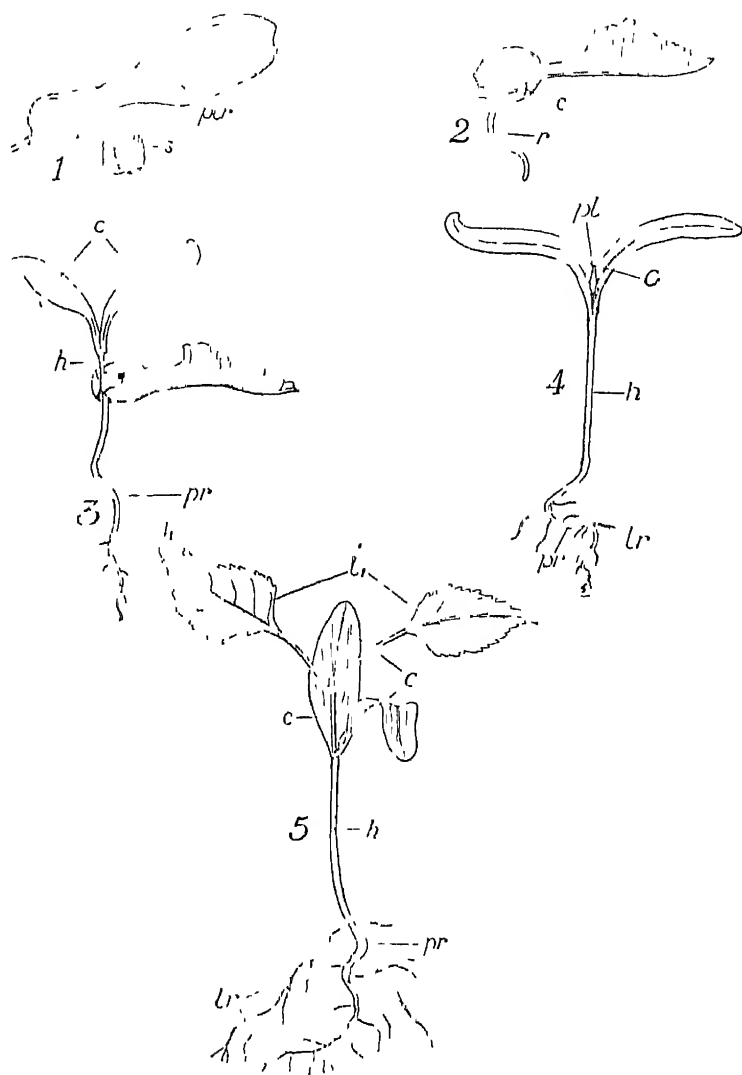


FIG. 1

SYCAMORE (*ACER PSEUDO-PLATANUS*) FRUIT, SEED AND SEEDLINGS

1. Half fruit (micro up) seed removed 2, 3, 4, 5. Seedlings in four successive
 stages of germination. *pr*, perianth, *h*, first foliage leaves, other lettering as in Fig. 3

K. P.

B

the pea consists already of a miniature plant with epicotyl, hypocotyl, radicle, and cotyledons attached to the stem at the junction of epicotyl and hypocotyl. The whole embryo, as the plant in the seed-stage is called, is enclosed in a seed-coat, which, except for a minute hole (micropyle) at the place where the root will emerge, forms a continuous envelope about it. The structure of the pea or bean seed is now clear, except in one particular. We recognise in the embryo all the parts present in the young plant with the exception that nothing comparable with the cotyledons occurs in the latter. We must therefore attempt to discover what the cotyledons are, and what are their particular functions.

First, we will see whether we can find out something about the cotyledons by examining the young pea or bean-plants which we have raised from seed.

Dig up a plant which is from six inches to a foot high, and, observing that the remains of the seed are still attached to the stem, remove the seed-coat and note the shrivelled cotyledons. Sometimes we may see that a bud or even a small branch springs from the angle (axil) which the stalk of a cotyledon makes with the stem. If we do not find such axillary buds or branches, we have a ready means of causing them to grow large enough to be seen, namely, by preventing the growth of the main stem.

10 Thus, having selected a seedling bean-plant about three inches in height, growing in a pot or in the open, cut away or pinch off the stem at the ground level. After some weeks, we note, on digging up the plant, that the two shoots which have formed arise, each in the axil of the stalk of a cotyledon. Now take a branch of a tree and examine it to see how its lateral buds and branches arise. They will be found, in the vast majority of plants, to occur only on the stem just where a leaf is borne, in other words, lateral buds are borne in the leaf-axils. We may argue thus, lateral buds and the branches which they produce arise in the axils of leaves: a lateral bud, which develops to a branch, arises in the axil of a cotyledon: therefore the cotyledon is a leaf. But such an argument, to be convincing, requires to be supported by other evidence. Let us therefore see if further evidence is to be

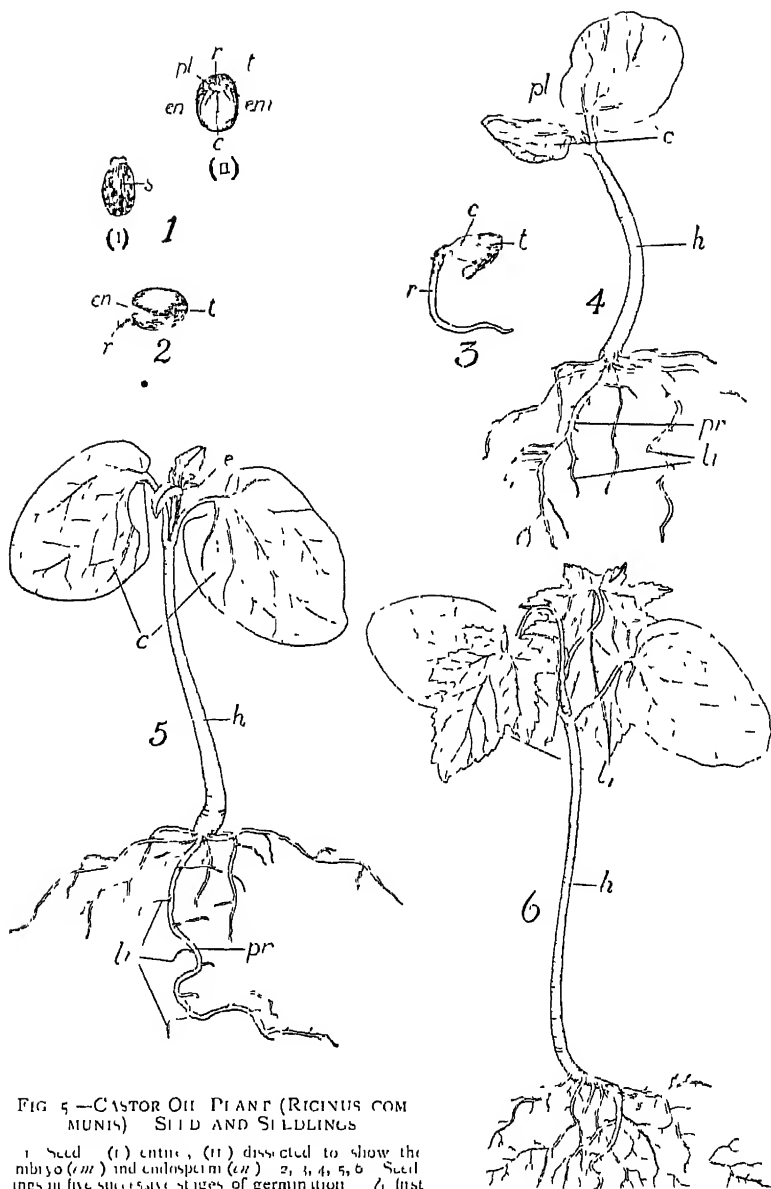


FIG 5 —CASTOR OIL PLANT (*RICINUS COMMUNIS*) SEED AND SEEDLINGS

1. Seed (1) entire, (2) dissected to show the embryo (*cn*) and endosperm (*en*). 3, 4, 5, 6. Seedlings in five successive stages of germination. *h*, first

found. The kidney bean (*Phaseolus vulgaris*) is very like the broad bean (*Vicia faba*). They both belong to the same family, and are likely, therefore, to be similar in the essential characters.

11 Sow three or four kidney beans and study their germination. It will be found that they make no concealment of the foliar nature of their cotyledons, for, as germination proceeds, the cotyledons are drawn out of the seed-coat, borne upward above the ground, turn green, spread out flat and show themselves to be leaves. Cotyledons which rise above the surface of the ground are said to be epigeal; those, e.g. of the broad bean, which remain below the ground, are termed hypogeal.

As an exercise in observation—and observation requires exercise and frequent exercise for its development—determine the nature of the cotyledons of the castor-oil plant (*Ricinus communis*, Fig. 5), firstly, by dissecting carefully soaked seeds after removing their coats, and, secondly, by observing germination-stages.

It remains to find an answer to the question—why admitting that it is a leaf, is the cotyledon apt to depart so much from the conventional form of leaves?

12 If, during early spring, we look at a lilac or privet bush just when the buds are breaking, we find that the outer leaves of the bud, though green and like the inner leaves, remain small, and may fall off as the bud grows to form a branch. But if we cut away the top of the bud with its group of very young leaves just after it has opened, the outermost leaves, which normally remain small and fall away, grow into ordinary foliage leaves (museum). Now repeat the observation and experiment on the buds of the flowering currant (*Ribes sanguineum*). By dissecting away the leaves of the just-opening bud and laying them out in order on a sheet of paper, we see that they form a series which when we read it backwards, from the innermost to the outermost leaves, comprises fully-formed foliage leaves consisting of blade, stalk, and leaf-base, leaves, the base of which is well developed and the blade only just recognisable, and leaves consisting of leaf-base only (see Fig. 6). These last never grow into ordinary leaves, and

having served the purpose of protecting the bud during winter, are cast off on the opening of the buds in spring (museum). From these examples we learn that a plant-member, such as a leaf, may be constrained, according to the need of the plant, to change its function. We learn

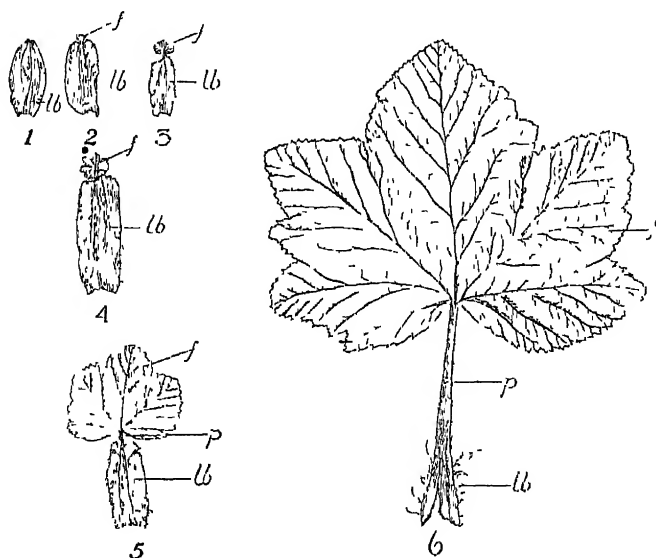


FIG. 6—FLOWERING CURRANT (*RISES SANGUINEUM*)

1, 2, 3, 4, 5. Series of leaves from opening bud, showing transitions from scale leaf (1) to lobed leaf (5). lb, leaf blade, p, leaf stalk, f, blade.

further that when the function changes, the form may also change.

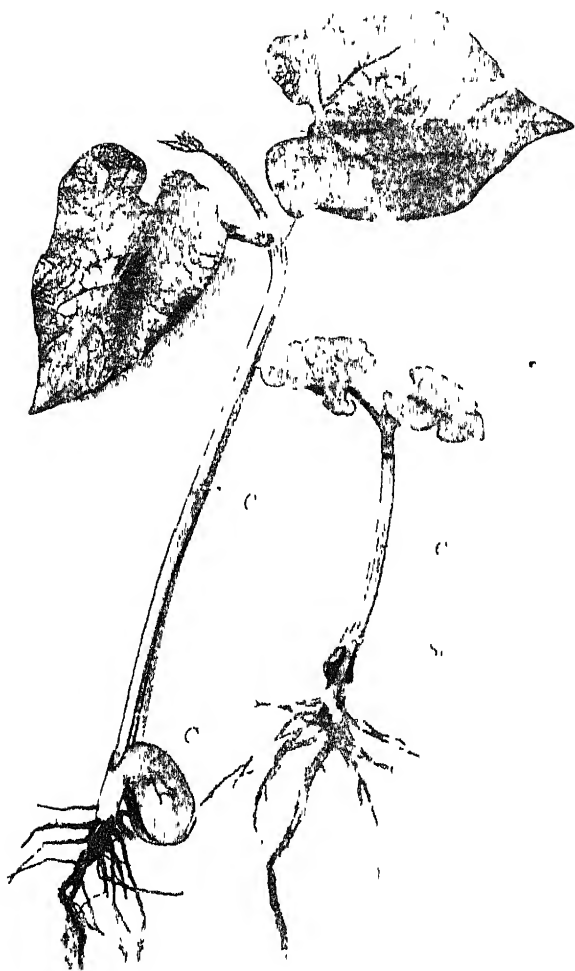
Many important facts follow from this conclusion. For example, it is evident that the particular form assumed by a leaf, or stem, or root, has some relation with the kind of work it has to do. Hence, just as we infer when we see a sharp-edged tool, that it is used for cutting, so, when we have had some general experience of the kinds of work to which the different parts of a plant's machinery are

put, we are able to infer from a departure from the normal usual form displayed by a member of the plant-body if that member is charged with a new, unusual duty. If power of adjusting means to ends is possessed by plants and animals to an extraordinary extent, and is spoken of as adaptability, the modification being called an adaptation. The power of adaptation by which an organism may modify the function of an organ and effect change of structure and of shape serving to fit the organ to its new work, should be studied by the student in suitable seasons of the year. For example, he should determine the morphological nature of the tendrils of pea plant, asking himself, are these structures stems, leaves, or parts of leaves, or roots? When he has solved this morphological problem, let him ask himself whether this departure from the normal structure fits the organ for the work it has to do. Among the innumerable subjects for such morphological exercises we may mention potato tubers, opening buds of beech, onion or hyacinth bulb, strawberry runners, and double flowers (stocks, roses, etc.), but the best subjects are those which the student discovers for himself. The method he must use is that of comparison. He must compare the thing with itself at different stages of its development, *i.e.* he must study its developmental history or embryology, and he must compare the thing with its nearest allies, *e.g.* a double flower with a single flower of the same species—a garden rose with a wild rose, and so on. This is the method of comparative anatomy, which may be pursued further and made to include microscopic as well as a naked-eye investigation.

Now to apply our conclusions to cotyledons. In some plants, the foliar nature of the cotyledons is obvious, *e.g.* in plants of the cabbage tribe—turnip, cress, cabbage, etc.—and in the castor-oil plant, etc., in others, though not apparent at first, it becomes so during germination, *e.g.* kidney bean, in yet others, *e.g.* the pea, bean and horse chestnut, it is by no means apparent. Keeping in mind the cases of the buds of the lilac or privet and of the flower of currant, we surmise that the reason why some cotyledons have retained their foliar character whilst others have, in large measure, lost it, is that the former have

retained the functions of ordinary leaves, and that the latter have exchanged these functions more or less completely for others. Change of form is an outward and visible sign of change of function. What then is the new function which the thick, unleaf-like cotyledons of the pea have assumed? When the castor-oil seed was dissected, the student must have been struck by the fact that the embryo did not take up the whole of the space within the seed-coat. He saw that, covering the thin, delicate seed-leaves (cotyledons), which lie pressed together in a plane median to, and parallel with the flatter surfaces of the seed, there is a mass of soft, white tissue. This tissue, called endosperm, from which the castor oil of commerce is extracted, has no counterpart in the ripe pea or bean seed. On the other hand, whilst the cotyledons of the pea and bean are thick and unleaf-like, those of the castor-oil seed are thin, and so leaf-like as to show even their "veins." Let us state the facts thus —Castor-oil seed, cotyledons thin and leaf-like, a mass of tissue (endosperm) external to the embryo making up the larger part of the seed. Bean seed, cotyledons fleshy, no corresponding endospermous tissue external to the embryo, the seed-leaves making up the larger part of the seed. If we could discover the use of the endosperm to the seedling, we could make a good guess why the cotyledons of the bean are fleshy. Conversely, if we could find out what purpose is served by the fleshy cotyledons, we should know probably why the castor-oil seed contains endosperm. Again, as always, when confronted with problems in plant-physiology we have recourse to our scientific method of guess and experiment. Now it cannot have escaped our notice when we were germinating various kinds of seeds that some are small and produce small seedlings, and some are large and produce large seedlings. If we examine seeds of the latter kind, we see that they have invariably either large and fleshy cotyledons or much endosperm, and if we dissect seeds of the former kind we find that the cotyledons are thin and that, if any endosperm is present, the amount is but small.

As a study of the museum specimens shows, the size of a seed has a relation with the size of its cotyledons



A

B

FIG. 7 — KIDNEY BEAN (*PHASEOLUS VULGARIS*)
From a Plate cut

A Seedling germinated under normal conditions. B Seedling from seed, at the same time as A, but from which the cotyledon was removed at germination. c, cotyledon, e, epicotyl, Se, scutellum left after removal of cotyledon.

of endosperm, and not with the size of the mature plant which it produces. Moreover, a large seed produces a large seedling, and a small seed a small seedling. We can scarcely doubt that the big seedling is due in some way to the big cotyledons or to the large amount of endosperm, and, thinking of the effect which proper feeding has on the size of young animals, the idea dawns upon us that perhaps the large cotyledons and the endosperm contain food supplies on which the seedling is enabled to feed during germination. This hypothesis we can put to the test. If it is true, then, as the seedling grows, the cotyledons should be found to shrivel and the endosperm to disappear. We have already seen the shrivelled cotyledons of the young bean plant (p. 18), and so have some confirmation of the correctness of our hypothesis.

13 To obtain complete proof we may proceed to weigh the shrivelled cotyledons of a bean which is about a foot high. After drying them in a drying oven, we compare their weight with that of the cotyledons of ungerminated bean seeds, similarly dried. We find that the dry weight of the shrivelled cotyledons is considerably less than that of the fresh cotyledons of ungerminated seeds. Another and more striking method, as soon as the epicotyl appears in each of six kidney bean seedlings sown in germinators, remove, by means of a sharp knife, the cotyledons from three of them, and plant all six seedlings in a pot with garden soil. Determine the rate of growth of the two sets (Fig. 7). Record, draw and preserve one plant of either set for the museum. (For this purpose the plants may be dried between blotting-paper and mounted on cards.) A similar experiment may be made with maize or wheat seedlings, in which case it is the endosperms which must be removed (cf. Fig. 8). The result in either case is the same. The seedlings formed from the seeds deprived of their fleshy cotyledons or of their endosperm, though they grow, increase in size much more slowly than those from intact seeds. Whence we conclude that endosperm and fleshy cotyledons serve as reservoirs of food-materials on which seedlings draw for their nutrition. Presently (Chapter III) we shall have to enquire into the nature of these food-materials and how the seedling obtains them.

It is interesting and typical of the ways of living that the same end, in this case that of endowing the young with a capital of food wherewith to start in life, be secured by different ways—the one, by converting

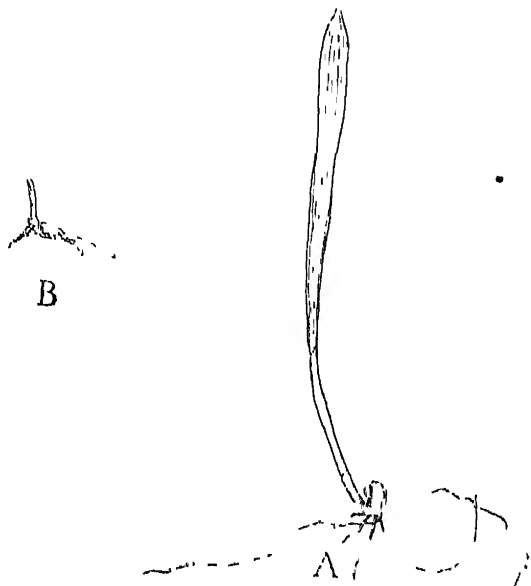


FIG. 8—WHEAT (*TRITICUM Aestivum*)

Showing the relative size of seedlings grown with and without endosperm.

A, Seedling, germinated on moist blotting paper. B, seedling of the same age, germinated under similar condition, except that the endosperm was removed at germination began.

seed-leaves into storerooms, the other, by the provision of a separate, nutritive tissue (endosperm) external to the embryo. In point of fact, the difference between wheat, and seeds without endosperm only amounts to that the embryo of the endospermous seed waits until it begins to grow before taking up the provision of material made for it by the parent plant, whereas

embryo of the seed without endosperm takes up the food-substances and stores them in its cotyledons before it ripens and comes to rest. If the parent makes a liberal provision, the cotyledons become thick and fleshy, if the provision is niggardly, the cotyledons remain thin and the seed is small. With a knowledge of the meaning of big cotyledons and large masses of endosperm, we obtain an understanding of the significance of the great variability of seed-production shown by different plants. The parent plant provides the material out of which the tissues of the cotyledons or endosperm and embryo are formed. The parent may produce either many small seeds or fewer large seeds: compare, for example, the poppy and pea. The one or the other habit runs in plant-families, and it is interesting to note that, of two of the most successful plant-families, the members of one, Leguminosae (peas, beans, clover, vetches, etc.), produce large and relatively few seeds, those of the other, Compositae (daisy, dandelion, etc.), produce small and relatively many seeds. We can also understand that the work of seed-production entails a certain amount of exhaustion to the parent plant, and that, therefore, a copious crop of seeds and fruits one year may use up so much of the food-material at the disposal of the parent that but few, or even no seeds may be produced the following year, *e.g.* beech, etc. So much is this the case that there is a great group of plants which die after once flowering. To this group belong our annual garden plants and weeds which live for one season or less, produce seed and die. If they are prevented from flowering, their lives may be prolonged, as any one with a garden or even a space for pot-plants may prove for himself. Not all once-flowering plants, however, are annuals, various tropical or sub-tropical shrubs and even trees grow for many years and ultimately flower, set seed, and die (*e.g.* species of aloe, palms). Other plants are biennials, that is, flower in their second year (turnip, carrot, long-love), whilst others again are perennials, and may live for centuries and flower again and again.

Having accomplished our objects, of learning something of the general nature of plants and of beginning our studies on nutrition, we will conclude this chapter with

the remark that the pea or bean type of seedling has its axis terminating above in a plumule and below in a radicle and bearing two cotyledons laterally, is the only type of seedling to be met with among the plants. If we call this type the *Dicotylous type*, we may say that there is another, very varied type, a

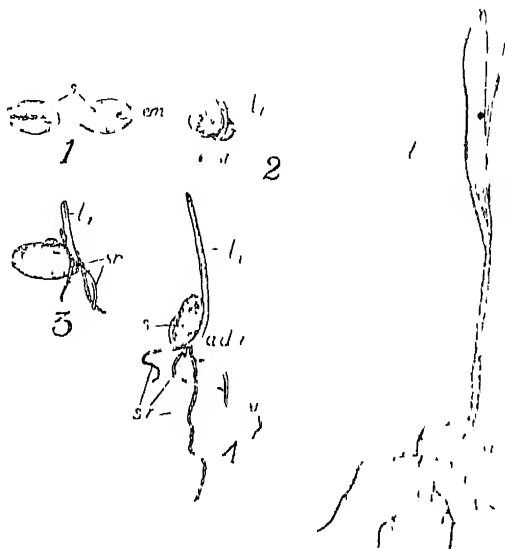
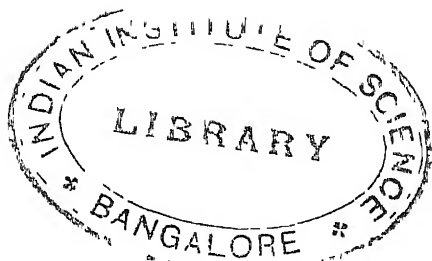


FIG. 9.—WHEAT (*TRITICUM SATIVUM*)—FRUIT AND SUCCESSIVE STAGES OF GERMINATION.

1, Grain of wheat (*fr*, fruit enclosing one seed). 2, 3, 4, 5, 6, 7, 8, 9, Successive stages of germination. *pl*, plumule; *co*, cotyledon; *ra*, radicle; *fr*, fruit; *co*, cotyledon; *ra*, radicle; *st*, shoot; *rt*, root; *ad*, adventitious root; *st*, shoot system; *l₁*, *l₂*, *l₃*, 1st, 2nd, and 3rd leaf whorls.

Monocotylous type. Seeds of the latter, such as wheat, oats, barley, maize, and date, should be examined, then seedling stages studied, and put in the museum (Fig. 9). Some help will be given in making out the parts of such seeds as there may be obtained from a text-book of general Botany (Bibliography, 3, 5).



CHAPTER III

THE nature and chemical properties of the food substances contained in
• the cotyledons and endosperm of seeds

WHEN we reflect on the results of our observations and experiments on fleshy cotyledons and on endosperm, we are bound to be struck by the fact that those seeds which contain large quantities of food-material serving for the nutrition of the seedlings are also the seeds which are used most largely by man and animals for food. When cereal crops are ripe, that is, when the plants have transferred stores of food-materials to the endosperm of the seed, man intervenes and, gathering in the crop, makes flour from the grains. Flocks of birds anticipate man, and by their depredations bring serious loss to the farmer. Teeming populations in the East support life solely on rice the seed-like fruit of a grass (*Oryza sativa*). The oilcake on which cattle are fattened is derived from the remains of the reserve-materials of the seeds of rape, cotton, etc. Thus the conclusion impresses itself upon us that the food materials of the seedling serve also as food for man and animals. That this is so, will add interest to our present enquiry into the nature of these food-materials and the way in which they are used by the seedling. Since a thorough examination into the chemistry of the food-materials, or, as we may call them, the reserve-materials of the seed, involves both a knowledge of chemistry and also the occasional use of the microscope, and since some students may lack the necessary chemical knowledge or be unskilled in the use of the microscope, we will indicate by means of an asterisk (*) the experiments which may be omitted by beginners.

531-1

2829

14 Soak one or two wheat or barley grains in water until soft, cut them in halves, and smear some of the endosperm on a saucer or porcelain slab. Add a few drops of a solution of iodine dissolved in water with potassium iodide (Appendix A), note that a blue colour is produced. Obtain some finely powdered, pure starch from a dealer and repeat the potassium-iodide-iodine test (1) by adding the solution direct to the powder, (2) by first boiling a small quantity of the starch with water in a test tube or beaker, then add the boiled liquid with several times its volume of water, and, after cooling, adding the iodine solution. If

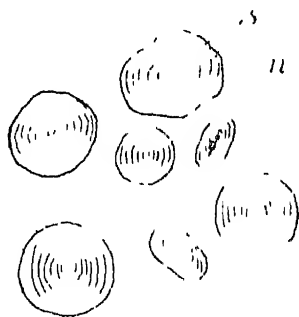


FIG. 10.—STARCH GRAINS FROM THE ENDOSPERM OF BARLEY. *a*, hollow of grains; *b*, lines of refraction due to different layers in successive layers.

blue liquid, note that the colour disappears on cooling; the tap, note that the colour reappears. We thus confirm that the blue colour obtained in the previous experiment is due to starch. We speak of this reaction as a test for starch, or as the *iodine-reaction* for starch.

15* Scrape the cut surface of a soaked barley grain with a clean knife, transfer a few scrapings to a small drop of water on a slide. Dry on a cover glass and examine the preparation at first with a low and then with a high power of the microscope. Several starch grains, indicating the characteristic concentric markings (Fig. 10).

16* Cut sections through the endosperm of wheat

cotyledons of the bean and the tuber of the potato. Note the positions of the starch grains, groups of which lie in the compartments of which these tissues are composed. Draw. Run in under the cover glass of one of these microscope-preparations a drop of potassium-iodide iodine solution. Note the blue or blue-black colour of the grains. Crush soaked wheat grains in a mortar with a little water, transfer some of the pasty mass to a test-tube, shake well and filter through a filter funnel lined with filter

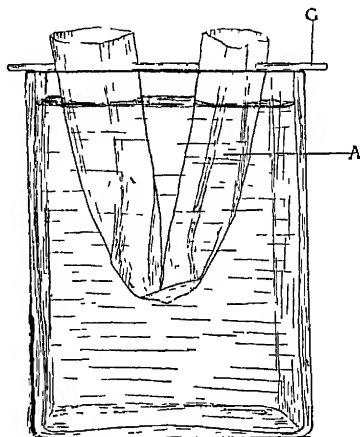


FIG. 11.—DIALYSER
A, parchment tube, C, glass rod

paper. Determine that the filtrate, *i.e.* the liquid which passes through the filter paper, gives no blue colour with the iodine solution, in other words, that starch is insoluble in water.

17 Boil some commercial starch with water. note that it passes into an opalescent, thin, jelly-like condition, but does not dissolve. Dilute some of this starch-containing liquid with water and pour it into a parchment shell or tube of parchment paper (Appendix B), bent in the form of a U, and supported in a beaker of water by means of a glass rod passing through the upper part of the limbs of the U (Fig. 11), leave it for twenty-four hours. If

starch is a diffusible substance it will pass through parchment paper into the water in the beaker. Prove the iodine test that starch is not diffusible.

18 Add a little starch—as much as will go on end of a knife blade—to about a half-pint of water in a beaker, stir well, and boil. Add by means of a glass rod two or three drops of a mineral acid, e. g. hydrochloric acid, boil, and at hourly or convenient intervals test the liquid for starch thus:—by means of a glass rod, disperse drops of the liquid on a porcelain saucer, dip another glass rod into the iodine solution, and mix drops of latter with those on the saucer. Observe that the reaction becomes fainter as time goes on, and finally boiled with small quantities of mineral acid, is decomposed. Keep the solution in a clean bottle (p. 18, exp. 25). It is important to find out to what substances the starch has given rise. We may hint with respect to this in the course of the following experiments which we perform in order to find out if it is generally present in seeds, and if it is widely distributed in other parts of plants.

19 Apply the iodine test to the cut and moist surfaces or to thin slices of the following seeds: nut, date, maize, turnip, almond, horse-chestnut, rape, castor-oil plant, and others which can be spared the seed-collection. Dissect out the embryos of germinated barley or wheat grains, put them for about an hour into a solution consisting of equal parts saturated chloral hydrate and potassium-iodide (Appendix V). Note that the embryos give a beautiful starch-reaction.

20 Cut thin slices of unripe and ripe apples or oranges, and compare them as to starch content. Test thin slices of the stems and roots of various flowering plants for starch. Find out whether starch is present in leaves, e. g. of clover, lime, lily, American water-lily (*Nelodea canadensis*), snowdrop, iris, and other plants. For this purpose, proceed as follows: throw the leaves, or other parts, into boiling water for a minute or two. Take them out and put them into a wide-mouth bottle containing methylated spirits. Stopper or

the bottle, and expose it to sunlight after a day or two, remove the colourless and brittle leaves carefully from the alcohol, wash them in water, and pour over them chloral hydrate, iodine solution as used in the previous experiment. Record which leaves contain starch, and which do not. Test similarly pieces of fern fronds, moss leaves, the green threads of any algæ which may be found growing in ditches, and pieces of mushroom or of toadstools.

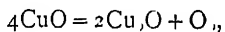
When ripe and unripe fruits, apple, etc., were tested for starch, it was noticed that the unripe fruit contained much starch and the ripe fruit far less or even none. On the other hand, the ripe fruit is sweet, and the unripe fruit is not. It therefore appears likely that the sweetness of the ripe fruit is due to the conversion of starch into sugar. This is rendered the more probable in that both starch and sugar consist of the same three elements, viz. Carbon (C), Oxygen (O), and Hydrogen (H), and have this further in common, that the proportion of hydrogen to oxygen in their respective molecules is as 2 : 1 (just as it is in water, the chemical formula of which is H_2O). Bodies having this constitution are classed in the chemical group of the carbohydrates. The differences and similarities of the constitution of the molecules of starch and of a sugar may be seen if we write down their respective formulæ

Starch, $(C_6H_{10}O_5)_x$ Sugar (grape sugar), $C_6H_{12}O_6$

The x outside the bracket in the starch formula means that a starch molecule has not 6 carbon, 10 hydrogen, and 5 oxygen atoms, but some multiple of these numbers, e.g. x may equal 100 or more (its exact value is not certain), and assuming that it is 100, the composition of the starch molecule is $C_{600}H_{1000}O_{500}$. If we neglect this complication of the bigness of the starch molecule, and write it $C_6H_{10}O_5$, then it is evident that it only differs from the grape sugar (glucose) molecule by H_2O , in other words, if we could add a molecule of water to one of grape sugar we could represent the conversion of starch into grape sugar thus $C_6H_{10}O_5 + H_2O = C_6H_{12}O_6$. When water is induced to combine with such a body as starch, the change undergone by that body is described as one of hydrolysis. These considerations with respect to

the chemical constitutions of starch and sugar is there is no inherent improbability in the view that when it disappears from the ripening fruit, it is because it has been converted into sugar.

Inasmuch as sugars are widely distributed in plants and evidently play a part in plant- as well as in human nutrition, we must consider their properties. We will begin by testing grape sugar (glucose) and cane sugar (sucrose) for the experiment. We also require a ready test for the identification of this important class of substances. The test we employ is based on the fact that some sugars have the power of taking oxygen from certain substances, and hence of reducing these substances to a less oxidised condition. For example, such reducing sugars, under suitable conditions, take oxygen from cupric oxide (CuO), reducing it to cuprous oxide (Cu_2O), thus

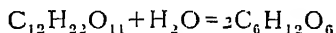


and inasmuch as cuprous oxide is an insoluble, yellow substance, we are able to see if it is produced when a solution of sugar is added to one containing cupric oxide. Thus, we have a basis for a test for reducing sugars. In the test we apply as follows:

21 Pour the liquid to be tested into a test tube. Add an excess of potash and then a few drops of a solution of copper sulphate. The precipitate is copper hydroxide $\text{Cu}(\text{OH})_2$, which is formed by the interaction of copper sulphate and potash in the presence of the excess of potash. Boil. If a reducing sugar is present, a yellow-red precipitate of cuprous oxide is formed. Dissolve grape sugar (which is readily obtainable from sugar cane) in water and apply the above test. Repeat with cane sugar. Observe that grape sugar is a reducing sugar, while cane sugar is a non-reducing sugar.

22 Pour another portion of the solution of cane sugar into a beaker and add a drop of hydrochloric acid. Boil for some time, and apply the potash and copper sulphate test. Observe that a well-marked precipitate is formed. It is evident from the experiment that cane sugar, when boiled with a trace of mineral acid, yields a

sugar. The process whereby a non-reducing is converted into a reducing sugar is called inversion. In the case of cane sugar, the chemical change may be expressed thus



The change is one of hydrolysis, that is, one in which water is caused to combine with the original body, in this case cane sugar. Instead of using potash and copper sulphate separately in testing for reducing sugars, a ready-made solution called Fehling's solution which produces the same result (Appendix A), may be employed.

23 Test by means of potash and copper sulphate or Fehling's solution the following plant tissues or extracts: the rind of ripe and unripe apple, beet, carrot, onion.

24 Crush with water in a mortar grains of barley which have been soaked in water for one or two hours. Filter the extract, label it Extract A. Now crush grains of barley which, germinated in a germinator, are showing leaves an inch or two long, prepare a watery extract, label it Extract B. Test samples of A and B for sugar. If neither gives a distinct reaction, concentrate A and B each to a small bulk by heating them in porcelain dishes over a bunsen flame or water bath and then repeat the sugar test. It will be found that the extract of the soaked ungerminated grains contains no sugar (or at most a trace) whereas the extract of the germinated grains contains a considerable amount. Prove that the sugar is contained in the endosperm. It looks certain from these results, and also from those obtained by testing ripe and unripe fruit (apple, etc.), that the starch contained in the ungerminated barley and in the unripe fruit becomes converted during germination and during ripening into sugar. We have a means at hand of demonstrating that starch *does* give rise very readily to sugar.

25 Take the liquid obtained by boiling starch with water and a mineral acid (Exp 18). Test a sample in a test tube for sugar. If no marked reaction is obtained concentrate the remainder of the liquid to a small bulk and repeat the sugar test. Learning thus that boiling with a

trace of mineral acid suffices to hydrolyse starch to sugar, we shall be the more ready to believe that the sugar which appears in ripening fruit or germinating barley is produced by the hydrolysis of starch. Presently we shall have to determine what agent in the tissues of the plant is responsible, like the mineral acid of Exp. 18, for this conversion of starch to sugar, and of what significance to the seedling is this conversion. When we do this we shall require to know the various properties which distinguish sugars from starch. One such property, that of solubility in water, is well known.

26 Another, that of diffusibility through a parchment membrane, is no less important, and must be demonstrated by the use of the dialyser (Exp. 17). Some hours after the sugar solution has been placed in the parchment tube, the liquid contained in the vessel in which the tube hangs is tested for sugar either directly or after concentration to a small bulk. The result of the experiment proves that a solution of sugar separated from water by a parchment membrane passes across the membrane into the water. This passage is called osmosis, and substances possessing this property are called osmotic substances (see Chapter VII). Let us now refer to our list of seeds tested for starch (Exp. 19), and note those from which starch was found to be absent. They should include the following — rapeseed, castor-oil, almond, date, Brazil nut. But these seeds all have either endosperm or fleshy cotyledons, and it is therefore more likely that they contain some form of reserve food-material other than starch, than that they contain no reserve food-material at all. A scrutiny of the list shows us that certain of these seeds contain fat or oil, indeed, as we know, they owe their use in commerce to this fact. We will select one kind of seed from the above list for examination, e.g. the castor-oil seed (*Ricinus communis*).

27 Pound a seed in a mortar with a few drops of ether or benzol (Appendix A) (this must not be done near a flame, for ether and benzol are highly inflammable, it is safest to do it out of doors). Pour the ether or benzol extract on a sheet of paper. Note that, as the liquid evaporates, it leaves behind a grease spot. To the grease spot or to the remains of the crushed seeds, add a drop of

two of osmic acid (Appendix A) Note that the osmic acid produces a black or brown colouration Repeat the test on a drop of salad or linseed oil in a white saucer

28^{*} By means of a dry razor or knife cut thin sections of the endosperm of the castor-oil seed, mount them on a slide, examine them microscopically and run in osmic acid note the brown-black masses of oil

29 Now examine, by means of the above tests, the remains of the endosperm of seedlings of the castor-oil plant in different stages of germination note that, as germination proceeds, the oil disappears from the endosperm From analogy with the occurrence of starch in other seeds and with its behaviour during germination, we conclude that oil and fat are the forms in which some seeds store their reserves of food Like the carbohydrates, fats and oils contain the three elements, carbon, oxygen, and hydrogen, but whereas in the carbohydrate molecule there are twice as many atoms of hydrogen as there are atoms of oxygen, in fats the number of hydrogen atoms per molecule is more than twice the number of oxygen atoms For a fuller description of the chemical and physical properties of fats, and also of sugars, the student should consult a text-book of Organic Chemistry (Bibliography, 10, 11)

Among the seeds which were germinated in the course of experiment (Exp 19), and tested for starch (Exp 26), was the date—the seed of the palm, *Phoenix dactylifera* The seed itself is remarkable It germinates very slowly, the endosperm is flinty hard, and the embryo, placed about the middle of the length of the seed on the side opposite the groove, is extremely small, and shows, in the ungerminated condition, no distinction into the usual parts Germination-stages of the date seed should be obtained, and put up in museum jars We are not now, however, concerned with the morphological peculiarities of the seed, but with the nature of the reserve materials contained in its endosperm

30[†] If a thin section of the endosperm of a soaked date seed is examined in a drop of water under the microscope, it will be found that the walls which chamber up the tissue into a number of compartments are extraordinarily thick

If now a similar section is made through the soft part of a date seed which has been germinating for months it will be observed that the walls, so thick in ungerminated seed, have become thin. It might be posed that this was merely a symptom of decay, but equally open to us to infer that the disappearance of substance of the walls is evidence that this substance consists of reserve food-material which serves to nourish seedling. This substance does not give the iodine reaction for starch, but if a thin section of endosperm is first treated with strong sulphuric acid and then with iodine solution (Appendix A), a blue colour, like that given by starch, is produced. This substance of the walls, insoluble in water and giving a blue colour with sulphuric acid and iodine, is called reserve-cellulose. It may be found not only in seeds of palms, but in others such as lupine and *Ononis galium*, seeds of which plants, both ungerminated and germinated, should be examined.

31. If the sulphuric acid and iodine test is applied to sections through the soft parts of plants free from starch, even to small pieces of plant tissues, e.g. grape or apple, a blue colour, like that shown by the date endosperm, is produced, and if sections are treated in this way and examined microscopically, it is found that the walls of which chamber up the tissue into compartments consist of cellulose substances. Though the cellulose of the walls of plant tissues gives the same sulphuric acid and iodine reaction as the reserve-cellulose of date and lupine seeds, it is not identical with this latter substance. The chief difference between ordinary cellulose and reserve cellulose consists in this, that the former undergoes a chemical change less readily than the latter. Indeed the function of the cellulose substances of the walls of plant tissues is not to serve as reserve food-material, but to act as a scaffolding to the compartments of these tissues.

Having made the discovery that the seeds of plants contain stores of carbohydrate or fatty material which serve in some way or other for the nutrition of the seedlings, we might conclude that our task of investigating the function of endosperm and cotyledons was at an end. But we must remember to apply the principle which we have laid down

and assumed to be true, that the modes of nutrition of plants and animals are fundamentally alike, we shall be struck by the fact that, so far, we have discovered in the seed no reserve food-substance of a kind similar to that which is so characteristic of the eggs of animals. For example, the hen's egg contains substances which have the property of setting to a solid mass (coagulating) when they are heated. We know from experience that these substances are highly nutritious, and since they disappear from the yolk during the hatching of the eggs, we may be fairly certain that they contribute to the formation of the developing chick. We must therefore first consider the nature of these coagulating substances, and then determine whether similar bodies occur in the seeds of plants.

32 We prepare a solution of white of egg by cracking the shell of a fresh egg and letting the white fall into a dish, beating it with five to ten times its volume of water, and filtering. Heat one part of the solution in a test-tube: note that a white coagulum is formed. A similar precipitate is produced by the addition of alcohol to the white of egg solution. By the use of the dialyser (Exp. 17) we demonstrate that the solution made from white of egg does not pass across the membrane, *i.e.* though soluble, it is not diffusible. Apply the following tests to samples of the white of egg solution in test tubes.

33 * *Xanthoproteic reaction* add a few drops of strong nitric acid: a white precipitate is produced, which becomes yellow on heating: add ammonia cautiously: the colour of the precipitate changes to orange (Appendix A). *Biuret reaction* add a trace of copper sulphate solution, then caustic soda or potash: a violet colouration is produced. By the substitution of ammonia for the soda or potash, a reddish violet colour is obtained (Appendix A). *Millon's reaction* add Millon's reagent (Appendix A); a white precipitate is formed, which becomes brick-red on boiling.

Iodine reaction add iodine, a yellow brown colouration is produced: the tint being considerably deeper than when a similar amount of iodine solution is added to pure water.

The group of chemical substances which have the above

properties and give these reactions is called the proteins and the chief protein contained in white of egg is called albumin (Appendix A). If, as we have already reason to believe, the proteins contained in eggs serve as reserve food-material we shall expect to find them also in the bodies of animals.

34 That they do occur in the adult body we prove by mincing fresh meat and pounding the fragments in a mortar with water. The extract is then filtered and the filtrate tested as in Exp. 33. We determine also that proteins occur in milk by diluting fresh milk with water and adding *dilute* acetic acid. Filter and test the precipitate for proteins. The curdling of milk is evidently connected with the presence of proteins.

The results of chemical analysis of the proteins show that beside carbon, oxygen, and hydrogen, they contain nitrogen and sulphur, and that some also contain phosphorus.

35 Demonstrate that proteins contain carbon, nitrogen, and sulphur thus: heat chopped, coagulated white of egg in a porcelain dish: note that it becomes dry and subsequently chars—an indication that it contains carbon. Chop finely some pieces of the white of a hard-boiled egg, dry in a desiccator (or better, use dry, powdered albumin instead, see Appendix A), grind in a mortar with soda-lime (Appendix A), transfer to a short tube of hard glass closed at one end. Heat in a bunsen flame, if necessary using the blow-pipe. Note the smell of ammonia: hold a piece of red litmus paper over the open end of the tube, observe that it becomes blue. Since ammonia (NH_3) is produced, albumin must contain nitrogen. A piece of lead acetate paper blackens when held in the fumes escaping from the heated tube, the blackening being due to the formation of lead sulphide (Appendix A), hence proteins contain sulphur. We recollect that a silver spoon left long in an egg becomes black (owing to the formation of sulphide of silver), and we recall the smell—of sulphuretted hydrogen—of a bad egg.

To study further the properties of the proteins is beyond the scope of our present work (Bibliography, 10, 11). We can only say that they may be classified according to

then properties, particularly according as they are soluble in water (albumins), insoluble in water but soluble in dilute solutions of such salts as sodium chloride, magnesium sulphate, etc (globulins), soluble only in strong salt-solutions, insoluble in any of these reagents (*e g* coagulated albumin). One substance with essentially protein-like properties, but with marked peculiarities, must be mentioned, and its reactions noted.

36 * Take a small quantity of peptone (which is obtainable in commerce, Appendix A), add water. Note that it dissolves. Heat; note that it does not coagulate. Apply the Xanthoproteic and Millon's tests for proteins. Test a solution of peptone and another of albumin by means of the biuret test. Compare the violet reaction of the albumin with the rose-pink reaction of the peptone solution.

37 * Demonstrate by means of the dialyser of parchment tube that peptone, unlike proteins, is capable of osmosis. In performing this experiment, which will last several days, owing to the fact that the rate of osmosis of peptone is slow, a *trace* of an antiseptic such as thymol or eucalyptus oil (Appendix A) should be added to a strong solution of peptone. At daily intervals, withdraw some of the liquid from the outer vessel and test a sample of it for peptone by means of the biuret test. If, owing to the diluteness of the solution, no reaction is obtained, concentrate to a small bulk and repeat the test.

We learn from our study of the proteins that they are complex bodies containing the elements Carbon (C), Oxygen (O), Hydrogen (H), Nitrogen (N), Sulphur (S)—some also contain Phosphorus (P)—that they have remarkable properties (of coagulation, etc.), that they may occur in the eggs and also in the mature bodies (*e g* the flesh) of animals (Exp 34), and that there is good reason to believe that they serve for the nutrition of animals. We have in the next place to enquire whether the seed and the mature plant also contain proteins.

38 * Cut thin sections of the tip of a root or stem of any young plant or of the young stamens of a lily or other flower. Mount in water on a slide in the usual way. Examine by means of the microscope. Run in iodine, and note that the contents of the compartments into which these

tissues are chambered give the yellow-brown reaction to proteins

39 * Repeat Exp 35, using, instead of white of egg or dry albumin, chopped and dried grass, or pea or bean meal. Note that these vegetable tissues yield evidence that they contain nitrogen and sulphur. Thus from the result of the last two experiments we infer that proteins are present in plants. Confirm this as follows. —

40 * Extract pea or bean meal or soaked and pounded wheat grains (1) with water, (2) with a dilute (5 %) solution of magnesium sulphate. Filter, and demonstrate by means of the protein tests applied to the filtered extracts that proteins are contained in the seeds of bean or pea or wheat.

41 * Put a little flour in a fine muslin cloth folded to form a bag. Hold the muslin bag under a tap, and allow water to run on the flour, which should be kneaded and squeezed between the fingers. Observe that the starch is washed away, leaving a sticky mass of gluten. Demonstrate that this residue contains proteins (Exp 33).

Hence we conclude that the reserve-materials contained in the seeds of plants are of like nature to those contained in the eggs of animals, and consist not only of carbohydrates (or fatty substances), but also of proteins.

42 * In order to determine the condition of, and the place in the seed occupied by, the protein reserve material we cut, preferably by means of a dry razor, sections through small pieces of the cotyledons of a bean or pea seed. Mount on a slide in a drop of iodine solution. Observe that, lying among the starch grains (stained blue by the reagent), there are many much smaller grains which show the yellow-brown colour-reaction characteristic of proteins. Similarly, cut sections of the dry endosperm of wheat or maize or other grass grains. Mount the sections in iodine solution, and note that the protein grains or, as they are sometimes called, aleurone grains are confined to the layer just beneath the seed coat. Apply the biuret test to other unstained sections. In adding the copper sulphate and potash to the sections placed on a glass slide. Heat, and, when cool, drain away the fluid, add a drop of water, put on a cover glass and

examine under the microscope. Treat other fresh sections with Millon's reagent and examine with a lens; note the red colour of the aleurone grain layer.

It is, however, in oily seeds that the protein grains reach their highest development. In order to prevent the solution of the water-soluble parts of the aleurone grains, we adopt the precaution of moistening with oil the razor used for cutting sections of the endosperm of such seeds, e.g. castor-oil and Brazil nut. Sections obtained from these seeds, mounted in oil (e.g. olive oil) and examined microscopically, show numerous, large, oval, transparent aleurone grains, each with a darker granule (the globoid) at one end. If such sections are compared with others cut in water, it will be seen that, in the latter, a part of the contents of the aleurone grains has dissolved, leaving behind a crystal-like body—the crystalloid—which occupies the larger part of the grain. Apply to sections cut with a dry razor the various tests for proteins. The globoid, to which reference has been made, has been proved to contain various mineral substances, e.g. compounds of calcium, magnesium, and phosphorus, and it has also been shown that these substances disappear from the aleurone grains during germination, passing, as there is reason to believe, to the embryo. Hence to our list of reserve food-materials, carbohydrate, fat, and protein must be added mineral substances, though, in most cases, seeds contain reserves of mineral substances in quantities insufficient for the requirements of the seedling.

CHAPTER IV

THE changes undergone by the reserve food materials of the seed during germination—the mode of passage of food materials from the place of storage (endosperm or cotyledons) to the place of consumption (the growing embryo)

We learned in Chapter III that even a very small seed contains a certain amount of those substances, starch (oil, etc.), which constitute reserve food-materials, and that the larger seeds possess rich stores both of carbohydrate or fatty and of protein (nitrogenous) food-substance accumulated either in the endosperm or in the fleshy cotyledons. We learned further (Exp. 13) that, in the course of germination, as the embryo develops to the seedling, the food-materials disappear from the endosperm or cotyledons. We inferred, therefore, that they are transported thence to the growing parts of the seedling.

This inference receives support from the experiment which we now perform, of germinating large and small seeds, e.g. broad beans and poppies or "California poppies" (*Eschscholtzia californica*) in darkness.

43 Sow the seeds in small pots containing ordinary soil, and place the pots in a light-proof box (Fig. 12 or cupboard). The pots must be taken out of the dark box from time to time, in order that the seedlings may be observed and measured, but they should be replaced immediately after inspection. As the experiment proceeds, we observe that, though the seeds germinate quite well, the seedlings produced in darkness are curious, pale plants, and that, whereas the bean plants live for three, four, or more weeks, the poppy or *Eschscholtzia* seedlings live only for a limited number of days. In other words, as we should guess from their appearance, both beans and poppies die

ultimately of starvation, the latter very soon, the former only after a fairly long interval of time—not indeed, as we determine by observation, till the reserve food-materials originally contained in them have disappeared from the cotyledons.

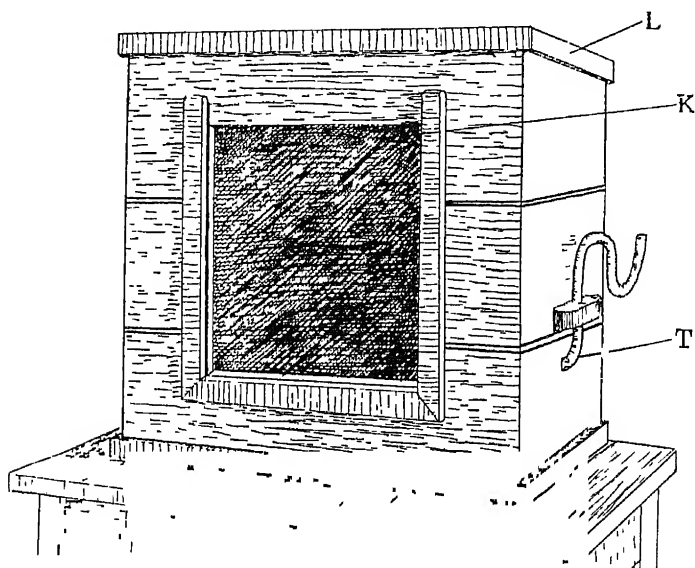


FIG 12—DARK BOX (See Appendix, p 234)

L, lid, K, shutter, T, gas tube

We cannot fail to attribute the longer life of the dark-grown bean seedlings to the large stores, and the shorter life of the dark-grown poppy or *Eschscholtzia* seedlings to the slender stores of food-materials which the seeds contain.

But if we recall the conditions in which the reserve food-materials exist in the seed, and the properties of these substances, we find ourselves confronted with a difficulty, namely, that of understanding how these food substances pass from the one place to the other—from the tissues of

the endosperm or cotyledons, to the growing root and shoot of the seedling. We know that the starch consists of solid grains which lie in compartments bounded by cellulose walls, we know that starch and oil are insoluble in water, and that much of the protein of the aleurone grains is also insoluble (Exp 42). Since the reserves are insoluble, it would appear to follow that they are irrevocably imprisoned in the tissues of the endosperm or cotyledons. Yet we have just obtained evidence (Exp 43) that the reserve-materials are liberated from the tissues and distributed to the embryo. If, however, we state the apparently contradictory facts in a somewhat modified form, we shall see that they are not irreconcilable. The reserve food-materials of the resting seed consist of solid granules (starch and protein) or of droplets of oil, while the substances are—except some of the proteins—insoluble in water, and therefore incapable of traversing the cellulose walls of the compartments in which they lie. In the active, germinating seed, food-materials reach the embryo and, at the same time, the reserves in the endosperm or cotyledons decrease and tend to disappear. Therefore, the active seed must have a means of so changing the reserve food-materials that they are no longer compelled to remain imprisoned in the tissues in which they were laid down, but are free to travel then to the embryo. It is evident that, for substances to pass from the tissues of the endosperm to those of the embryo, they must be soluble in water and capable of osmosis. If we accept this reconciliation of our facts, we proceed to ask, into what soluble and diffusible substances may the reserves of the seed be changed? For instance, into what soluble, diffusible substance may starch be transformed? The answer is suggested already by the results of Exps 18 and 25, which show that starch, boiled with traces of mineral acids, becomes converted to sugar. But if this hydrolysis of starch to sugar actually occurs in the endosperm, it certainly is not due to the agents, heat and traces of mineral acids, employed in the experiment. Granting, therefore, that our reasoning has been sound, we are driven to conclude that there must be present in the germinating seed some special agent which

like the mineral acid, is capable of effecting the hydrolysis of starch to sugar, but which, unlike the trace of mineral acid, is capable of effecting this change with rapidity at ordinary temperatures. Thus we arrive at a point in our argument at which we formulate a definite hypothesis, and put that hypothesis to the test of experiment. We state our hypothesis thus: the starch contained in the endosperm of resting seeds is converted during germination, by some unknown agent, into sugar. To test our hypothesis we ask (1) does this transformation actually take place? (2) if so, by what agent is it effected?

44 To obtain a positive answer to the first question we have only to repeat Exp. 24, viz. pound up in a mortar about $\frac{1}{4}$ of a pound of the grains of sprouting barley or wheat, add a little water, and continue the pounding. filter, and test a portion of the filtrate for sugar (see Exp. 21).

To obtain an answer to the second question, we take the remainder of the filtered extract just obtained, put it in a large beaker or bottle, add gradually, several times its volume of methylated spirit till a heavy precipitate is produced. Allow the liquid to stand for an hour or more till the precipitate has settled. Pour off the bulk of the liquid, taking care not to stir the precipitate. Transfer what remains of the liquid, together with the precipitate, to a filter. As soon as the liquid has drained away, take the filter paper from the filter funnel, open it out, spread it on a plate, scrape off the precipitate by means of a wooden spatula, and transfer it to a small, wide-mouthed, stoppered bottle. Add about 20 c.c. of distilled water, shake thoroughly, and filter, if necessary. Put the solution thus obtained into a clean bottle, add a *trace* of an antiseptic, e.g. thymol or eucalyptus oil, and label it. Before investigating the action of this extract on starch, it will be necessary to test a sample of it for sugar. For this purpose take out about 1 c.c., dilute it with 5 c.c. of water, and apply the sugar-test. If only a slight sugar reaction is obtained, it may be disregarded, but, if the test proves that sugar is present in considerable quantities, the remainder of the extract must be treated again with an excess of alcohol (methylated spirit), the

liquid filtered, and the precipitate redissolved in water the same way as before.

45 Put a very little, pure starch powder in a beaker, add water, boil thoroughly, dilute the boiled liquid, it is almost clear, take three test tubes labelled A, B, C. To A add some of the starch liquid only, to B starch liquid and about 2 or 3 c.c. of the extract prepared in Exp. 44. To C add water and 2 or 3 c.c. of the same extract. Put the tubes in a warm place at any temperature between 20° - 30° C. At hourly or convenient intervals, test the liquids in A, B, C, for starch, using the method described in Exp. 18. Determine that, whereas the blue starch reaction of the liquid in A remains, the reaction given by B gradually becomes fainter. C, which serves the all unnecessary purpose of demonstrating that the alcohol extract itself contains no starch, gives no starch reaction. When B gives only a faint starch reaction, concentrate by heating in porcelain dishes, each of the liquids in A, B, to a small bulk, and test them for sugar. The fact that A gives a well-marked sugar reaction provides us with proof that our barley or wheat extract contains the substance for which we are seeking, viz. an agent which at moderate temperatures such as those at which grains germinate, converts starch into sugar. The experiment proves, moreover, that this agent is extremely potent, the amount contained in the 2 or 3 c.c. added to B being but little.

46 Whilst the preceding experiment is going on, prepare an extract from *unsprouted* barley or wheat which has been soaked just long enough to become soft. Ascertain by similar experiments to those made in the preceding experiment, that the amount of the starch-hydrolysing agent present in the ungerminated grains is less than in the sprouted grains.

The starch-hydrolysing substance which we have extracted from germinating barley is known as diastase, and may be obtained in commerce. For subsequent experiments, commercial diastase may be used.

It will be interesting to observe diastase at work on individual starch grains. For this, use commercial diastase (Appendix A) (or our own extract, Exp. 44).

CHANGES DURING GERMINATION

47.¹ Prepare sections or a fair quantity of scrap wheat endosperm—take two watch glasses A and B, two other watch-glasses to serve as covers. To A some of the scrapings or sections and a little diastase solution, to B, add water only. Cover the watch-glass to check evaporation, and put them in a moderate warm place (temperature 20° - 35° C). After several hours mount the sections or the isolated grains of the scrap

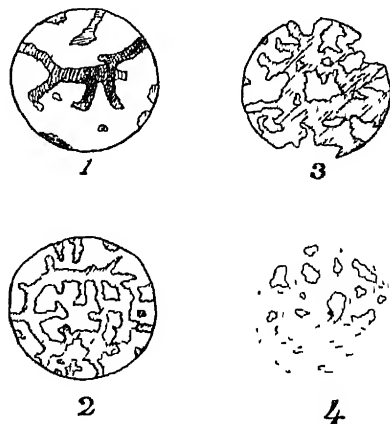


FIG. 13.—STARCH GRAINS OF GERMINATING BARLEY, TO ILLUSTRATE THE ACTION OF DIASTASE.

1, 2, 3, 4 Successive stages of corrosion of the grains. Cf. Fig. 10.
(After Strasburger, *Textbook of Botany*)

a drop of water, examine them microscopically, and observe that the grains which have been acted on by the diastase begin to show signs of change—they swell, radial fissures appear, each grain comes to have a corroded appearance and subsequently nothing remains of it but a pale oval or ghost (Fig. 13). Apply the iodine test to the grains at various stages of dissolution. (Avoid in this experiment the use of starch grains, such as those of the bean, which show radial fissures in their natural condition.)

We have thus obtained the experimental proof of

truth of our hypothesis, and may now, therefore, as a fact that the germinating grains of barley contain an agent which converts starch into sugar using other seeds, e.g. beans, peas, etc., we get diastase, the agent of this change, occurs in starchy seeds. A visit to a maltster's works will show us in the malting of barley, as carried out preliminary to brewing, a large-scale illustrative action of diastase. In the preparation of malt, grain is moistened thoroughly, and allowed to germinate. When germination has proceeded to a certain extent, the temperature of the room is raised gradually to that at which the barley is killed. The action of diastase is allowed to continue till most of the starch is converted into sugar, which serves, in the subsequent operation of brewing (see p. 58), for the production of alcohol.

Having proved that, during the germination of seeds, insoluble starch is converted into soluble fusible sugar, we conclude that it is in the fact that carbohydrates migrate from endosperm or cotyledon to the several parts of the growing embryo.

We now proceed to learn something more of the properties of diastase and determine first that it is a hydrolysing agent, not only the starch of seeds, but also that which occurs in mature plants.

48 * This we do by treating with diastase, material described in Expt. 17, sections or scrapings of the growing potato tubers. After a certain time, the grains show corrosion figures similar to those which have already been seen in germinating seeds.

We note, moreover, that diastase has no action on the cellulose of the cell walls which are to be seen in sections. The starch grains contained in leaves disappear, shown, by means of sections of leaves treated with diastase, to disappear under the action of this agent. Thus, the conclusion that whenever the plant needs to get rid of starch, it first hydrolyses it to sugar by the action of diastase. We may say that starch is the reserve form, and sugar the translocation (travelling) form.

by the carbohydrate material of a great number of plants. Before such a carbohydrate as starch—and the same applies to reserve-cellulose (Exp 30)—is used as food-material by the plant it is converted into sugar. Hence we recognise that diastase must be as widely distributed in plant-tissues as starch itself.

Now we have used more than once what we know to occur in animals as a starting point for our enquiries into the processes which take place in plants. We will here reverse the procedure and ask: has what we have learned concerning diastase and its action on starch any bearing on animal physiology? We know that starch enters very largely into the composition of animal food, and that the food which we swallow does not pass directly into the body, but into a tube—the alimentary canal—the parts of which are œsophagus, stomach, and intestines. We know further, that the food undergoes digestion, and that not till it has been digested does it pass through the wall of the alimentary canal into the blood. We suspect, therefore, that a knowledge of the action of diastase on starch may help us to an understanding of the significance of the process of digestion which food-substances, such as starch, undergo in the alimentary canal of animals. Just as insoluble starch cannot pass across the walls of the endosperm tissue, so it cannot pass through the complex wall of the stomach or small intestine, and just as soluble, diffusible sugar passes across the cell-walls of the plant, so it may pass across the walls of the absorptive part of the alimentary canal and of the blood vessels, etc., and thus, reaching the blood stream, be distributed throughout the body of the animal. It will be interesting, therefore, to discover whether diastase, which is present so generally in plants, occurs also in animals.

49 We do this in the following ways. Prepare a solution of saliva. By chewing a piece of indiarubber or by inhaling a little vapour of glacial acetic acid, a flow of saliva is induced. Collect the saliva in a small beaker or porcelain dish: add an equal volume of water. Stir thoroughly, and filter. Using the method described in Exp 45 demonstrate the diastatic action of saliva, *i.e.* its power to hydrolyse starch to sugar. A sample of commercial, salivary dias-

tase, known as ptyalin (Appendix A), should be obtained and its starch-hydrolysing properties tested.

Food is retained in the mouth for such a short time that much of the starchy material which it contains escapes action of the diastase (ptyalin) of saliva, but on reach the small intestine, the partially digested food is acted by pancreatic juice, which also contains a diastase. By action of the pancreatic diastase, the conversion of starch into sugar, which was begun in the mouth, is completed.

The diastase contained in pancreatic juice is present in the extract of the pancreas, known commercially as liquor pancreaticus (Appendix A), and the hydrolytic effect of this liquid on starch may be demonstrated by the method of Exp. 45.

The above experiments throw a clear light on what is meant by digestion. As far as starch is concerned, digestion means the conversion of this substance, by means of the special agent diastase, into soluble, diffusible sugar. They also provide yet another illustration of fundamental agreement between the nutritive processes of plants and animals. Both use starch as a food-material and both prepare this substance for distribution through the tissues in precisely the same way. By similar methods of procedure, to some of which reference will be made subsequently, it may be shown that the digestion of other substances, such as fats and proteins, means, in the same manner, the conversion of the fat or protein, each, by means of a special agent, into soluble and diffusible substances. Each of these agents is a specialist; the diastase acts on starch and starch only. We want a general name to include these specialists in chemical change. They used to be called unorganised ferments but are now known as *enzymes*. We may form a rough and unfinished picture of the mode of action of the enzyme diastase, on the starch molecule thus: imagine it holding out one hand to a molecule of water and the other to a part of the many $C_6H_{10}O_5$ groups which make up the starch molecule (see p. 33). Swinging its two imaginary hands, diastase brings the two $C_6H_{10}O_5$ molecules and the H molecule together, unites them as $C_{12}H_{22}O_{11}$ (malt sugar) and, having done so, is free to lay hold again of a wa-

molecule and another pair of the $C_6H_{10}O_5$ groups, and to treat them in a similar way. Thus we can picture diastase effecting the piecemeal disintegration of the starch molecule by hydrolysing its constituent groups. In this sense we may describe the action of diastase on starch as clastic (breaking down) and hydrolytic (water-adding). It is by similar clastic and hydrolytic processes that other enzymes act.

To return to the study of the properties of enzymes as illustrated by diastase—as we have already remarked, a little of the enzyme goes a long way in producing its specific change. Now, having regard to our description of the mode of action of diastase, is this altogether incomprehensible for, according to that description, the enzyme only plays the part of an intermediary, presenting a molecule of water and two $C_6H_{10}O_5$ molecules to one another. The presentation effected, the enzyme is free to bring about another, similar union between water and two more $C_6H_{10}O_5$ molecules. In Exp. 45 we found that even though the diastase was allowed to act on starch for several days, not all the latter was converted into sugar. That this was due not to the exhaustion of the enzyme but to the conditions of the experiment we demonstrate by repeating the experiment thus.

50 Prepare solutions of diastase and of starch (as in Exps. 44 and 45). Add a known amount of the starch liquor to a large test tube or beaker (A), and a corresponding amount to a parchment tube or diffusion shell (B), as described in Exp. 17. Add a measured quantity of the diastase solution to A and an equal quantity to B. Stand A and B in water kept at a moderate and as uniform a temperature as possible. After some hours, and then at convenient intervals, test A and B for starch. At the end of 12-24 hours replace the water in which the parchment tube dips by fresh water. Determine that, whereas A continues to give some colouration with iodine, B ceases, after a time, to give the iodine reaction. This means that, though some starch remains in the test-tube, all traces of it disappear from the parchment tube. Since the only difference between A and B is that, whilst the sugar formed by the action of the diastase on starch accumulates in A,

it passes by osmosis from B into the water surrounding the parchment tube, we conclude that the accumulation of the products of enzyme activity tends to bring the action of that enzyme to a standstill. In other words, enzymic action is inhibited by the products of that action.

By a suitable and easily planned series of experiment we may determine that diastase is very susceptible with respect to its starch-hydrolysing powers to external conditions. Thus diastase acts slowly at a low temperature, more rapidly as the temperature increases to about 60°C , and slower again at yet higher temperatures. Heated to about 85°C it is 'killed,' *i.e.* its hydrolysing powers are destroyed.

Again, diastase—like other enzymes—may be shown to be very susceptible to other conditions, *e.g.* acidity or alkalinity of the medium in which it acts.

We will now summarise what we have learned concerning the properties and mode of action of the enzyme, diastase. Diastase is a soluble, indiffusible substance (see Exp. 4) occurring in both plants and animals, among which it is widely distributed. Whenever starch is undergoing digestion, diastase is present as the active agent of that change. Starch is a reserve form of carbohydrate food-material. It is insoluble, and hence, though excellent for storage purposes, is immobile and unusable as food. Sugars are plastic forms of carbohydrate. They are soluble and diffusible, and hence are mobile and usable as food. The rôle of diastase is to convert the reserve carbohydrate (starch) into plastic carbohydrates (sugars). Its mode of action is hydrolytic and elastic. A specialist among chemical agents, it acts exclusively on starch. On that substance its action is potent given suitable conditions of temperature, slight alkalinity of medium, removal of the product of its activity. A very little diastase suffices to effect the conversion of large quantities of starch to sugar. The action of diastase on the starch contained in starchy seeds begins as soon as the latter are placed under conditions which admit of germination. For further information on the subject of diastase, see Bibliography, 10, 11.

Our study of the action of diastase on starch has served to give us a clear and precise idea of what goes

on in the course of the digestion of food by an animal. We have seen that both saliva and pancreatic juice contain diastase, and that both, in consequence, convert starch to sugar. If we extend our ideas gained from the mode of action of diastase on starch to digestion in general, we are led to conclude that digestion in animals means—apart from any mechanical action—the production, by the action of special enzymes, of soluble, diffusible, plastic substances, from insoluble, indiffusible materials supplied to the animal in the food. Similarly, applying the same conclusion to plants, we predict that just as a diastatic enzyme converts insoluble starch into soluble, diffusible sugar, so, in oily seeds, a specific enzyme effects the conversion of the oil into soluble plastic substances, and in seeds containing protein (aleurone) grains, other enzymes effect similar changes in the indiffusible and, for the most part, insoluble proteins. Not only shall we expect to find these enzymes in the germinating seeds, but also in any parts of the plants which contain reserves of food-materials liable to be drawn upon for the purpose of nutrition. This hypothesis we proceed to put to the touch of experiment.

51 * Prepare solutions of commercial diastase and of liquor pancreatici (Appendix A). Cut out small cubes of the white of a hard-boiled egg which, as we know, contains insoluble, coagulated proteins. Place a cube in each of two test tubes, and also in a diffusion shell or parchment tube standing in water, add diastase solution to one test tube and diluted liquor pancreatici to the other test tube, and also to the diffusion shell, add a drop of thymol to each of the tubes to prevent the growth of bacteria, and keep them at a temperature of 35° - 40° C. We note after some hours that, whereas the white of egg to which diastase has been added undergoes no change, that exposed to the action of liquor pancreatici becomes irregular in outline, smaller and translucent. Noting these changes, we recall the fact already discovered, that protein-like substances (peptones) exist which are both soluble and diffusible (see p. 41, Chapter III). Now, from our study of diastase, we should not be unprepared to find that if pancreatic juice digests proteins, soluble and diffusible bodies result from this digestion.

52 Therefore, after the pancreatic juice has acted on the white of egg for some time take a sample of the liquid and test it for peptones by the biuret test. The rose colour which results from the application of this test indicates that the pancreatic juice has effected conversion of insoluble, indiffusible, coagulated albumen into soluble, diffusible peptone. After 12-24 hours, add the peptone test to a sample of the liquid in the vessel which the diffusion shell (or parchment tube) dips in. If a peptone reaction is produced, concentrate half of the liquid to a small bulk and repeat the test. Note that though the liquid is shown to contain the protein-like bodies, no coagulation occurred during the operation of boiling down the liquid to a small bulk, for peptones, unlike the soluble proteins, are not coagulated by heat. It may be necessary to allow the experiment to continue several days before peptones are recognisable in the water of the outer vessel.

53 Start another digestion experiment with white of egg, or curdled milk, and liquor pancreaticus, using, however, a *very small quantity* of the substance to be digested. Before all the solid white of egg or curd of milk has disappeared from the test tube, test for peptone. Then replace the tube and test again at intervals of six or eight hours. It will be found that, as the action of the pancreatic juice continues, the peptone reaction gets fainter, and finally fails. Peptones, therefore, are not the final products of digestion of proteins by pancreatic juice. By appropriate experiments it may be shown that the peptones give place to simpler nitrogen-containing bodies, among which may be mentioned leucin, tyrosin, and asparagin (Bibliography 11). It would take up too much time to investigate the final products of the pancreatic digestion of proteins, so we may state that they have been shown to belong to certain groups of nitrogen-containing, organic substances known as the amino-acids and their derivatives (Bibliography 11).

The main facts which we have learned from the study of the pancreatic digestion of proteins are as follows. During digestion, soluble and diffusible substances are produced. The digestion takes place in two stages, in the first

CHAPTER V

THE meaning of the term Nutrition is the use which the plant makes of food substances. The germinating seed considered as a machine. The source of the power which drives the machine and the conditions under which it works.

We concluded from the results of our experiments in Chapter IV that the reserve materials of the seed are rendered soluble and diffusible by the action of enzymes and that the plastic food-substances thus produced serve to nourish the seedling. We have now to discover what is meant by the nourishment of the seedling. In other words, we have to find out to what uses the plant puts its plastic food-materials.

It would be possible to begin this work by experimenting with seedlings, but our task will be simplified if we choose not one of the more complex, higher plants, but one of the simpler, lower plants.

Of the lower plants which may be used for a study of nutrition, the most convenient is that used by brewers to set up alcoholic fermentation in sugar. The plant in question is known as yeast (*Saccharomyces cerevisiae*), and may be obtained in commerce.

55 Having purchased a small quantity of active yeast we prepare a nutrient solution, e.g. Pasteur's nutritive solution (Appendix A), in which it is known that yeast grows well. Half fill a small flask with the solution, boil it, remove the flask from the bunsen flame, plug the neck with a wad of cotton wool, and set it aside to cool. Pass a needle mounted in a pen-holder once or twice through the flame of a bunsen burner in order to sterilize

its point. Allow it to cool, take up a little of the yeast on the point of the needle and transfer it to the cooled liquid in the flask. Plug the neck of the flask again with fresh cotton wool, label it appropriately, and stand it in a warm place (25° - 30°C). After some days, note that the contents of the flask have become turbid. By mounting a drop of the liquid on a slide and examining it under the high power of the microscope, it will be seen that the turbidity is due to the presence of innumerable spherical or oval bodies, which, if the magnification is sufficiently great, present the appearance shown in Fig. 14.

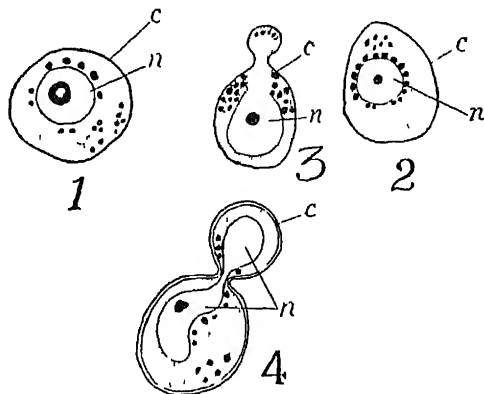


FIG. 14.—YEAST CELLS

1, 2. Living yeast cells, from a sample of brewers' yeast. Very highly magnified.
3, 4. Living yeast cells, growing actively (budding) in Pasteur's nutrient solution.
c, cell wall, n, nucleus. (After Wager, 1910)

We note that these oval bodies are of different sizes, and that some of them are in process of division or of budding off cells like themselves (Fig. 14). Hence they are evidently alive. Each oval or spherical particle which constitutes a complete yeast plant is called a cell or protoplast, and since a single protoplast is capable of nourishing itself, growing, dividing, and, in fact, leading an independent existence, we say that yeast is a unicellular plant.

56 If a drop of potassium-iodide iodine solution is put under the cover glass, the granules become of a very brown colour. Similarly, if some of the turbid liquid filtered through glass wool—or if some of the old yeast is taken and pounded up with clean silver sand in a mortar, it may be shown, e.g. by Millon's reagent (p. 33), that the pounded contents of the yeast cells contain proteins. We thus confirm the result of the iodine applied to the individual yeast cells. We conclude that a cell which, since it feeds, grows and divides, is alive, takes substances which, after being killed, give the actions characteristic of proteins. Since the amount of protoplasm borne on the needle point has sufficed to produce a relatively enormous quantity now present in the flask, it is clear that there has been a very considerable augmentation of the amount of proteins, etc., present in the old cells placed in the flask. Therefore, the living yeast supplied with appropriate nutrient materials, which include sugar and also traces of nitrogen-containing bodies, builds up proteins and incorporates them into its living substance. In this process of incorporation, both carbohydrate and nitrogen-containing materials are employed. As a result of this process, the amount of living substance is increased. Of the chemical constitution of living substance we know little or nothing. We know that when it is killed it gives rise to invariably protein reactions, and hence infer that proteins are among its essential constituents. We call the living substance protoplasm. We conclude, therefore, that the most important use to which the yeast plant puts its nutrient materials which it derives from the nutrient solution is the manufacture of fresh living substance.

Applying these conclusions to the seedlings of flowering plants, we may conclude that one use of the plastic materials is to supply the protoplasm of the seedling with materials for the construction of new protoplasm. In this process of manufacture goes on in the tissues of the seedling we already have evidence. We know that during germination the embryo increases enormously in size. We find also, from a microscopic examination of sections through the young root and stem of seedlings, that their tissues are made up of mu-

masses of substance, giving, like the yeast cell, the yellow, iodine reaction. The number of these masses is far greater in the germinating seedling than in the embryo contained in the seed. Therefore, we conclude that, just as the yeast plant consists of protoplasts which undergo division, so the tissues of the seedling consist of protoplasts which undergo division. The only essential point of difference lies in this, that, whereas the yeast cells, after division, separate from one another and constitute each a unicellular plant, the protoplasts constituting the tissues of the seedling do not separate, but remain connected with one another. In other words, the seedling is a multicellular organism and not a unicellular organism like yeast.

When we examined the sections of plant-tissues (Exp. 31'), we found that the tissues are divided by walls of cellulose into small compartments, and we recognise now that the protoplasm which gave the yellow-brown colouration with potassium-iodide iodine solution is the living thing—the protoplast, and that the cellulose walls are but a scaffolding or supporting framework to the protoplasts. Since, as the seedling grows, the number of cells increases, there must be, beside the increase of living protoplasm, an increase of the cellulose substances which form the walls around each protoplast. These walls are formed by and from the protoplasm. Part of the latter, undergoing breaking-down processes, produces cellulose, which is deposited either between two cells which have just divided, or on the already formed walls of young cells. Thus, the protoplasm makes from the plastic food-substances, not only more or less permanent additions to its own substance, but also uses up some of its substance in the production of cell-walls. Such manufacture by the protoplasm of the various substances needed by the organism for special purposes is called secretion. We may say that the cell-wall substance is secreted by the protoplasm, that diastase or any other enzyme, such as we know to be produced by plants or animals, is secreted by the protoplasm, and so on. The plastic food-substances supplied by the endosperm or cotyledons to the growing seedling provide materials which serve for the increase of the protoplasm of the seedling.

But not all this increase is permanent, for some of protoplasm is devoted to the secretion of special substances such as cellulose, enzymes, etc. Hence some of the constituent molecules of the plastic food-substances migrate from the endosperm become, in the seedling, integral parts of protoplasm, and some, having been for a time constituents of the protoplasm, serve for the formation of various secretions. Moreover, if the rate of passage of sugar from endosperm to seedling is faster than the rate at which the sugar is being used by the seedling, the surplus may be reconverted into starch and stored temporarily in the seedling in that form (see Exp. 19).

We thus obtain a definite idea of what is meant by nutrition. This process involves the incorporation of some of the constituents of plastic food-substances with protoplasm, which becomes thereby increased in quantity. It means also the provision of materials not only for permanent increase, but also for the manufacture, by living protoplasm, of the many various substances which will play a part in its life.

Now, if this were a complete account of the fate of plastic food-materials which the seedling derives from its reserve food-materials, it would mean only that the elements contained in the reserve materials pass to the seedling, and, undergoing recombinations, become parts of the protoplasm, of the cell-wall, of enzymes, and of other secretions. If this were the case, there would be, at best, merely a transmigration of matter from the endosperm or cotyledons to the seedling, and the dry weight of a seedling which has access to the reserve-materials for its supplies of food, would neither increase, nor decrease, as germination proceeded. Whether this is the case or not we can put to the proof of experiment.

The problem which we have to solve is as follows: If a seed is germinated under such conditions that it can obtain supplies of food-materials except from its reserves in cotyledons or endosperm, does the dry weight of the seedling remain constant, or is there an increase or decrease in dry weight?

We know that the germinating seedling takes up water, and, therefore, its undoubted increase in weight might

duc merely to the water which it absorbs. We know also that the cotyledons, e.g. of the bean, shrivel as they lose their reserves of starch, etc. Is the loss of dry weight by the cotyledons exactly balanced by the gain in dry weight by the rest of the seedling?

This is a matter of such importance that though the experiment is tedious, and requires fairly accurate weighing by means of a chemical balance (Appendix B), all students should attempt it.

To carry out the experiment we proceed as follows.

57 Pick out about 60 seeds of kidney bean (*Phaseolus vulgaris*) or other plant with large, epigeal cotyledons. Soak the seeds for twenty-four hours in tepid water. Select 50 seeds of approximately similar size, divide them into two lots of 20 each, reserving the other 10. Weigh the two lots of 20. Plant one lot in pots in moss or coco-fibre, and plant also the 10 reserved seeds in case any of the 20 fail. Put the pots in a dark room or box, being careful that no light falls upon them. Give water when necessary.

Squeeze out each seed of the other lot of 20 from its seed-coat, pound the seeds thoroughly in a mortar, and, when they are thoroughly crushed, transfer them to a weighed porcelain dish and dry in a drying oven at 100°C . After a day, take the dish from the oven, place it in a desiccator to cool, weigh, replace in the oven for twenty-four hours, then remove, cool, and weigh as before. We thus obtain the dry weight of ungerminated seeds (less the weight of the seed-coats, which does not concern us). After a fortnight, when the dark-grown seedlings have grown to a considerable size, turn them out of their pots, remove, by washing, all pieces of moss, etc., from their roots, select 20 seedlings, weigh, chop, and then grind them in a mortar and determine their dry weight.

The difference between the fresh weight and dry weight of the seedlings gives a measure of the amount of water taken up in the course of germination. A comparison of the dry weight of the 20 germinated seeds with the 20 similar, ungerminated seeds, shows that in the course of germination the dry weight of the seedlings has decreased.

We prove, thus, that during the work of germination

there is an actual loss of material by the plant. Not only is food-material passed from cotyledon to seedling, but also the plant, if prevented from getting food-material from other sources, loses some of the material which it contained before germination began.

In order to form an idea as to the meaning of this loss, let us imagine that we are engaged, not in the study of the physiology of plants, but in that of the physiology of motor-cars, and let us ask what we should require to learn before we could claim to understand the mechanism of cars. We should have to find out what uses the several parts serve, *i.e.* the functions of those parts, and also how they contribute to the work of the car as a whole. But we should require to know more than this, namely, the source of the energy which serves to drive the mechanism. If only from the smell left in their wake, we are familiar already with the fact that most cars consume petrol. We know also that the petrol poured into the tank of the car has to be replenished at intervals, and that, the faster the car is driven, the quicker is the petrol consumed. We conclude that the petrol furnishes the energy which drives the car. We find in the car arrangements for heating the petrol and for mixing it with the oxygen of the air, so as to cause it to undergo combustion—that is, to unite chemically with the oxygen. As a result of this combustion, the petrol disappears, being replaced by waste, oxidised substances formed by the union of the constituent elements of petrol with oxygen. It is not, however, to produce these waste substances that the motorist purchases petrol. The waste substances are the unavoidable consequences of the combustion, and have to be got rid of. It is the energy liberated during the combustion that the motorist requires and employs, by means of appropriate mechanical devices, to do the work of propelling his car. If we burn a little petrol in a lamp, we find that energy, in the form of heat and light, is produced, that the petrol disappears, and that gases, carbon dioxide and water vapour, are given off. If we shut off the supply of oxygen completely, the petrol fails to burn, if we reduce the supply of oxygen, it burns with a smoky flame—the combustion is incomplete—and some of the carbon of the petrol, instead of uniting

with oxygen to form carbon dioxide, is deposited as soot. The energy liberated during the combustion of the petrol in the lamp may be caused to do mechanical work, to convert water into steam, to make the heavy lid of a kettle containing water heated over the flame of the lamp to bob up and down—so also the energy liberated during the combustion of the petrol in the car is caused to do the work of propulsion.

Now petrol consists of a mixture of substances having the general formula, C_nH_{2n+2} , and when it is burned completely it gives rise to water and carbon dioxide. Moreover, it can be proved experimentally that a definite amount of petrol unites with a definite amount of oxygen, and, as the result of its complete combustion to carbon dioxide and water, liberates a definite amount of energy. Further, whether the oxidation is slow and carried out in several stages, or whether it is rapid and carried out apparently in one stage, the amount of energy liberated is the same. Just as we pour a pint of water from a pot, either in a rush or drop by drop, and in either case get no more and no less than a pint, so the energy liberated by the oxidation of a given amount of a substance is fixed in amount, no matter whether it is liberated in one or more stages.

If we turn our attention again to the seedling, we recognise as we note its root penetrating into the soil, its stem lifting itself into the air, its protoplasts dividing and laying down new partitions of cell-wall substance between the daughter protoplasts, that the plant, like the motor in action, is doing work. But to do work is to expend energy. From what source does the plant derive the energy necessary to carry on the work of living? The loss in dry weight undergone by the seed during germination now becomes significant. The liberation of energy by the combustion of petrol in the motor-car, or in the lamp, involves a consumption of petrol and of oxygen. The activity of the germinating seed involves a loss of substance. May not this loss of substance be associated with the liberation of the energy by means of which the plant does its work? And further, from the analogy with the working of the car, may not this energy be derived from the union of

certain substances contained in the seed with oxygen obtained from the air? Thus we arrive at an hypothesis which is capable of verification, and which, moreover, does no violence to our knowledge of seedlings. For we know already that a loss of substance, analogous to the loss of petrol in the running car, occurs in the germinating seed. We know, moreover, that seeds contain substances capable of uniting readily with oxygen—for example, the oil of seeds may be used for lighting purposes. A handful of sifted sugar thrown on a smouldering fire burns with bright flame. It seems reasonable to suppose, therefore, that the reserve food-materials supply the material for the process, akin to combustion, in the course of which energy is furnished to the plant. But if this or something like this, is the mode whereby the plant derives energy, we shall expect to find that in doing so it consumes oxygen, just as the petrol-engine consumes oxygen. Thus, we may test our hypothesis by determining the relation of seeds to oxygen. If the seed depends for its supply of energy on the union of certain combustible materials with oxygen, and if the energy so derived gives the seedling its power of doing work, then, just as without oxygen, the petrol will not burn and the car refuses to run, so, without oxygen, seeds should refuse to germinate. To determine whether or no this is the case, we proceed as follows —

58 Obtain seeds which are capable of germinating under water—rice is good for the purpose, but not always easy to get in a living state. Failing the seeds of water-marsh-plants, onion seed will serve, though it only germinates so far under water as to show the white tip of its radicle. Next, prepare a quantity of boiled water by the method described in the Appendix B. Into each of two similar small flasks or beakers, one filled with unboiled, the other with boiled (and cooled) water, put a dozen grains of rice or other seeds known to germinate under water. Pour carefully a thin layer of oil on the surface of the boiled water in order to check absorption of oxygen from the air. Observe the seeds at intervals, and note that germination does not occur in the absence of oxygen.

The following more striking method of illustrating this dependence of germination on oxygen-supply is due to Dr. F. F. Blackman —

59 Two large beakers are placed in a sink, soaked barley grains (about a dozen) are put in each, and a wide-mouthed funnel is inverted over the grains in each beaker. The beakers are filled with water so that the grains are submerged. One beaker is put under an open tap so that a constant stream of water passes down the neck of the funnel and overflows from the top of the beaker.

To the other beaker, no fresh water is supplied. Hence the grains in the first beaker receive a constant supply of oxygen dissolved in the water of the stream and those in the other receive only such supplies of oxygen as are contained in the water in the beaker, or are absorbed by the water from the air. In the course of a week or so, the barley in the continuous water-stream, although submerged, germinates and puts forth green leaves which grow to a considerable size, that in the standing water, though germination may begin, is unable, in the absence of sufficient oxygen, to produce any but the most diminutive of seedlings.

In addition to these methods, the following experiments may be employed to demonstrate that oxygen is necessary for germination —

60 Fill two gas jars with water and invert them so that their open ends are downward and beneath the surface of the water in a pneumatic trough or large dish. Displace the water by carbon dioxide (Appendix B). Impale two soaked bean seeds on a long blanket- or hat-pin. Fix the pin to the inner surface of a cork which fits the mouth of the jar, and then push the cork with pin and seeds into the open end of one of the gas jars, keeping the lower end of the jar under water during the operation. Proceed in a similar manner with the second jar, but before pushing its cork home, introduce, by means of a pipette, a little air into the jar.

Leave the jars standing in the water in the dish or pneumatic trough to prevent any air from entering, and proceed in a similar manner to prepare two jars containing each, two soaked, impaled bean seeds. Replace the water

in each jar by hydrogen (Appendix B), and, having so, allow a little air to enter one of the jars, since, how hydrogen and oxygen form, in certain proportions, an explosive mixture, the hydrogen experiment should be omitted by those unaccustomed to chemical manipulation. Instead of hydrogen, nitrogen gas (Appendix B) may be employed.

As a contrast, set up a similar gas jar containing and pass soaked beans, transfixed by a blanket-pin, into the jar. Record from day to day the germination of the seeds in the jars. From the fact that germination proceeds fairly uniformly in all the jars containing air, we infer that the various gases (CO_2 , H_2 , etc.) do not of themselves affect germination adversely, and since it proceeds not at all, or at most very slightly, in the jars from which oxygen is absent, we conclude that, as our hypothesis led us to expect, oxygen is necessary for germination.

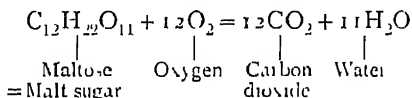
Another method which, though less direct, is interesting because of its bearing on garden-practice, may be tried.

61 Prepare enough puddled clay, by kneading clay with water, to fill two fair-sized marmalade pots. Lay a layer of the wet clay at the bottom of one pot, plant a dozen barley grains, and press wet clay closely over it till the pot is full. Cover the pot with a sheet of glass. Fill the second pot with puddled clay, and plant barley grains just below the surface. Cover each pot with a glass plate. After a week or more, investigate the germination of the barley in the two pots. (The puddled clay must be prevented from drying during the course of the experiment, otherwise it will crack, and the experiment will fail. Drying may be prevented by covering the two pots with bell-jar, under which a saucer of water is placed.)

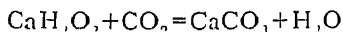
The results of the preceding experiments show that there can be no doubt that oxygen is necessary for germination. If we succeed in establishing our hypothesis, we shall have the right to conclude that the failure of seeds to germinate in the absence of oxygen is due to the fact that, no combination between oxygen and combustible substances being possible in these circumstances, no energy derived from this source is available, and no work can be carried on. But if oxygen is used for this purpose, in which seedlings are growing must in consequence

come to contain less oxygen. By taking advantage of the well-known fact that a taper burns in ordinary air, but not in air free from oxygen, we can determine whether germinating seeds extract oxygen from the air.

62 Half fill each of two wide-mouthed, well-stoppered bottles with soaked peas or barley. Replace the stoppers and put the bottles in the dark in a warm place. After twenty-four or forty-eight hours, bring *one* bottle into the room. Having lit a taper, remove the stopper, and plunge the lighted taper into the air of the bottle. The fact that the flame is extinguished gives us proof that oxygen has disappeared. Since oxygen has disappeared from the air in which the seeds have germinated, it is in a high degree probable that it has entered into combination with some oxidisable substance contained in the seeds. We will assume that this has been the case and also that the oxidisable substance is sugar. Now a sugar, when burned, gives rise to carbon dioxide and water according to the equation —



Therefore, by ascertaining whether carbon dioxide is produced by germinating seeds, we have a method of testing the truth of the assumption which we have just made. For the purpose of this test, we prepare a solution of lime-water or of baryta-water (see Appendix A), pass into a sample of either of these substances a stream of carbon dioxide from a CO_2 -generating apparatus (Appendix B), and note that the lime-water (or baryta-water) becomes cloudy. The cloudiness is due to the production of calcium (or barium) carbonate —



By blowing through a glass tube, dipping into another sample of lime- or baryta-water in a test tube demonstrate that the air expired from the lungs is rich in carbon dioxide. We now apply the test for CO_2 to the air in the second bottle of seeds which has remained in darkness. remove

the stopper, pour in a quantity of lime-water, replace the stopper, and shake the bottle. The marked turbidity of the liquid indicates that carbon dioxide (CO_2) was present in the air of the bottle in considerable quantity. Since, however, ordinary air contains CO_2 , it will be well to test a sample of air thus empty out the bottle, wash it, and then pour in lime- or barlyta-water, replace the stopper, shake, and note that the amount of CO_2 enclosed in the bottle is sufficient to produce, not a heavy precipitate like that obtained before, but merely a faint cloudiness.

The consumption of oxygen and the production of carbon dioxide by germinating seeds may be demonstrated in various other ways, *e.g.* —

63 (1) Prepare an apparatus (see Fig. 15) consisting of a glass flask (A) fitted with a rubber cork. By means of stout rubber tubing connect the side-tube of the flask with a glass tube of small bore (B), bent at right angles. Introduce into the flask a piece of moist blotting-paper and also one or two dozen barley grains which are beginning to germinate. Stand in the flask a tube containing a strong solution of potash (C), to absorb the carbon dioxide. Replace the cork, warm the flask by dipping for a minute the lower part in water heated to about 40°C and place it so that the open end of the bent tube dips below the surface of mercury contained in a small vessel (D). As the air in the flask cools, it contracts, and the mercury rises up the tube to a certain height. Mark the height to which it rises, either by making a line with india-ink on the tube or by reading off the level on a millimetric scale fixed behind the tube. Cover the whole apparatus with a box or black cloth to exclude the light, and, at intervals, record the rise of the mercury in the tube. The rise indicates that the total volume of gas in the flask and tube decreases. We know from the preceding experiments that oxygen is used, and that carbon dioxide is produced by germinating seeds. Since the carbon dioxide is absorbed by the potash, the extent of the rise of the mercury in the tube gives an indication of the amount of oxygen consumed by the germinating seeds.

Let us now review the situation to which this series of

experiments has led us. We know that a germinating seed is doing work, and therefore expending energy. We know that, in machines such as motor-cars, the driving energy is derived from the oxidation of petrol (or coal, or similar bodies). We know that, unless oxygen is supplied, the petrol cannot develop energy, nor can the seed germinate. When oxygen is supplied, the petrol is consumed and the seed loses in dry weight. When the petrol is

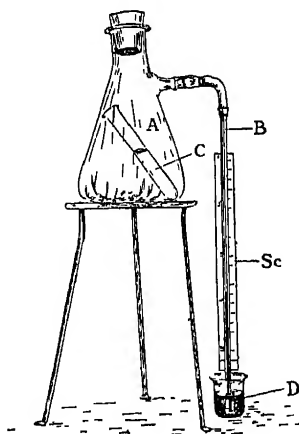


FIG. 15.—APPARATUS TO DEMONSTRATE EVOLUTION OF CARBON DIOXIDE BY GERMINATING SEEDS

A, flask, with germinating seeds, B, glass tube, C, test tube containing potash, D, mercury, Sc, scale

oxidised, waste products, such as carbon dioxide, are liberated. We cannot but conclude that the mode by which energy is obtained is similar in the motor-car and in the germinating seed, namely, by the union of oxygen with combustible materials and the consequent breaking-down of these relatively complex materials to simpler substances, such as carbon dioxide and water. Nor is it difficult to obtain evidence to show that the materials oxidised in the germinating seed are, as we have assumed, the plastic food-materials. If we look at the equation on p. 69, show-

ing the mode of oxidation of sugar, we note that number of molecules of oxygen consumed in oxid molecule of malt sugar is equal to the number of molecules of carbon dioxide produced in the process, and, since molecules occupy the same space, it follows that, in oxidation of sugar, the volume of oxygen which disappears is equal to the volume of carbon dioxide produced. We have a means of testing whether, in a starch-seed carbohydrate is the material, the oxidation of which liberates the driving energy whereby the plant-material is kept at work.

We use for the purpose the apparatus of Fig. 1, making, however, the following modifications:

64. A second flask of similar size and shape is used and placed beside that containing germinating seeds. If no potash is added to either flask, observation shows that the mercury column, though its height may fluctuate owing to temperature-changes in the air, does not move in the same manner as in the previous experiment. That such fluctuations as occur are to be attributed to changes of temperature causing expansion or contraction of the air in the flasks may be seen by observing that they occur alike in the empty flask and in that containing germinating seeds. Inasmuch, therefore, as the volume of air contained in the flask in which the seeds are germinating undergoes little or no change, we may infer that the amount of oxygen consumed is about equal to the amount of carbon dioxide liberated by the seedlings.

We conclude from the foregoing experiments that in starch-seeds, the carbohydrates supply the material which when oxidised, yields the energy whereby the plant keeps at work. Thus we have satisfied ourselves that the plant-food-substances which reach the embryo from the parent-sperm or cotyledons not only supply the protoplasmic embryo with materials out of which is formed more protoplasm, but also serve as "combustible" materials by oxidation of which the energy necessary for carrying on the life-work of the seedling is liberated. We may now inquire of the processes whereby the plastic food-material produced from the reserve-materials, and whereby the same is incorporated with the protoplasm as the nutritive pro-

or we may regard the former, digestive processes, as a preliminary to nutrition, and confine the term nutrition to the latter process only. If we adopt the second course, we require a term to include not only the digestive processes, but also other chemical changes which occur regularly in the living organism. We may use the term *metabolism* in this wide sense. To the process of oxidative decomposition of plastic food-substances—that is, the energy-producing process—the term *respiration* is applied. Since the whole significance of respiration lies in the liberation of energy, it is as absurd to say that respiration consists in the taking in of oxygen and the liberation of carbon dioxide as it is to say that a motor-car is driven by letting oxygen into the engine of the car and letting out carbon dioxide.

That the temperature at which respiration goes on in the seedling is far lower than that at which the petrol is burned in the car need not greatly trouble us. For, in the first place, chemists have shown that chemical reactions generally which go on at a high temperature, may go on, also, though at a slower rate, at a low temperature, and, in the second place, just as we found the enzyme, diastase doing at a low temperature what a trace of mineral acid could do with equal rapidity only at a high temperature, so it may be that enzyme-like bodies are excreted by the protoplasm for the special purpose of speeding up respiration.

There is still one aspect of the production of energy by the motor-car and by the seedling which requires attention. It is well known that the motor-car owes its existence to the improvements which have been made of recent years in the petrol-engine. The petrol-engine has been rendered more efficient by these improvements, that is to say, an increased percentage of the total energy developed by the combustion of the petrol is used for doing the work of propulsion. But, in spite of all the ingenuity expended on the construction of petrol, steam, and other engines, a very large proportion of the total energy produced is inevitably wasted—that is, is not used for the purpose which the engine is designed to serve. This wasted energy takes the form of heat, and the fact that it is wasted is due, of course, to no lack of skill on the part of engineers, but

to the properties of matter, e.g. to friction and to the power metals have of absorbing, conducting, and radiating heat, and so on. If our comparison of the plant with a machine is just, we shall expect to find that, of the total energy produced in respiration, a large proportion is not employed in driving the plant's machinery, but appears in the form of heat. That this is the case may be shown in a striking manner by the use of the familiar Dewar flasks (Thermo flasks) (Appendix B). As is well known, a hot substance put into a Dewar flask remains hot for a long time, and a cold substance remains cold.

65 Soak in tepid water for twenty-four hours enough peas to nearly fill two Dewar flasks. Wash the seeds in running water, and reject any which seem bad. Remove superfluous water by means of a clean cloth or blotting-paper. Nearly fill one flask with the seeds; whilst filling, insert a small thermometer in the flask so that the part of the stem which records temperatures above 30°C only remains visible. Push a large wad of dry cotton wool into the neck of the flask to close the neck and hold the thermometer in place. Wrap the whole apparatus in a large, loosely-fitting jacket of cotton wool. Boil the remaining peas, and, adopting similar precautions to those used with the unboiled peas, put half their number in the second Dewar flask with a thermometer and wad and covering of cotton wool. Hang up a third thermometer in an ordinary flask to give the air-temperature during the course of the experiment. During several days, and at as regular intervals as possible, record the temperatures registered by the three thermometers. Observe that the germinating peas produce sufficient heat to raise the contents of the flask to a temperature as much as 10°C , or more, above that of the air. Note that the temperature of the flask containing the boiled peas is for some time little above that of the air, but that, after a day or two, it may also show a rise of some degrees. If it does, turn out the contents of this flask and observe that the peas have become mouldy. Now take the remaining soaked peas and warm them for a few minutes in a beaker of water to which a trace of corrosive sublimate has been added. It should be borne in mind that

corrosive sublimate is in the highest degree poisonous that only a mere trace need be added to the water in the beaker, and that the vessels which have contained it must be thoroughly washed immediately after use. If there is any objection to the use of corrosive sublimate, the peas may be boiled in a solution of permanganate of potash instead. Rinse out the Dewar flask with a solution of corrosive sublimate or permanganate of potash. Transfer the poisoned peas, after wiping them between sheets of blotting-paper, to the clean flask. Arrange the thermometer and cotton wool as before, and determine that no mould being able to grow, no rise of temperature occurs. We learn from these experiments that germinating peas liberate a considerable amount of heat, and conclude that, just as in machines not all the energy is utilised for doing mechanical work, so, in germinating seeds, some of the energy developed in respiration appears in the form of heat, and is therefore not used directly in doing vital work. We learn, incidentally, that low forms of plant-life, such as moulds, also produce heat, and we suspect that they too respire, and that they, like the seedling and the petrol- or steam-engine, are of only limited efficiency. Under more natural conditions than those which obtain in a Dewar flask, the temperature of germinating seeds would, of course, not rise so high, for heat would be lost by conduction to surrounding objects. Although we may regard the heat produced during germination as energy lost, in the sense that it is not applied to the performance of the work of the seedling, yet the heat may be serviceable in other ways, namely, in providing a temperature at which the organism can work at its best. If the petrol in a car is frozen, the car cannot be started if the temperature of the car rises above a certain point damage ensues. In other words, the mechanism of the car works only within a certain range of temperature. That this is also the case with the seedling, we have already determined in part, for we know that seeds heated to the boiling point of water are killed.

66 By exposing seed in germinators to different temperatures, e.g. freezing point, room temperature, 25° - 30° C. and so on, we may determine that germination proceeds

best at a certain (moderately high) temperature. Like machines in general, the plant-mechanism would under certain definite temperature-conditions.

The facts we have learned respecting the nutritive respiration of seeds are fundamental facts. They tell us that the living substance (protoplasm) owes its increase to the plastic food-materials, and they tell us the source and mode of liberation of the energy with which the living mechanism works. Therefore, if we hold fast to our general hypothesis which has helped us so far already, and according to which the vital processes of plants and animals are fundamentally similar, we expect to find that the modes of nutrition and of respiration of the mature plant and of the animal are those of the germinating seed. The proof that this is the case with respect to nutrition we will defer to Chapter VI, and deal at once with respiration. With respect to air the necessary proofs lie close at hand. We have already traced the plastic food-materials to the blood-stream and know that the multitudinous protoplasts which make up the body are bathed with fluid derived from that source. We are aware that the body of the animal is constantly doing work, and therefore expending energy. We know moreover, that, in the warm-blooded animals, heat is produced in sufficient quantity to maintain the temperature of the body at a remarkably constant level and considerably above that of the surrounding air. We breathe on a dry day on a piece of glass and note the water which comes from our breath; we blow through a tube into lime-water and demonstrate, by the precipitate of calcium carbonate which forms, that the air we expire contains carbon dioxide. Putting these facts together, we recognise that the nutrition of animals is similar to that of plants. The apparent difference consists in this, that the waste products of animal metabolism include not only water and carbon dioxide, but also nitrogenous substances—urea, uric acid, etc. Since these substances are evidently derived from proteins, we are bound to infer that, in addition to carbohydrates and fats, proteins also contribute, by their decomposition, to the energy required by the organism for the purposes of its life-work. In point of fact, pro-

decompositions similar to those which take place in animals are known to occur in plants. We may perhaps express the facts thus: the bulk of the energy which is used by plant and animal is derived from the oxidative decomposition of carbohydrates or fats. In the course of its activity, the protoplasm itself wears out, and so produces waste nitrogenous substances. If plentiful supplies of proteins are at the disposal of the organism, as is the case with many animals and some plants, the proteins may be respired, *i.e.* they may be split up, certain of their products oxidised, and thus release energy. The only difference between the animal and plant with respect to these energy-releasing processes appears to be that, whereas many animals are protein-spendthrifts, most plants are protein-misers. This is a matter which, though interesting, is not essential, and need not therefore detain us. With respect to the respiration of the adult plant, we convince ourselves by the following experiments that what we have learned from the seedling applies word for word to the adult.

67 Grow two bean plants, each with its roots in tap-water contained in a wide-mouthed jar. When they have passed the seedling stage compare the plants as to root and shoot development, and then remove one of them from the jar and place it with its roots in another jar containing boiled water (Appendix B). A layer of oil is then placed on the surface of the water in this jar. The growth of the "transplanted" plant at once falls behind that of the untouched plant. In the absence of oxygen from the water the energy necessary for the growth of the root is not developed. By repeating Exp 62, using the opening buds of horse-chestnut or dandelion, etc., we prove that, like the seedlings, these parts of the mature plant absorb oxygen and evolve carbon dioxide. By the method of Exp 65, using buds or young leaves, we demonstrate that their respiration results in a production of heat. Thus we conclude that the energy-liberating processes, which we include under the term respiration, run the same course in both young and adult plants, and in animals. We conclude, in fact, that this process is general in all organisms, and that if there are any organisms in which respiration

does not take place in this way, they must be regarded as constituting special cases, and must receive special consideration

CHAPTER VI

THE seedling as an independent plant the lowest forms of plants and animals and the lines followed in the evolution of the higher plants and animals The distinguishing characters of root- and shoot-systems The mode of growth of the root the functions of its parts the root hairs, the absorbent organs of the root

THERE comes a time when the seedling, reared hitherto at the expense of the reserve-materials of the seed, cuts itself adrift from the latter and sets out on its career as an independent plant That career we will now follow The independent seedling grows, forms new members, performs remarkable movements, and becoming a fully grown-up plant, bears flowers and sets seed For these operations, it must be able to obtain large supplies of material, and, for them, no small amount of energy is required We have learned by experiment what substances the young plant and the animal use in constructing their tissues and how they obtain their supplies of energy, so, till we find that the hypothesis is wrong, we will maintain that the mature plant uses similar food-materials for its nutrition Since, however, the substances, such as sugar, proteins, etc., which, on this hypothesis, constitute the true food-materials of organisms, are not present in the soil or in the air, it follows that either our hypothesis is wrong, or that the mature plant manufactures these plastic food-materials from other substances which it obtains from the soil or air, or from both these sources

68 The following experiment affords evidence in support of our hypothesis —

Germinate two lots of radish seeds on damp blotting-paper When the seedlings have emerged, grow one lot with the roots in clean tap water, the other lot in tap

water to which has been added a *trace* of potassium nitrate. It will be found that the seedlings with their roots in the water to which potassium nitrate has been added grow far more vigorously than those whose roots are in ordinary water (Fig. 16).

What relation, if any, exists between the absorption of such a mineral salt as potassium nitrate (KNO_3) and the formation of plastic food-materials need not concern us now. We are content to note the evidence supporting the view that the root absorbs substances from the soil, and that these substances contribute in some way to the formation of plastic food-materials on which the growth and activity of the plant depend. Recognising, as we must, the vital importance of plastic food-substances, it will be evident that the first and constant care of the plant must be to obtain adequate supplies of the raw materials which it needs for their construction, and, if we bear in mind what we have learned already with respect to the phenomenon of adaptation—the automatic adjustment of the members of an organism to their special work—(p. 22) we shall expect to find that the plant-members concerned in the absorption of the raw materials of the food show adaptations tending to fit them for this work. That is, these members, or parts of them, present appearances which we may recognise as fitting them to serve as *absorptive organs*. Hence, by a careful and intelligent inspection of the members of a plant (roots, leaves, etc.), we shall obtain broad hints of the special functions which these members or their parts perform.

Our task, therefore, in this and in succeeding chapters is to discover what are the raw materials of the food, how they are absorbed, and how, from the raw materials, the plant manufactures the finished articles, the plastic food-substances such as sugar and proteins, on which the protoplasm subsists and on which, therefore, all growth and activity depend. In studying these problems, we shall learn how profoundly the form characteristic of plants is determined by the necessity under which they labour of producing organs suited for the absorption of the raw materials of the food. We shall even discover that in this necessity lies the origin of the remarkable

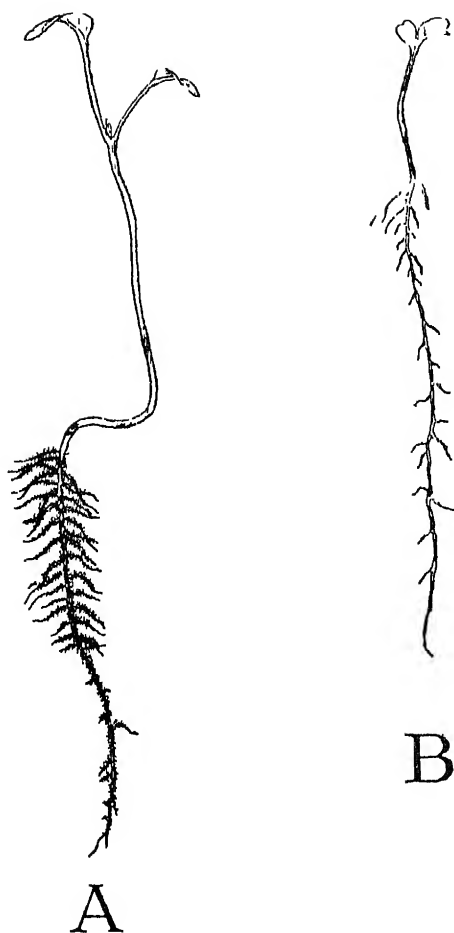


FIG 16

A, Radish seedling grown in tap water, to which has been added a trace of potassium nitrate; B, radish seedling, of the same age as A, grown in tap water only

K P

B

differences in form which distinguish the higher plants from the higher animals. Between animals and plants there are, as the evidence of former and later chapters proves, no fundamental differences except those arising from the obligation laid upon the plant to make its own plastic food-materials. The modes of nutrition of the protoplasm are identical in plant and animal; the respiratory processes are similar; the movements, though more conspicuous in the animal, are alike, and the modes of behaviour to external influences are the same. All the wonderful differences in form which stand out when we compare the animal and vegetable kingdoms are the outcome of this great fact that plants make their plastic food-materials, animals appropriate them ready-made. It is a fact of the profoundest physiological importance that "all flesh is grass."

Let us, in order to illustrate these remarks, compare one of the lowest plants with a very simple organism which, though undoubtedly an animal, stands very near the parting of the ways which lead respectively towards the animal and the vegetable kingdoms. The plant may be obtained often in great numbers in ponds, water-butts or rain puddles; the animal is found also in these situations. If water from one of these sources is stood in a glass dish near a window the presence of the plant and of the animal may be revealed by a green scum appearing on the side of the dish near the window. On microscopic examination of a drop of the liquid, minute green organisms in active, dashing movement may be observed. Now and again one of them rests awhile and then may be examined in greater detail. It consists of a single protoplast the greater part of which is not colourless like that of the yeast plant, but green, the green part having a somewhat cup-shaped form (Fig. 17 A). In the narrow cavity of the "cup," the protoplasm is colourless, and, by suitable modes of preparation, a denser, more or less spherical part of the protoplasm—the nucleus—may be seen lying toward the lower end of this neck of protoplasm. At the free end of the neck is a slight projection or knob, from which extend two protoplasmic threads (flagella), each as long, or longer, than the body. It is by means of the flagella

that the organism rows itself along. On one side of the body, near the anterior end, a red speck—the eye-spot—may be seen; and suitably stained microscope-preparations show that the protoplast is enclosed by a delicate wall of cellulose, through which only the flagella project. By hunting through drops of water obtained from puddles of rain water—and if the student has a microscope at his disposal the hunt is well worth making—examples of the second organism may be obtained. It consists of a single green protoplast, which has the form of an elongated cup or flask (Fig. 17 B), but instead of the narrow neck being occupied by colourless protoplasm, it is hollow, and thus forms a gullet. Flagella, attached near the bottom of the gullet, project beyond the body of the animal. As in *Chlamydomonas*—the plant previously examined—so in this animal, *Euglena viridis*, a nucleus and an eye-spot are present, but, unlike that of the plant, the enclosing wall of the body of *Euglena* is not of cellulose. Upon occasion, both *Chlamydomonas* and *Euglena* may lose their green colour, become colourless and yet increase in size and even multiply by division. So alike are the two organisms, that it might seem mere hair-splitting to describe the one as a plant and the other as an animal. Nevertheless, it is proper and useful to distinguish them in this manner, for, it is possible to imagine that, by a long series of changes, all of the nature of so many adjustments to the more adequate winning of raw material from water or from land, or to the more efficient elaboration of plastic food-substances from this raw material, the higher plants, even the highest flowering plants, have been derived from a *Chlamydomonas*-like ancestor. So it is conceivable, also, that all animals have been derived from some lowly form not far removed from *Euglena viridis*. What then is the essentially plant-like character of *Chlamydomonas*, and what the essentially animal-like character of *Euglena*? *Chlamydomonas* encloses its protoplast within a definite wall of cellulose. *Euglena* has a sac or gullet open to the outside. Hence *Euglena* can engulf solid particles, digest them, and so obtain plastic food-substances. *Chlamydomonas* debais itself from all solid food supplies, for only substances in

solution can pass its wall. Euglena can, upon occasion, prey on other organisms, and pick up nutritious morsels. Chlamydomonas cannot. Chlamydomonas, by surrounding itself completely by a wall, condemns itself to the labour of constructing its plastic food-substances from such soluble raw materials as its environment provides. Euglena, though

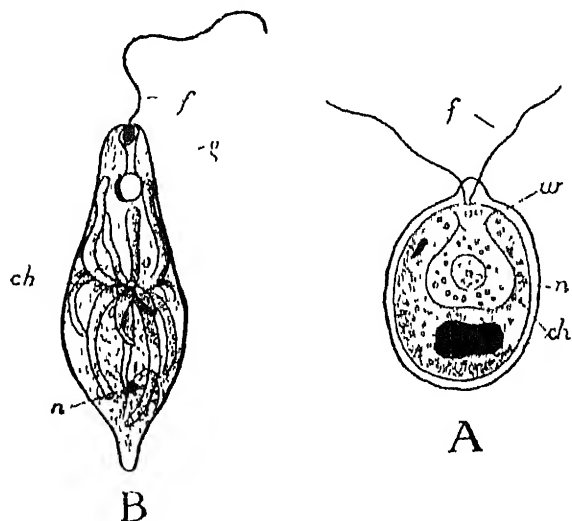


FIG. 17

A, *Chlamydomonas* sp.; B, *Euglena viridis*. (Highly magnified.)

w, wall; ch, chloroplast; n, nucleus; f, flagella; g, gullet.

it has not lost this power altogether (see Chapter XI), can swallow any likely particle, and thus get its food ready-made. These are the salient, differentiating characters not only of *Chlamydomonas* and of *Euglena*, but also of plants and animals in general. It is the investing wall which determines the mode of life of *Chlamydomonas*; it is the gullet which makes *Euglena* free of all solid food-supplies which it can engulf and digest. It is the fact that *Chlamydomonas* has a continuous solid cell-wall, which is the essential fact, not that the wall is of cellulose,

though, since it is of cellulose and since animal protoplasts have no cellulose walls, it is sometimes said that the cells of plants are characterised by cellulose walls, and those of animals by the absence thereof. Evidently this is true, but evidently also the wall is the thing, the chemical nature of the wall is a matter of less importance.

We may summarise the whole difference between *Chlamydomonas* and *Euglena* thus. *Euglena* would, if it could swallow it, feed on *Chlamydomonas*, *Chlamydomonas* cannot swallow, and, therefore, cannot feed upon *Euglena*, nor on any other undissolved substance. For "*Euglena*" write "animals," and for "*Chlamydomonas*" write "plants," and the statement holds good. Animals feed on plants, plants cannot feed on animals unless, as happens in the case of certain parasites, a plant gets access to the soluble substances contained in the body of an animal.

Our excursion into the world of microscopic plants and animals has been of use, not only in presenting our immediate problems to us more definitely, but also in enabling us to take a more comprehensive view of the plant and animal kingdoms.

We return now to the work of flowering plants, and proceed to enquire into the nature and mode of absorption of the raw materials of their food.

It is a significant fact that plants in general live, as it were, in two worlds. One-half of the plant leads a subterranean life, the other half an aerial life. It is noteworthy, also, that the underground root-system differs markedly in form and aspect from the above-ground shoot-system.

69 A careful examination of a young plant, e.g. sunflower or bean, etc., dug up from the soil or turned out from its pot, or, better, raised for the purpose in a box of coco-fibre or sawdust (Appendix B), shows us that, whereas the green shoot-system consists of a main axis bearing two kinds of lateral members, namely, lateral axes (branches) and leaves, the colourless, or non-green, root-system consists of a main axis (tap-root) bearing lateral members (lateral roots) of one kind only. Moreover, an examination of the lateral members of the root and shoot shows that they differ from one another with respect

to origin, structure, and behaviour. Note, for example, each lateral shoot begins as a bud standing in the axil of a leaf (see p. 18), that these axillary buds arise as somewhat superficial outgrowths from the main axis, that the main and lateral shoot-axis terminates in a bud, that such a bud may develop into either leaf-bearing or flower-bearing branches, now observe that the lateral roots do not arise from buds, but push themselves out from deep-lying tissues of the main root, and that, instead of terminating in a bud, each lateral root, like the main-root, has over its tip a thin, brownish, thimble-like covering called a root-cap, nor are flowers borne on the root.

70 Beside observing these morphological differences between the members of the root- and shoot-systems we prove by the simple experiment of turning a plant on its side and leaving it in a horizontal position for several days that remarkable physiological differences also exist between them. Whereas, in consequence of this change of position, the main stem or ascending shoot curves till its young tip points once again vertically upward, the main root (the descending axis) curves in precisely opposite manner till its tip points vertically downward.

Whilst drawing the specimen, we note, also, that the lateral members of root- and shoot-systems exhibit characteristically different curvatures as the result of this change of position.

When we study the curvatures of root and shoot we recognise that they serve to bring the root into the soil and the shoot into the air. If we observe an animal making similarly purposeful movements, for instance, a worm thrusting its head out of its burrow, we say that the movements are instinctive. Now, in point of fact, certain fixed animals perform movements, in response to change of position, quite like those of the root and shoot of our plant. Hence there is every reason for us to regard the upward and downward bending of shoot and root respectively as examples of nervous actions of the same order as those which in animals are spoken of as instinctive. Like instinctive actions in general, the root- and shoot-curvatures serve evidently to secure the well-being of the organism. The well-being of the plant

then, is secured by the penetration of the root into the soil and by the exposure of the shoot to the air. In other words, the shoot has work to do which can best be done in the air, the root has work to do which it can perform best in the soil.

Thus we come to ask, what is the nature of the work performed by the root-system? The most commonplace observations suffice to convince us that the root-system of a plant serves to fix it firmly in the soil. The tree resists all but the most violent hurricanes, the place on the wall where last year's ivy grew is marked by the remains of the roots, which broke rather than relax their hold when the ivy was torn away, the delicate root of the tender seedling is held so firmly in the soil that the pull of the shoot extricating itself from the seed does not disturb it. We must discover by what peculiar mode of growth the root, though elongating all the while, maintains its hold upon the earth, and so serves as a support to the plant.

71. To this end we choose several germinating bean seeds, the roots of which are about $1.1\frac{1}{2}$ inches long. The seeds for the experiment may be germinated either in moist air, or, since roots often fail to grow straight in air, in boxes of moistened coco-fibre or of sawdust. Dry with blotting-paper the surface of a straight root about $1.1\frac{1}{2}$ inches long, and, by means of the method given in Appendix B, make a series of 10-12 india-ink marks at distances of about 1 mm from one another, starting from the tip of the root. Lay a millimetre scale beside the root, measure, and record the distances between the consecutive marks. Pin the seedling in moist air, or, when the ink is dry, replant it in a seedling "observation-box" (Appendix B) in moist coco-fibre or sawdust, with its root vertically downward and so near to the glass face as to be visible from the outside of the box. Measure, at intervals of twelve or twenty-four hours, the amount of elongation of the several marked zones. Tabulate the results and plot them in the form of a curve. Determine that the elongating region begins about 1 mm behind the root apex, extends over a very short section (10-20 mm), and ends near that part of the root where the root-

hairs begin. Note also that, further back, the root, tho it has ceased to grow in length, increases in thickness. Measure the distance from the apex of the region of maximal elongation.

72 Cut longitudinal sections through the root-tip through the region of greatest elongation—note (if necessary staining with iodine) that the cells of the former are small, and that the protoplasm occupies the whole of cell, that the cells of the latter are large, that the protoplasm forms a layer against the cell-wall, and that it encloses a large space (vacuole) containing liquid (cf p. Fig. 20). Observe that, when treated with iodine, protoplasm, which is killed by the reagent, contracts away from the wall, and, staining yellow or yellow-brown, comes readily recognisable. The extreme tip just behind the root cap, composed of small cells with much protoplasm, constitutes the formative region, i.e. the region where the protoplasts are undergoing division, and thus forming new protoplasts. Behind the small formative region (of about 1 mm) is the region of elongation (some 20 mm in length), and in this region, where so cell-division also occurs, the increase in length of the whole root is effected by that of the constituent cells. We can now understand how, from the very start, the growth of the root makes for the steadiness of the plant and secures its fixation. The old part of the root—the part in connection with the stem—having ceased to elongate, is wedged, as in a vice, in the soil. Nor does the tip elongate. The growth in length of the region behind the tip rises in a pressure which, since the apical region is free to move, drives the tip through the soil, the shortness of the elongating region serves to prevent that region, in the event of the tip meeting with considerable resistance, from being itself bent. Each day, a section ceases to elongate, each day, a newly formed region just behind the growing point begins to increase in length. Somewhat in the way that a worm, fixed by its tail-end, pushes, by the elongation of its body, its head from a burrow, a root pushes its way through the soil.

73 By the use of a seedling observation-box, on the glass front of which we trace each day the outline of

roots, we follow the course of elongation of the main and lateral roots of a bean or pea seedling.

We learn from these observations that each root consists of the following regions: a formative region (growing-point) covered by the root cap, a region of elongation, a region of thickening, the lower part of which constitutes the root-hair region.

74 In order to compare the mode of elongation of the shoot with that of the root, mark the young shoot of a seedling bean or pea, from base to apex, with a series of india-ink marks about 8 mm distant from one another. Proceeding as in the case of the root, determine the amount and distribution of the growth of the marked shoot. We thus discover that the growing region of the shoot is as extensive as that of the root is limited: that, in the young stem, parts far distant from the apex, as well as parts near it, are in active elongation; that the parts which elongate most rapidly are situated between the nodes, *i.e.* the places of insertion of the leaves, and that therefore these internodal regions come each to be of considerable length. The stem in cleaving the thin air encounters but slight resistance, and is therefore under no necessity, as is the root, to concentrate its elongating region.

The significance of the root-cap now becomes evident. Behind it, lies the delicate growing-point, in front, are grains of sand, small stones and other hard substances of the soil which the tip is bound to encounter. As a thimble protects the finger, so the root-cap protects the growing-point of the root. It can be shown, moreover, by simple experiment that the root-tip has another and more subtle means of protecting itself from mechanical injury.

75 Plant bean seeds in a seedling observation-box containing, beside coco-fibre, a layer of small stones on the top of which more coco-fibre, in which the seeds are planted, is laid and pressed down. Follow the behaviour of the roots when they come to the layer of pebbles. Note that, when obstructed, the root curves in such a way that the tip is first moved away from the obstruction, and then, brought again into the vertical position, resumes its downward progress. Determine that the curvature in

response to contact and that which results in the restoration of the vertical position, take place, not at the itself, but in the region of elongation. Some of plants raised for the purpose of the above experiment may be allowed to grow on, in order that the origin and order of development of the lateral roots may be observed. As already indicated, the lateral roots are identical in structure with the main root, and grow in precisely the same manner. Thus each lateral root, wedged like the main root firmly in the soil, contributes to the firm hold which the whole root-system has on the soil.

76 Onion or hyacinth bulbs, started into growth in hyacinth glasses, or similar vessels filled with water, provide objects for the study of the development of another type of root-system. Observe that a number of independent roots which, since they arise on the bulbous stem (arise), are called adventitious roots, make their appearance and grow all at about the same rate. Examine the root-systems of several Dicotyledons and Monocotyledons, refer them to the bean type (tap root-system, cf. Fig. 18) or to the onion or lily type (adventitious root-system, cf. Fig. 19). Note also cases (e.g. sunflower) in which the tap root soon ceases to grow, its place being taken by numerous, lateral roots (fibrous roots). Observe that plants with a tap root are deep-rooting (mallow, etc.), that with the adventitious types of root are more surface-rooting. Note the curious behaviour of the roots of various rosette plants (dandelion and plantain), which, after penetrating the soil to a certain depth, shorten so much as to throw the outer tissues of their older parts into transverse folds, and observe that, as a consequence, the rosette leaves and the epicotyl are dragged down close to the ground. Note, on a lawn, how the plantain and the dandelion, by this root-contraction, press down the grass in their neighbourhood, and so secure space and light for themselves.

77 Dig up a dandelion plant. Cut its roots into short lengths. Plant the pieces of root in moist, sifted soil. Observe that, after some weeks, each root has developed into a plant. Follow the process by which this comes about. Note that the wounded, cut surfaces are made smooth.

healed by the formation of tissue, called callus or wound-tissue, that, later, roots develop from the new tissue of the lower end, and that buds, which are called adventitious buds, develop from the new tissue of the upper end. We conclude, therefore, that the morphological "points" (p. 85), which distinguish roots from shoots, *e.g.* that the root does not bear buds, are the expression of general, but not of irrevocably fixed habits on the part of these members. We note, in confirmation, that many trees, *e.g.* elm, may throw up suckers from injured places on their more superficial roots.

78 In order to study further the process of healing of wounds and the formation of adventitious roots from callus-tissue, cuttings of geranium, willow, rose, etc., should be made, planted, and examined in their different stages of growth. Mount a typical series of such cuttings, and add them to the museum. Observe and record the healing of wounds made when large branches are lopped from trees. If possible, obtain a series of photographs of such wounds and of the appearances they present in the slow course of their healing. Put the date and other details on the photographs, mount them, and add them to the museum.

Having studied the fixing function of the root-system, we must next direct our attention to an even more fundamental part of its work, that of absorbing substances from the soil. From general knowledge of plants and gardens, such as the wilting and dying of plants in periods of drought, and from the large percentage of water contained in plant-tissues (see Exp. 5), we are bound to infer that a plant absorbs water from the soil. We shall have an opportunity later (Chapter ix.) of determining the quantities of water taken up by the roots, and will now confine ourselves to enquiring how the absorption of water is effected.

When growing hyacinth or tulip bulbs in water in hyacinth glasses for the purpose of Exp. 76, it may have happened that, owing to neglect in keeping the glasses quite full of water, the later-formed roots found themselves compelled to grow some distance through the air before reaching the water. If such was the case, it may have



FIG 18

Broad bean (*Vicia Faba*) Root system of seedling, showing elongation of the radicle to form a tap root (*br*) with numerous lateral roots (*lr*)

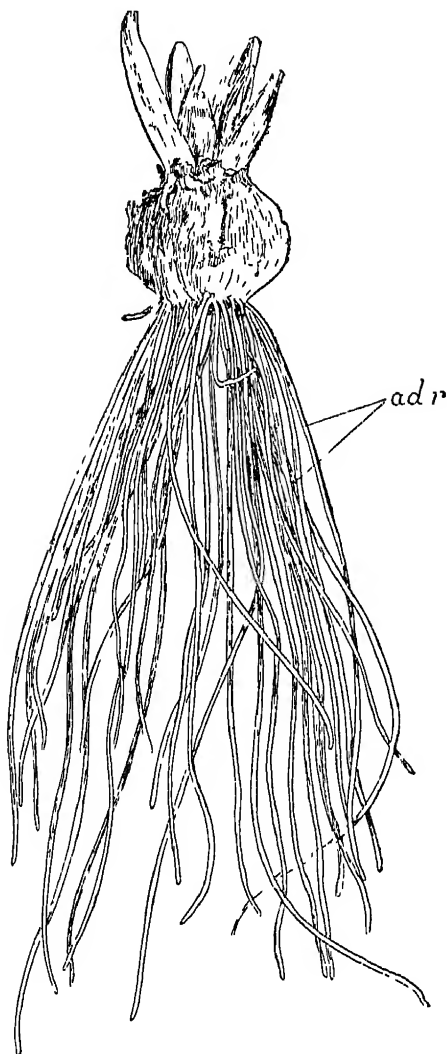


FIG 19

Hyacinth (*Hyacinthus orientalis*) Bulb with adventitious root system,

been noticed that, whereas the surfaces of the roots which were submerged from the first are quite smooth, those of the roots which grow through the air are covered with fine, white root-hairs.

In order to obtain material for the study of this phenomenon, we germinate oat or bean seeds, some with their roots in water, others with their roots in moist air.

79 * Make transverse sections across the beginning of the root-hair region of a root growing in air, mount in water, and examine under the microscope. Observe, with low and high powers, the root-hairs in all stages of development. Put the preparations aside and make others through the corresponding regions of a root grown in water which presents to the naked eye no sign of root-hairs. Determine, by microscopic examination of the sections, that slight, dome-shaped outgrowths from the superficial cells occur, and that they are identical with those on the youngest part of the root-hair region of the root grown in moist air.

80 Germinate two lots of maize or oat grains, one, so that the young roots are in moist air, the other, so that they are in moist sand or sawdust. Contrast the development of the root-hairs in the two cases. If, now, we assume that the plant possesses a certain power of regulating the development of its organs in accordance with external conditions, we are led to infer from these facts that the root in water or thoroughly wet soil, being able to carry on its work without their help, does not put itself to the trouble of forming root-hairs. Since, however, countless generations of plants have required root-hairs and have formed them, these plants—like land-plants in general—have acquired the fixed habit of producing root-hairs. Hence a hyacinth or bean root in water is, as it were, the subject of divided counsel. The ingrained habit urges the formation of root-hairs. They begin to form. The presence of much water renders root-hairs unnecessary. They cease to develop. We may suppose that the formation of the incipient root-hairs is due to habit, but that the root, before it obeys the dictates of habit, requires a special stimulus, perhaps, in this case, a slight drying of its surface. In water, this stimulus is lacking, and the

root-hairs never get beyond the rudimentary stage. The bodies of plants and of animals contain many rudimentary structures. the calyx of the flowers of the daisy family is often represented by mere hair-like structures. whales and various snakes have hind limbs which are ridiculously small, and are never used. our ears have sets of muscles which do not work. the vermiform appendix, the rudiment of an accessory stomach, is a nuisance. The hypothesis just suggested to account for the rudimentary root-hairs of the water-roots of hyacinths helps us, when applied to these cases, to form some idea why rudimentary organs may outlive their uses. The habit of leaving them down belongs to the race. the business of their further development, to the individual. This further development depends on some precise stimulus either from without or from within. In the absence, more or less complete, of this stimulus, the organ remains rudimentary.

Our observations on the presence or absence of hairs on the hyacinth roots lead us to conjecture confidently that the root-hairs are organs which serve to secure the absorption of ample supplies of water. We argue thus. roots grown in water have access to plentiful supplies, roots in moist air to but meagre supplies of water. If the special business of the root-hairs is to secure an adequate amount of water, then, in the case of water-roots, root-hairs will have but little to do, but in the case of air-roots, their water-absorbing capacities will be exercised to their full extent.

81 We test this hypothesis by an examination of the roots of various water-plants, and discover that, in the large majority, e.g. water buttercup (*Ranunculus aquatilis*, etc.), root-hairs are absent. though, in the case of some aquatic plants, such as the frog-bit (*Hydrocharis Morisus-ranae*), root-hairs are produced in large numbers. (The aquatic plants collected for this purpose may be examined also with respect to their shoot- and root-systems, which should be compared with those of land-plants, and, with examples of the latter, dried and preserved in the museum, together with notes of their several characters.)

We may take it, then, as probable that the function of root-hairs is to secure a plentiful supply of water for the

plant, that, when the root finds itself in water or wet soil, the root-hairs are not required, and may remain rudimentary, but, when the root is in ordinary soil, in which there is no excess of moisture, root-hairs develop and play an important part in the work of water-absorption.

A naked-eye examination of root-hairs suffices to show that they are extremely delicate structures. Microscopic examination confirms this. How delicate they are we demonstrate by pulling up and exposing several of the oat or maize seedlings (of Exp. 80) to the dry air of a room. We observe that the root-hairs soon shrivel and die. Herein lies the explanation of the reason why gardeners, when transplanting seedlings, and indeed any actively-growing plants, are careful to perform the operation quickly, and to leave a ball of earth attached to the root. For, otherwise, the root-hairs are injured and the plant suffers owing to its inability to form a new crop of root-hairs as fast as it needs supplies of water. We can understand, also, why cuttings which have, of course, no special absorbent organs, should be planted in moist soil and protected from loss of water.

82 The maize and oat seedlings of Exp. 80 serve to show that root-hairs are not only delicate, but ephemeral structures. By making sketches of sections of the roots daily, and indicating in the sketches the root-hair region, we discover that, although day after day the extent of the root-hair region remains fairly constant, this is due, not to the permanence of the individual root-hairs, but to the formation of new hairs, just behind the elongating region, to take the place of those which, the side distant from that region, are shrivelling and dying. Each day, the root-hairs in the older part of the root-hair region wither, and, each day, others are formed in the younger part of that region. If, therefore, we determine the time that the root takes to increase its length by an amount equal to that of the root-hair region, we have determined also the average length of life of a root-hair. By measurements of this kind we discover that the life of a root-hair is generally a matter of a day or two. So that the root of a plant bears root-hairs throughout its

it follows that, during that period, countless numbers of root-hairs arise, do their work, and die.

The evidence which we have brought together in support of the water-absorbing function of the root-hairs, though strong, is not direct. We have now to attempt to prove definitely that a root-hair cell possesses the power of absorbing water, and, if we succeed in this, we shall have to enquire how the absorption of water is effected. Our enquiry will necessitate the examination of root-hairs, and also other cellular elements of plants, and will give us information which will be of service to the understanding, not only of the mode of water-absorption by root-hairs, but also of many other phenomena exhibited by plants—as, for example, the method of growth of cells, the origin of the pressure exerted by roots in penetrating rigid soils, the means by which delicate seedlings hold themselves upright, and so on.

83 ¹ We begin this work by making a more thorough examination of preparations of root-hairs (see Exp 79¹). On examining them microscopically, we observe that the first sign of the root-hairs is a slight raising up or projection of the free, outer wall of a surface cell of the root-hair region. We recognise that, since the cellulose wall of a cell is dead, the rounded projection must be due to pressure from within. This pressure continuing, the cell-wall bulges outward till it forms a finger-like projection. By treating preparations with iodine solution, we find that, as in all other protoplasts, that which constitutes the root-hair cell consists of a mass of protoplasm. In the very young cell, the protoplasm forms a more or less dense mass, taking up practically the whole of the space occupied by the cell, but, in the older root-hair cell, the protoplasm, unrecognisable, perhaps, before the addition of the iodine solution, is seen, after its action, to form, and, indeed, to be confined to, a thin layer just beneath the cellulose wall. Within this protoplasmic layer is a clear space or vacuole occupied by a watery fluid, the cell-sap (see Fig 20). As the action of the iodine continues, this layer, or film of protoplasm stains brown and contracts away from the wall.

84 ¹ In the next place, in order to satisfy ourselves that

structures like root-hair cells are able to absorb water and pull off some of the violet hairs which occur on the filaments of the stamens of *Tradescantia virginiana*, or, if this material, we make sections parallel to the surface of the leaves of *Primula sinensis* (or cross sections of the stem of this plant). surface sections of the petals of clove flowers or, if nothing better is obtainable, we use thin sections of beetroot. Our object in choosing such material is to obtain cells which are similar in structure to root-hair cells, but in which the cell-sap is coloured, and hence is easily recognisable. Examine microscopically and draw the preparation. Observe the cell-wall and vacuole with its cell-sap and note that no protoplasm is to be seen. Now place under the cover-glass a 5-10 % solution of common salt or a 3-5 % solution of potassium nitrate (Appendix A). Note that, as the salt solution passes into the cell, the cell-sap of the vacuole contracts away from the wall and ceases to occupy a smaller space than before. Since the vacuole contains fluid, and since fluids are incompressible, it follows that the addition of the salt solution has caused so much of the liquid of the cell-sap to pass out from the vacuole. But the space which now exists between the cell-wall and the coloured blob of cell-sap is colourless, a fact which indicates that none of the dissolved pigment to which the cell-sap owes its colour has escaped from the vacuole. By repeating the experiment using the appearance of the cell in water with that of the cell acted on by salt solutions of increasing strength (Fig. 20), we are led to the conclusions (1) that the vacuole is surrounded by a membrane through which water may pass, but through which the dissolved pigment does not pass. (2) That, in the intact cell, the protoplasm is pressed close against the cellulose wall, and is therefore invisible. We treat the sections in salt solution and iodine solution, and observe that the layer in question gives the yellow-brown reaction characteristic of protoplasm. As the action of the iodine continues, observe that the protoplasmic layer, now killed, shrinks away from the wall, and, becoming disorganised, allows the contents of the vacuole to escape. We may note, also, the discoloured stained nucleus lying either in the protoplasmic membrane

or slung in 'bridles' of protoplasm stretching across the cell from one part of the membrane to another.

85 * Make another microscope preparation of the hairs of *Tradescantia*, etc., in dilute salt solution, and, whilst the coloured sap is still enclosed in the protoplasmic layer of the cell, add a drop or two of water to one edge of the cover glass, and, by means of blotting-paper placed at the other edge, withdraw the salt solution so that the cell is

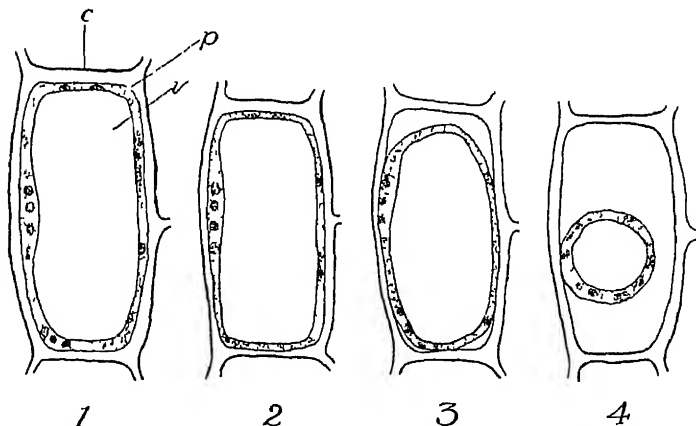


FIG 20 —YOUNG PLANT CELLS FROM THE TISSUE OF A FLOWER PEDUNCLE TO SHOW STAGES IN PLASMOLYSIS

1 In water 2 In 1 per cent solution of potassium nitrate 3 In 10 per cent solution of potassium nitrate 4 In 20 per cent solution of potassium nitrate c, cell wall, p, protoplasm, v, cell vacuole

(After De Vries.)

again in water. Note that the vacuole increases in size till once again its outline comes to lie close against the cell-wall. We thus learn that, when the salt solution is replaced by water, the latter substance passes through the membrane of protoplasm (or plasmatic membrane as it is called) to the vacuole. Thus, the volume of the cell-sap increases and, as a consequence, the membrane is stretched till it comes to lie close against the cellulose framework of the cell. We infer that, just as salt solution withdraws

water from the vacuole, so something in the vacuole may cause water to pass into it from the outside.

It follows from these facts that a vegetable cell is an apparatus of a remarkable nature, and that it consists of a protoplasmic layer across which water may pass, but across which other substances, *e.g.* dissolved pigment of the cell-sap contained in the vacuole, may be unable to pass. This layer is stretchable. Hence if water passes into the vacuole, the increased pressure thus set up stretches the protoplasmic membrane till it is pushed firmly against the cellulose wall. Since also the cellulose cell-wall is stretchable—though less so than the protoplasmic membrane—it more water passes into the vacuole, the cell-wall itself is stretched and the cell becomes larger (cf. Fig. 20, 1 and 2). If, whilst the cell-wall is thus stretched, the protoplasm secretes a new layer of cellulose on the inner surface of the old layer, and if this new layer is sufficiently rigid to maintain the whole cell-wall in its present position without the help of the pressure of the cell-sap in the vacuole, then this distension of the cell by the fluid pressure of the vacuole is made permanent, and so, if now the protoplasmic membrane is caused to shrink away from the wall, the latter, being more rigid by reason of the new layer of cellulose, does not collapse as much as before. There has been, therefore, a permanent increase in size of the cell: in other words, the cell has grown. In the case of the root-hair cell, the stretching of the wall takes effect only at the dome-shaped projection which, in consequence, comes to form the tubular outgrowth characteristic of the root-hair.

A living cell, the vacuole of which is so full of sap as to press the plasmatic membrane close against the cellulose wall of the cell and to stretch the latter to its full extent, is said to be turgid, and is described as being in a state of turgidity or turgor. If water is withdrawn from the vacuole, as, for example, by exposing the cell to the action of salt solution, the volume of the cell-sap decreases and the vacuole becomes smaller. The previously stretched protoplasmic membrane recoils, as a piece of stretched elastic recoils when released, and so contracts away from the cellulose wall. A cell in this condition is

said to be plasmolysed, and the process by which this condition is reached is called plasmolysis. The turgid cell may be compared with an inflated bicycle tyre, and the plasmolysed cell with a deflated tyre. As in the inflated tyre, the pressure of the air in the inner tube distends its wall, presses it against that of the outer tube, and stretches the wall of that tube so that the tyre becomes rigid, so, in the turgid cell, the pressure of the fluid in the vacuole stretches its wall (the plasmatic membrane), and, pressing it against the cell-wall, stretches that wall so that the cell becomes rigid and as a reduction of air-pressure in the inner tube—due to escape of air—results in the recoil of the inner tube, and a relaxation of the distended outer tube, with the result that the tyre becomes flabby, so a reduction of fluid-pressure in the vacuole—due to loss of water—results in the contraction of the previously stretched plasmatic membrane, and consequently in a relaxation of the cell-wall, with the result that the cell loses its rigidity. Consequently, if the elements of a cellular tissue are turgid, the tissue has a certain rigidity, if they are plasmolysed, the tissue becomes flabby.

Since, for reasons which will appear presently, these matters of turgidity and of plasmolysis are of fundamental importance, we will determine by experiments on various plants that the facts we have learned from the study of cells with coloured sap hold good for living vegetable cells in general. We note, for example, the supple sturdiness of delicate seedlings

86 Lay one or two cress seedlings on the table, and observe that, as they wither, they become flabby. As they dry, the cells lose water and the protoplasmic membranes cease to press on the cellulose walls. The walls are too delicate of themselves to form a sufficient scaffolding or skeleton for the plant, and the seedling is too young to have formed any supporting internal skeleton of woody tissue. Hence the whole plant becomes soft and droops. Note the wilting of leaves of plants in dry weather and infer the cause.

87 Take two bean seedlings, each with roots about $2\frac{1}{2}$ inches long, make on each an india-ink mark about 2 inches from the tip. record the exact distance between

mark and tip. Plunge one root into a 3% and the other into a 10% solution of potassium nitrate, leave for several hours. Note whether the roots have become flabby. Measure the distance between the tip and ink mark in each case. Refer the contraction to plasmolysis. Plunge roots into ordinary water. After some hours, remove and measure them. The root which was partially plasmolysed in the 3% solution has recovered its normal length. That subjected to 10% has not recovered. Plant both seedlings and determine that the root of the former grows, but that of the latter does not, having been killed by the stronger salt solution.

88 Make an India-ink mark beyond the growing tip of each of two bean roots growing in moist air. Determine the rates of elongation of the roots. Plunge one root in a 4% solution of potassium nitrate, the other into water. Determine that the root in water continues, and that in salt solution ceases, to grow in length. Transpose roots and observe that the partial plasmolysis due to salt solution is sufficient to check growth; thus prove that turgor is a condition for growth.

89 Cut two square chunks of beetroot of about equal sizes. Wipe their surfaces. Plunge one into hot water, the other into cold water. Note that the beet escapes from the former and not from the latter. The escape being due to the destruction of the protoplasmic membrane. Hence determine the approximate temperature at which the cells of the beet are killed.

90 Cut a cucumber in slices, throw some of the slices on a plate, and cover them with salt. After some hours compare their flabbiness with the natural firmness of untreated slices. Consider the influence of thus plasmolysing a cucumber on its digestibility.

91 Select several flower-stalks of dandelions bearing open buds, remove the flower-head and split a stalk lengthwise into four sections. Note that the pieces curl so that the inner tissue is outside. Cut the curls into rings, then several into water, and note that the rings coil up in the same sense as before. The coiling in water is due to the fact that the cells of the originally inner surface absorb water, and so increase in volume, whereas those of

originally outer surface of the stalk absorb no water, or little, do not stretch, and hence are obliged, by the stretching of the inner rows of cells, to bend. Make a model of wire and india-rubber tubing to illustrate this bending.

Inasmuch as a tissue, when turgid, is longer, owing to the stretching of the cell-walls, than when it is plasmolysed, it follows that we can measure the amount of pressure necessary to produce this difference of length by first plasmolysing a piece of a plant consisting mainly of cellular tissue and then determining how big a pull is necessary to stretch it to its original length.

92 A cowslip flower-stalk serves well for this purpose. Two india-ink marks are made at about two inches distance from one another, the distance between the marks is measured accurately, and the stalk is put into a strong solution of potassium nitrate (15 %). When the stem has become flabby, the amount of contraction is determined by remeasuring the distance between the marks. The stalk is then laid horizontally on a plate of cork or wood, one end, just beyond an ink mark, is fixed securely to the plate by means of pins or thin nails, and to the other end, just beyond the other mark, a stout thread is tied, the thread is passed over a simple pulley fixed to the edge of the table and attached to a scale pan taken from a chemical balance. By placing a millimetre scale beside the stalk, and by adding weights to the scale pan, determine the weight required to stretch the stalk to its original length. To express the result more definitely, measure, *e.g.* by means of a microscope provided with a micrometer (Appendix A), the area of a cross-section of the stalk at the beginning of the experiment, and, neglecting the fact that the stalk is made up of a vast number of cells, consider it as a single cell and argue thus —the pull on the end walls necessary to stretch the plasmolysed stalk to its original length is so much; but this original length of the turgid stalk was maintained by the pressure of the cell-sap in the vacuole; therefore, the latter pressure, causing turgor, is equal to the mechanical pull due to the weights on the scale pan; since we know the area of the cross-section we can express this pressure in pounds per square inch; and since we know that the atmosphere exerts a pressure of about 22 lbs

to the square inch, we can express the turgor-pressure in terms of atmospheric pressure, e.g. as equal to 1, 2, 4, 5, or more atmospheres. By means of measurement made in this and other ways, it has been found that the turgor-pressure in plant-tissues may be very great indeed, reaching as much as 10, or, in some cases, even 20 atmospheres. It is by reason of these great pressures that relatively delicate plant-structures may, when they encounter obstacles to their growth, perform remarkable feats of strength: thus they may cleave rocks or, like the thistles of the American prairies, raise up and bend railway lines.

Our investigation of the mode of absorption of water by the root-hair cell has carried us a long way from our immediate subject, nevertheless, it has not been irrelevant. For it is evident that, just as the partially plasmolysed cell of Exp. 85¹ absorbs water and recovers its turgidity, so may a root-hair cell absorb water and become distended. If a cell neighbouring on a root-hair cell has been losing water, it may withdraw water from the vacuole of the root-hair cell just as the salt solution withdraws water from the turgid cell in Exp. 84¹. Then the root-hair cell, having lost water to its neighbours, may take up from the soil more water to make good that loss. In this way is obtained the large quantity of water absorbed by the root and in this way also the water, thus obtained, passes from the root-hair cells into the wood-vessels of the root (Bibliography, 5, 6).

when the stopper is taken out of the bottle in which it is contained. The rapid dissemination of the odour is doubtless due to air currents which carry the odouriferous substance; but, even if the air of the room was perfectly still, the vapour given off by the scent would be distributed gradually throughout the space.

In order to account for the phenomenon of gas diffusion, and also for other phenomena exhibited by gases—a theory—the kinetic theory of gases—has been put forward, which suggests that, though, in any gas, there are no visible mass movements, nevertheless the invisible molecules are in a state of constant movement. Apply this theory to liquids, we may suppose that, in a solution of salt, the molecules of salt are moving through the water in all directions like those of a gas liberated in the water, though, owing to some form of obstruction on the part of the water molecules, not so freely. From this hypothesis it should follow that if a solution of salt is brought in contact with water without shaking, the salt molecules travel gradually throughout the water till, ultimately, they become uniformly distributed in it. That this actually takes place, we demonstrate in the following way:—

93 A long cylindrical jar is filled with water and set on a steady table. By means of a funnel (e.g. the funnel, see Appendix A) a strong solution of a coloured salt, e.g. bichromate of potassium is poured into the vessel so as to form a distinct layer at the bottom. A white paper millimetric scale is fixed behind the vessel and a daily record taken of the height reached by the potassium bichromate.

This way of measuring the rate of diffusion is open to the objections that changes of temperature set up convection currents in the liquid, and that, unless the table on which the vessel stands is perfectly steady, vibrations caused by people walking in the room, etc., increase the rate at which the liquids mix. Both sources of error lead to a fictitiously high result.

The following method is simpler to carry out and is open to these objections:—

94 To leaf gelatine, obtainable in commerce, add enough water to form a liquid which, when it cools, sets to a firm jelly. Before the liquid gelatine cools, add

small quantity of thymol or other antiseptic (Appendix A), and also a few drops of phenolphthalein (Appendix A), a substance which undergoes a marked colour change—from colourless to rose—when acted on by an alkali. Pour the warm, liquid gelatine carefully into a tall jar and, when it has set, invert the jar over a dish containing a fairly strong solution of potash (KOH) or soda

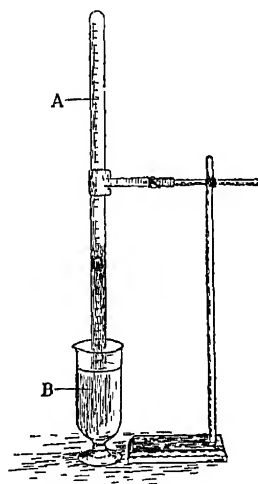


FIG. 21—APPARATUS TO ILLUSTRATE DIFFUSION OF LIQUIDS

A graduated glass tube (A) containing an indicator (phenolphthalein) is inverted in a vessel (B) containing a solution of potash.

(NaOH), which give a strong alkaline reaction. Determine the rate of diffusion of the alkali by tracing the ascent of the rose colour. Record the results. The fact that the experiment lasts for weeks should not lead to neglect in the matter of recording, for it is important for us to learn that this liquid diffusion is a *slow* process. The experiment demonstrates also that the gelatine does not prevent the diffusion of water and potash.

We know from previous experiments (*e.g.* Exp. 26) that a similar intermixing of water and sugar takes

place when these substances are separated from each other by a parchment membrane. Nor is this surprising if we suppose that diffusion of gases or liquids is due to molecular- and not to mass-movement. Although such a parchment membrane, when used to contain water, does not leak, its substance must be pictured riddled with a system of ultramicroscopic spaces, too small to allow the smallest drop of water to pass, but am large enough to allow of free movement of molecules H_2O . Inasmuch as the molecules of fluids are, according to our assumption, like those of gases, in constant movement, some water-molecules pass into the interstices of the membrane, and, arriving at the other side, come in chemical contact with the sugar-molecules. The fact that sugar is soluble in water means that sugar- and water molecules exercise some attraction on one another. We may conceive of the sugar-molecule linking with the water molecule and joining it in the molecular dance; though being heavy, the sugar-molecule will cause a slowing of the rate of movement. The molecular dance is chaotic. Sugar and water partners career in all directions, so back through the interstices of the parchment membrane some on through the sugar solution. Since the molecular movements are unending, the result is that, in course of time, there are as many sugar-molecules per unit volume on one side of the membrane as on the other. Looking at the final result, we say that sugar and water have passed by osmosis each from one side of the membrane to the other. Even after equilibrium has been reached, that is, when the liquid on one side of the membrane contains per unit volume as much sugar as that on the other side, the molecular movements do not cease. But, in these circumstances, the number of sugar-molecules conveyed in a given time across the membrane in one direction is equal to the number which pass in the opposite direction. Just so may the population of a town remain remarkably constant though the townsmen come and go.

In order to study the phenomenon of osmosis more closely, with the object of learning more about the mechanism of absorption by the plant, we make an osmotic apparatus thus —

95 Cut off a length of about eight inches of parchment tube (Appendix B). If dry, soak the parchment tube in water when thoroughly wet, fix in one end a rubber cork without holes and make the joint good by winding tightly round the parchment tube many turns of stout silk thread or florist's buttonhole wire. Stand the tube upright, open end uppermost, in a pail or other suitable large vessel. Make two holes in opposite sides of the parchment on a level with the rim of the pail. Pass a glass rod through the holes, and let the rod rest on the rim (Fig. 11). Fill the outer vessel with water up to about two inches below the level of the glass rod, then fill the parchment tube nearly up to the level of the holes through which the glass rod passes. Leave the apparatus for a short time and observe whether any water has escaped from the tube. If so, there is a hole in it, and the tube must be replaced by another. If it proves to be sound, empty the tube and cut away the part pierced by the holes through which the glass rod passed. Choose a two-holed rubber cork which just fits the tube. Insert through one hole the tubular limb of a separating funnel, seeing that it makes a good joint. Through the second hole in the cork, pass a glass tube about three feet long so that its lower end is flush with the lower surface of the cork. Stand the empty parchment tube in the vessel of water, insert the cork, and make good the junction between cork and tube by many turns of silk thread, supplemented, if necessary, by wax (Appendix A). Fix the glass tubes, or the longer one only, in a clamp, so that the lower cork rests on the bottom of the outer vessel and the parchment tube is steadied (Fig. 22). Prepare a strong solution of sugar, colour the sugar with a dye such as methylene blue. Pour the coloured sugar solution into the separating funnel. Open the stop-cock of the latter and run in the sugar solution till, the parchment tube being full, the liquid rises in the long glass tube to about half an inch above the level of the cork. Close the stop-cock of the separating funnel. Pour water into the outer vessel to about the level of the top of the parchment tube. If any sugar solution has been spilled into the outer vessel, transfer the osmotic apparatus bodily to another, similar vessel of clean water.

The apparatus is laborious to make, but is so instructive in action that no trouble should be spared to get it set. An improvement which enables us to use it again and again is illustrated in Fig. 22. The outer vessel, which serves

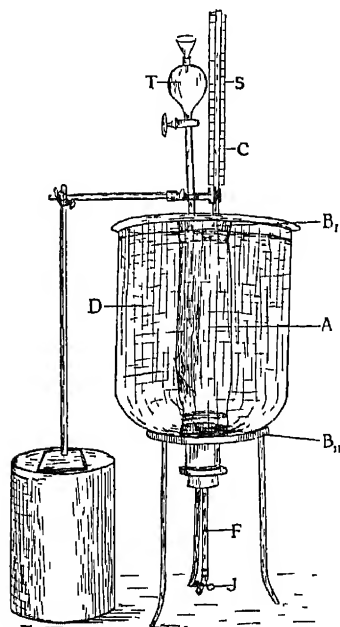


FIG. 22 — APPARATUS TO DEMONSTRATE OSMOTIC PRESSURE.

A, parchment tube, B₁, B₁₁, rubber corks, I, thistle funnel, C, glass tube, D, vessel containing water, F, exit tube, J, clamp, S, scale.

to contain the parchment tube, consists of an inverted bell-jar with an open neck (Fig. 22 D), and is supported on a strong iron tripod-stand. The lower end of the parchment tube is closed by a rubber cork (B₁₁) with a hole, into which is fitted a short glass tube, the upper of which is flush with the upper surface of the cork. A tube is passed also through the hole in a rubber cork which

closes the neck of the bell-jar. To the free end of the short glass tube (F), a short piece of rubber tubing is attached and closed by means of a clamp (J). When the upper cork (B₁) with its funnel and long glass tube has been inserted in the upper end of the parchment tube, the bell-jar is filled with water and the apparatus is ready for use. The coloured sugar solution, to which a trace of an antiseptic may be added, is poured through the funnel and the air contained in the short tube F is expelled by opening the clamp J. After the apparatus has been used in the way about to be described, the liquid in the parchment tube may be withdrawn through the tube F and the parchment tube refilled with sugar or other solution, the osmotic properties of which it is required to test.

An osmotic apparatus of this kind, once fitted up, will last for months if the parchment tube is not allowed to become dry, and may be used for various interesting experiments in osmosis.

If all the joints are good and the parchment tube intact, the fluid begins to rise almost at once in the long glass tube. Fix a paper scale behind the long tube and take time-records of the height to which the liquid rises. After some time, it becomes necessary to add another length of glass tubing, which is connected with the long tube by a short piece of rubber tubing and supported, e.g. against the wall of the room. The liquid continues to rise to a height of many feet. Then, after some time, it begins to fall and finally descends to the level of the liquid in the outer vessel. Before this takes place, note that the dye has escaped through the parchment-wall into the water outside, and prove by the sugar test (Exp. 21) that sugar also has passed into the outer vessel.

The rise of the water in the long glass tube means that the volume of liquid in the osmotic apparatus has increased, and hence that water has passed through the parchment-wall. Therefore, in the course of osmosis there is, as indeed we should expect from our previous studies, a movement in both directions: sugar and soluble pigment pass from parchment tube to the outer vessel, water passes into the long tube, the amount of water which passes in is greater

than the amount of sugar which passes out. We may say that the sugar solution exercises an osmotic attraction on water, and that it sets up an osmotic pressure sufficient to hold up the high column of water in the long tube.

But since, from the moment the apparatus is set to work, sugar begins to diffuse out into the surrounding liquid and since, in that situation, it sets up a counter-attraction tending to withdraw water from the parchment tube, the apparatus is useless as an osmometer,—that is, as a measure of the actual osmotic pressure exerted by the original sugar solution. It serves only to indicate that such pressure exists and can be made to do work, e.g. lifting water. That the sugar which escapes into the outer vessel does set up a pressure tending to neutralise that in the parchment tube we demonstrate thus.—When the liquid in the long tube has reached a certain height, has become fairly stationary, siphon off that in the outer vessel and replace it by water. Note that the liquid in the long tube begins again to ascend. The level previously reached represented the resultant of the osmotic pressure exerted by the sugar solutions inside and outside the parchment tube. By removing the sugar from the outer vessel, the pressure exerted by that in the tube is no longer in part counteracted, and so is able to hold up a longer column of water.

96 By using solutions, e.g. of cane-sugar and grape-sugar successively, we may demonstrate by means of our osmotic apparatus that the rates at which substances in solution pass through a parchment membrane depend, among other things, on their molecular weight. The molecular weight of cane-sugar ($C_{12}H_{22}O_{11}$) is 342, that of glucose ($C_6H_{12}O_6$) is 180, therefore, if we dissolve the molecular weights of cane-sugar and of glucose in equal volumes of water, say 1000 c.c., and use the one and then the other solution in the parchment tube we find that the height of the water column supported by the cane-sugar solution is greater than that supported by the grape-sugar. For the heavier cane-sugar molecules diffuse more slowly than the lighter molecules of glucose. Hence before the latter has had time to get up its full osmotic

pressure, much of it has diffused into the outer vessel. Similarly if we fill the osmotic apparatus with a solution of potassium nitrate of corresponding strength, *i.e.* one containing the molecular weight of KNO_3 in grams, viz. 101 grams, per 1000 c.c. of water, we find that, owing to the high rate of diffusion of potassium nitrate, the osmotic pressure which the solution sets up is very small indeed and by no means a true measure of that which it can exert.

We have, however, in plant-cells and tissues an apparatus ready-made for comparing the osmotic pressures exerted by different substances. For a vegetable cell is an osmotic apparatus strikingly similar to that used in Exp. 95. The osmotic substances dissolved in the cell-sap correspond to the sugar or other solution in that apparatus, and the protoplasmic membrane (plasmatic membrane) of the cell corresponds to the parchment membrane. If a completely turgid vegetable cell is placed in water, it undergoes no change in volume since the cell-wall is already stretched to its limit by the pressure of the cell-sap; if, however, the cell is not completely turgid, water is absorbed, and the volume of the cell is increased—the osmotic pressure being used to do the work of stretching the plasmatic membrane and cell-wall. If such a cell is placed in a strong salt solution, the osmotic pressure of the latter causes water to pass out from the vacuole, across the protoplast and cell-wall. The amount of fluid in the vacuole being thus reduced, the pressure on protoplast and cell-wall falls off and the volume of the cell decreases. As water is withdrawn, the solution of osmotic substances in the cell-sap becomes more concentrated, and consequently the osmotic pressure of the cell-sap increases. If the osmotic pressure of the salt solution used is greater than that of the concentrated sap, more water is withdrawn from the vacuole, plasmolysis sets in, and, yet more water being withdrawn, the protoplast, shrunk away from the wall, may collapse and become disorganised. If, then, we place vegetable cells or tissues in salt solutions of different strengths, we find that, whereas, in strong solutions, plasmolysis is complete, and in very weak solutions it does not occur at all, there is one strength of solution which just suffices to produce the early stage of plasmolysis and no more.

NO_3 , and if each of these exercises osmotic pressure, we can form an idea of the reason why the osmotic pressure exerted by a solution of KNO_3 of known strength is greater than we should have expected on the basis of the "law," which we have established for cane-sugar and glucose, that equi-molecular solutions of osmotic substances exert equal osmotic pressures. It was by means of experiments such as these that the discovery was made that the osmotic pressure of a substance is a property of its molecules in the same way that weight, etc., are properties of the molecules. These discoveries have led to others of equal interest, and those in turn have served as the basis for various theories to account for the known facts. We cannot, however, pursue this part of our subject further, but refer students who wish to learn more about the physical aspect of osmotic pressure to the larger textbooks (Bibliography, 11).

The comparison of a plant-cell with the osmotic apparatus of parchment tube (see p. 113) fails in one very important particular. For, whereas such substances as soluble pigments pass readily across the parchment-wall (see Exp. 84*), the soluble pigment in the cell-sap remains, in the partially plasmolysed cell, enclosed within the plasmatic membrane. There must, therefore, be an important difference between parchment and protoplasmic membranes. The parchment membrane is indiscriminately permeable to diffusible substances, such as water, sugar, potassium nitrate, soluble pigments like methylene blue, etc., but the protoplasmic membrane is not. The former kind of membrane is called a permeable membrane; the latter kind, which is permeable to water but not to all diffusible substances, is called a semi-permeable membrane. The importance to the plant of this property of semi-permeability is very great indeed. For, were the osmotic membrane of the plant-cell permeable, there would be a constant osmotic leakage of osmotic substances from the cells of the plant into the soil just as there is a constant osmotic leakage of sugar from the osmotic apparatus of Exp. 95.

98 Instead of using cells with coloured cell-sap, the curved pieces of split dandelion stalk (Exp. 91) may be

employed in the above experiment. To use them, proceed as follows —

Split a dandelion stalk longitudinally into four pieces, cut the pieces into lengths of about an inch each, and having prepared the series of solutions the osmotic pressures of which are to be determined, trace on paper, by means of a brush and india-ink, the amount of curvature of each piece of stalk. Throw a piece into each of the solutions, after some minutes, take the pieces out, determine which have curved more, and which have become straighter.

The solutions which have caused a straightening have done so by withdrawing water from the cells of the dandelion. Their osmotic pressures are, therefore, greater than the osmotic pressure of the cell-sap of the cells of the stalk.

Those solutions in which the pieces have curved more have yielded water to the cells, that is, the osmotic pressure of these solutions is lower than that of the cell-sap of the dandelion cells. Hence, by finding a solution in which pieces of dandelion stalk retain their curvature unchanged, we find what strength of solution is in osmotic balance with the cell-sap.

99 By using pieces of dandelion stalk in the same way we may compare known strengths of different osmotic substances with one another, with respect to their osmotic pressure.

It is important to demonstrate that the property of semi-permeability referred to on p. 115 is not confined to living membranes.

100 Thus we may make a semi-permeable membrane called, because of its mode of preparation, a precipitation membrane, by causing solutions of copper sulphate and potassium ferrocyanide to interact. The precipitate of copper ferrocyanide which forms as the result of this interaction, is not granular, but skin-like. To prepare it, pour a 3% solution of potassium ferrocyanide into a wide mouthed bottle fitted loosely with a cork. Fix a glass tube, drawn out at one end to a fine point, in a hole in the cork so that its fine end dips below the ferrocyanide solution. Having withdrawn and cleaned the tube, draw a drop of strong copper sulphate solution into it, close the tub

by the finger, and fit the cork attached to it in the neck of the bottle so that the fine end of the tube is just below the surface of the potassium ferrocyanide solution. When the finger is removed from the end of the tube, the copper sulphate reacts with the ferrocyanide to produce a membrane, which closes the fine end of the tube. Inasmuch as the copper sulphate on one side is a strong solution and has an osmotic pressure higher than that of the weak potassium ferrocyanide solution on the other side of the precipitation membrane, water passes from the latter solution to the former. Hence the volume of the copper sulphate solution increases, its increased weight stretches the precipitation membrane and the "artificial cell" grows. As the pressure of the copper sulphate solution increases further, the membrane is ruptured, but it forms again owing to a new skin of copper ferrocyanide being produced as soon as the copper sulphate and potassium ferrocyanide come in contact. Again the "cell" grows, and again it is ruptured to be once more repaired. This membrane, permeable to water but not to such substances as copper sulphate or potassium ferrocyanide, is a semi-permeable membrane.

Another method of making a semi-permeable precipitation membrane is as follows:—

101. To ordinary gum, add a small quantity of gelatine, some sugar and a little colouring matter, *e.g.* aniline blue in solution. Take up a little of the gum-mixture on the rounded end of a piece of thick glass rod and expose it to the air till it is dry. Put the rod so that its gummed end is in a 2% solution of tannin. A membranous precipitate of tannate of gelatine is formed on the surface of the dried gum. The sugar contained in the gum exerts its osmotic pressure, water passes across the membrane, the latter is stretched and behaves like the membrane in the former experiment. The dye, however, does not pass across. Thus the precipitation membrane behaves like the plasmatic membrane of a vegetable cell (cf. Exp. 84⁺), permitting water to pass across, but preventing the osmosis of the dissolved pigment.

102 Again, we may illustrate the semi-permeability of precipitation membranes by the use of our parchment tubes.

Prepare two tubes, A and B, supported by glass rods as in Exp. 17. Into each pour a 1 % solution of calcium nitrate in A and B in a 1 % solution of disodium phosphate. Add a little methylene blue to the liquid in each parchment tube. After a day, note that the methylene blue has appeared in the water of the outer vessel of A, but not in that of B. In B, the calcium nitrate and disodium phosphate interact to form a precipitation membrane of calcium phosphate on the wall of the parchment tube, this membrane, though permeable to water, is impermeable to a solution of methylene blue.

On the lines of such an apparatus as that of Exp. 17, a perfect, working, osmotic model of a vegetable cell may be constructed.

Thus we reach the end of our enquiry. We discover that, by means of its root-hair cells, the root absorbs water together with any soluble osmotic substances which the plasmatic membrane of the root-hair cell is permeable to. What these latter substances are, we determine by other methods and in another chapter.

CHAPTER VIII

THE substances taken up by the roots of plants The composition of plant
ish Water- and sand cultures The soil in relation to plant life
The origin of soils their physical, chemical, and biological
properties

THE fact, established in the last chapter, that root-hairs are capable of absorbing water and any other soluble substances to which their plasmatic membranes are permeable, leads directly to the enquiry —What are the substances which the roots of plants absorb, and to what use does the plant put the substances which it obtains from the soil?

That water is absorbed by the root is evident without further experiment, for, on the one hand, we have found (Exp. 5) that plants contain large quantities of water, and, on the other, we know that unless water is supplied to their roots, plants wither and die.

Presently we shall have to enquire more closely into the relation of the plant to water, but, in the meantime, we will devote ourselves to discovering what substances, other than water, are absorbed by the root.

The same experiment which taught us that the tissues of plants contain considerable quantities of water, proved also that the dry matter, left after the water is driven off, consists largely of carbon compounds, and that, after these compounds are burned away, a relatively small amount of mineral matter remains behind as ash.

Now, carbon compounds of the kind contained in the plant do not exist in the soil, but inasmuch as such mineral substances as those found in the ash are always present in the earth, we may be fairly confident that the mineral substances of the ash of plants are obtained in one form or another from the soil.

To determine accurately and completely all the con-

stituent elements contained in plant-ash would require a somewhat elaborate chemical analysis. Though such an analysis is beyond us, we can demonstrate readily the presence of some of the more important elements by the following methods:—

103 Fill a large bottle with fine ashes from a bonfire, or from a bundle of dried hay burned for the purpose.

Place some of the ash on a piece of platinum foil (Appendix A), and heat it in a Bunsen flame. Roll up the platinum foil with the remains of the fused ash, place it in a test tube, add a little distilled water, boil. Test the solution so obtained with red litmus paper: a blue colour shows that an alkali is present. (Use the solution for Exp. 105.)

104 Moisten a platinum wire (Appendix A) with hydrochloric acid, dip it in the ash and hold it in the flame. Observe the yellow colour of the flame, due to the presence of *sodium*. Repeat, observing the colour of the flame through a piece of cobalt-blue glass: a violet colour indicates the presence of *potassium*.

105. To the solution obtained in Exp. 103, add an equal volume of dilute nitric acid. Pour the liquid into a clean test tube, add an excess of a solution of ammonium molybdate (Appendix A), boil: a yellow precipitate indicates that *phosphates* are present in the ash.

By appropriate tests, the presence in the ash of other mineral substances—compounds of *calcium*, *magnesium*, etc.—may be demonstrated.

It is to be noted, however, that these tests give no indication as to the form in which the elements, potassium, magnesium, calcium, etc., exist in the plant, for the compounds present in the intact plant are decomposed in the process of burning.

Since, however, as our tests indicate, potassium, sodium, calcium, phosphorus, etc., are present in combined form in the ash, these elements must have been present in some form or other in the plants which yielded the ash. Whence it follows that they were obtained from the soil.

The question therefore arises, are the mineral substances which plants contain mere accidental accumulations or are they of significance in plant nutrition?

that silicon, chlorine and sodium, though generally present in plant-ash, are not essential for the perfect development of flowering plants in general.

Since water- and sand-cultures are among the most instructive experiments in plant physiology, we proceed to carry them out with a view to ascertaining what symptoms are manifested by plants deprived of one or other of the essential, mineral substances. The salts and also the water used for the culture solutions must be pure. Full instructions as to the mode of preparation of solutions, see Appendix A.

It will be of interest to repeat, as the first of our sand-culture experiments, the original sand-culture made by Boussingault, who introduced the method about 1868.

106 Procure good silver sand, clean it (Appendix A), place pieces of clean cloth (broken flower pots) at the bottom of each of six, small, new pots and fill the pots with the sand. Plant one pea seed in each pot (the seeds, before planting, may be plunged whilst dry into hot water for a few minutes, soaked in tepid water for twelve hours and then sown). Water thoroughly and cover the top of each with a piece of glass. When the seedlings appear, water three of the pots with complete culture solution (Normal solution, Appendix A), and three with a culture solution complete, except for nitrates (Normal, minus nitrates). When planting the seeds in the pots of sand, plant others in pots of ordinary soil to serve for comparison. Record the rate of growth of the seedlings of the three series when the difference between them with and without nitrates has become marked, draw a photograph of the plants (Frontispiece). From the stunted appearance presented by the plants grown in sand lacking nitrates, and from the healthy appearance presented by those in sand containing nitrates, we conclude that a plant cannot live without nitrogen, that the supply of nitric acid which the seedling peas contain in the form of protoplasm in the aleurone grains of the cotyledons (cf. Chapter I) is soon exhausted, and that the plant then turns to the soil for further supplies of nitrogen, which supply the soil may furnish in the form of nitrates. Be thus demonstrating that plants derive nitrogen from

soil, we get a striking example of the complexity of the chemical process carried on by plants. For, from our study of the nutrition of seedlings, we have learned that plants build up their protoplasm from complex, organic nitrogen-compounds (e.g. proteins), and now we are driven to the conclusion that the raw materials, which provide the nitrogen contained in the proteins of plants, are the relatively simple, inorganic nitrogen salts which are absorbed from the soil. Though physiologists have spent much time in investigating the process of nitrogen-assimilation, i.e. the process whereby the nitrogen of the nitrates of the soil is combined with other elements to form amino-compounds and proteins, our knowledge of the subject remains fragmentary (Bibliography, 10, 11).

Beside its historical interest, this sand-culture experiment of Boussingault has yet another interest, and for that reason we shall return to it presently (p. 141).

107 Repeat Exp. 106, using plants with smaller seeds in order to avoid the complications introduced by reserve-materials. Among the various plants which may be used successfully for sand-cultures, poppies (*Papaver* sp.) or the Californian poppy (*Eschscholtzia californica*) are, by reason of the readiness with which they grow in sand, among the best.

108 By means of sand-cultures determine the effect of withholding certain of the other elements of our list, viz. — calcium, magnesium, and phosphorus (see Appendix A). Though the results are less striking than those obtained by withholding nitrogen, yet, in each of these cases, the plant fails to develop properly.

Mount the drawings or photographic records of the sand-culture experiments on cards, add details and preserve them in the museum. Record peculiarities as to leaf-development, colour of leaves, growth of stem, flower- and fruit-formation in each case.

Unless the sand used in the above cultures is pure, and unless special precautions are taken, an attempt to demonstrate the effect of withholding iron will fail, and this for two reasons. First, the amount of iron required by plants is extremely small, and second, traces of iron occur in most samples of sand. Hence to demonstrate the effect

produced in the plant by lack of iron we resort to the method of water-culture (Appendix A)

109 Procure several good-sized glass bottles—jam bottles answer the purpose well. Clean and sterilize them and immerse the bottles almost to their rims in hot water of a temperature of about 50°C . Melt a little soft paraffin (of melting point 40°C , see Appendix A) in a porcelain dish, and pour it into one of the bottles. Rotate the bottle so that the paraffin, as it cools, forms a thin layer over the whole of its inner surface. Do the same with the other bottles in a similar way. Having germinated maize so that the roots of the seedlings are in water, remove the endosperm from each of several seedlings, replace the seedlings and leave them for a day with their roots in water. Fill two paraffined bottles, one with a "normal" culture solution, the other with "normal, minus iron". Make several cuts in a plate of cork, passing from the edge to near the centre. Slip the lower parts of the shoots of healthy seedlings each into one of the groove-like cuts, wedge each seedling firmly by means of cotton wool. Place the cork over the bottle, the fluid in which is at such a level that it covers the roots but does not reach the shoots. By adopting the precautions mentioned in the appendix, the plants may be kept in a healthy state. Note that the leaves which form in the "normal, minus iron" solution are not green but yellowish or white. In the absence of iron, chlorophyll, the green colouring matter of plants, fails to develop. Paint a part of such a chlorotic leaf with a dilute solution of iron sulphate, and observe that it becomes green in the course of a day or so.

It is a curious fact that though iron is essential for the development of chlorophyll, that substance does not contain iron.

In Nature, certain plants on certain soils become chlorotic and fail to develop properly. It has been shown that this state is due—at all events in many cases—to the inability of the plants to obtain iron from the soil, and that it may be remedied by injecting solutions of iron salts into the stems. It must be borne in mind, however, that not all white leaves are due to the lack of iron. Many plants, e.g. maple, ivy, holly, pelargonium, etc., have the habit

forming races with variegated leaves, and, in these cases, we have no reason to think that the absence of colour is due to lack of iron: again, most plants grown in darkness (cf p 44) produce colourless leaves. Hence we may suppose that the presence of iron salts, like light, is a necessary condition for the development of chlorophyll.

By referring to the records made of the growth of the sand-cultures, it will be found that, when calcium is absent, the leaves tend to become yellowish, and that, in the absence of magnesium, they often take on a dirty-brown colour. Inasmuch as it has been shown recently that chlorophyll contains magnesium, it should be possible to demonstrate that, in the complete absence of that element, plants do not produce the normal green chloroplasts (chlorophyll grains).

The results of sand- and water-culture experiments prove that the following elements obtained in combination from the soil in the form of salts are essential for the growth and development of flowering plants—nitrogen, phosphorus, sulphur, calcium, magnesium, potassium and iron, together with hydrogen and oxygen obtained in the form of water (H_2O). As the result of chemical processes, of which at present the green plant guards the secret, certain of these elements—nitrogen, sulphur, phosphorus, and potash, as well as oxygen and hydrogen—become constituents of plastic food-materials such as the proteins, and, finally, are incorporated with the protoplasm. Others, such as magnesium, become part of the indispensable chlorophyll grains. Hence we must regard the essential mineral salts as contributing with water to the raw materials of the food.

The precise function of some of these mineral substances is not clear. Some of them, as, for example, iron, seem to condition the formation rather than to enter into the construction of the plant-machinery, others play yet other parts—thus calcium has a rôle in connection with the transport of plastic carbohydrates. For, in its absence, the accumulation of starch in abnormally large quantities in the leaves, indicates that something has gone wrong with the transport service (cf also p 185). Calcium occurs also in a combined form in certain layers of the cell-walls.

Our studies of the osmotic properties of cells have taught us that only such mineral substances as are soluble in water can pass into the root-hair cell. That the root has, however, the power of dissolving certain mineral substances, we demonstrate thus —

110 Pack a large flower-pot half full with coco-fibre, place a slab of polished marble so that one end rests on the coco-fibre and the other on the side of the pot, such a position that downward-growing roots come into contact with the surface of the marble. Press a layer of coco-fibre firmly into the pot, plant soaked bean seeds, etc. them with coco-fibre, and water thoroughly. After several weeks, turn the marble out of the pot and observe that the root-systems of the seedlings have etched upon the surface copies of their outlines. Since marble (CaCO_3) is soluble in water containing carbon dioxide (CO_2) in solution would suffice for the etching process for the root to produce—as we know it does—carbon dioxide. Whether root-hairs are able also to excrete other more powerful solvent substances is a matter of dispute.

That the mineral substances absorbed by the root-hairs are taken up in extremely dilute solution follows from the diluteness of the solutions which suffice for water-sand-cultures. Indeed, our study of osmosis would lead us to predict that, where the rooting medium contains osmotic substances in any considerable degree of concentration, water absorption is hindered (cf. also p. 128).

We demonstrate the fact that the addition of strong solutions of salts may act injuriously on plants, e.g. plasmolysing the root-hairs, thus —

111 Raise turnips or oats from seed planted in six inches of ordinary soil. When the seedlings are up, water each of the pots daily with one of the following —water, 1%, 5%, 10% and 20% potassium nitrate solutions, and determine that the strong solutions exercise a disastrous effect.

Gardeners have long since recognised the fact that a plant may be deficient in some one or other of the essential mineral substances required by plants, and that, although it may not have lacked such a substance originally, it may come to be poor in an essential mineral, owing to the absorption of this substance by previous crops. If

know, also, that animal manures contain relatively small quantities of such substances as phosphates. Hence, in order to restore the fertility of the soil, they apply to it both dung and artificial manures or fertilisers. When a fertiliser is readily soluble in water, e.g. nitrate of soda, sulphate of ammonia, etc., care is taken to add it in small quantities, for, otherwise, it is apt to "burn" the plants.

112 This burning effect may be demonstrated by adding powdered sulphate of ammonia to a patch of a lawn infested with daisies. After a few days, the daisies present on the lawn are brown and shivelled, and the grass itself may be also damaged temporarily. It is owing to the special susceptibility of daisies to damage by sulphate of ammonia that this substance is employed, mixed with sand, as a weed-killer on putting-greens and lawns.

Another matter of interest which our various observations now allow us to explain, at all events in part, is that known as *selective absorption*. Different crops make different demands on the mineral constituents of a soil. Some, like potatoes and root-crops generally, take up more potash salts, others, like leguminous crops, take up more phosphates, and so on. Now, if a substance, say a salt of phosphorus, is taken up by the root and distributed throughout the plant, a moment comes when, if the phosphates in the plant have not in the meantime undergone some change, there is as much phosphate in solution in the cell-sap as in the water of the soil. Hence, automatically, the accumulation of phosphates ceases. If, then, we consider two plants, one which makes use of phosphates largely, the other sparingly, then, though grown in precisely similar soils, the former takes up during its life large quantities of phosphates, the other takes up but little. It will be evident that the behaviour of different crops with respect to amount of absorption from the soil of the essential mineral substances must be a factor in determining the rotation of crops.

It remains to enquire how the dilute solutions of the various mineral substances are absorbed by the root. We have described the plasmatic membrane of the root-hair as a semi-permeable membrane, that is, one which, whilst

permeable to water and certain diffusible substances, impermeable to certain other diffusible substances. We have, therefore, to assume that the root-hair cell, while impermeable to the osmotic substances which it contains dissolved in the cell-sap of its vacuole, is, on the other hand, permeable to nitrates, and to the salts of the other elements essential to plants. On this view we can understand how certain plants take up large quantities of non-essential minerals—just as sea-weeds absorb large quantities of salts of iodine. We suppose that large differences of permeability exist, at least, that the protoplasmic membrane of the root-hair cell of one plant is permeable to substances which cannot pass the plasmatic membrane of the root-hair cells of other plants.

The study of the behaviour of a plant, the absorbent organs of which find themselves confronted with new osmotic conditions, throws light both on the extent and limitations of the plant's powers of adaptation. Thus, the roots of willows grow in the strongly saline waters of the Dead Sea, and hence the root-hair cells, immersed in the salt water, the osmotic pressure of which is very high, must exert a correspondingly high osmotic pressure in order to retain their turgidity and to effect the absorption of water. Some marine algae withstand a change from salt water to fresh water, a change which would suffice to produce death from "osmotic explosion" in the cells of most plants. Brought into solutions of higher osmotic pressure, the absorbent cells of some plants react by setting up a countervailing osmotic pressure by the secretion of osmotic substances into the cell-sap. In other plants reach a similar condition of osmotic equilibrium with the fluid which surrounds them by absorbing osmotic substances from it. On the other hand, plants are not always able to readjust themselves so as to overcome unfavourable soil-conditions. Thus, a certain number of plants are very intolerant of chalky soils. Among such chalk-shy, calciphobe plants are sphagnum moss, the sun-dew (*Drosera rotundifolia*), and other plants common in boggy and peaty land, foxgloves (*Digitalis purpurea*), rhododendrons, azaleas, etc. The leaves of azaleas and rhododendrons

grown in soil rich in chalk, turn a sickly yellow colour, whilst the sun-dew and various mosses may be killed outright. It seems probable that though, for some reason or other, an excess of lime is injurious to these plants, their root-hairs are, nevertheless, permeable to soluble salts of lime, and have no power of modifying this property. In the cultivation of azaleas and other chalk-shy plants in greenhouses, gardeners are careful to water them with soft water, and advantage is taken of the susceptibility of various mosses to lime to get rid of them from lawns by watering with lime-water or by dressing the lawns with chalk.

Other plants are indifferent to chalk, and yet others (calciphil plants) show a marked preference for chalky soil (Bibliography, 1, 9, 17). Hence, in the cultivation of alpine plants from limestone regions, it is sometimes necessary to use stone containing lime for the construction of rock-gardens.

Our sand- and water-cultures have convinced us of the fundamental importance of the soil to the life of the plant. From it, the plant derives not only its water supplies, but also the mineral substances indispensable to its existence.

In addition to these fundamental relations between plant and soil, there are others hardly less striking, for even in the course of a walk across a small tract of country, it may be seen that the different kinds of plants are not scattered uniformly over it, but group themselves into bands, and that the kinds or types of vegetation characteristic of these several plant-associations are very different from one another (Bibliography, 1, 9).

Over that small tract of country the climate is uniform, throughout it the amounts of rainfall and sunshine are about the same. Therefore, we are compelled to conclude that the differences in vegetation which its several parts present are due, in some way or other, to differences in the soil.

The same conclusion is reached from a survey of the vegetation of the whole world: woodland and grassland occur the world over, alike in the hot regions near the equator, and in the cool temperate regions. The fact that the species of plants which compose a tropical

jungle differ from those which live in an English wood is to be attributed mainly to differences of temperature. Only plants that can withstand a fairly low temperature survive in an English wood, and only plants that can flourish at a uniformly high temperature live in the tropical jungle. But the *tree-type* of vegetation is common to both, and the existence of this type in such different regions is to be attributed to similarity of soil conditions, particularly, as we might guess, to there being a supply of water adequate for the support of luxuriant tree-growth. Thus, the question of the plant's water-supply is one which has many bearings, and is worth careful consideration. That this supply depends ultimately on amount of rainfall—or, in some cases, on supplies of underground water—is evident; though it is also evident, from the facts we have learned (p. 128), that plenty of water in the soil need not necessarily mean plenty of water available for the plant.

Consider, for example, how the vegetation of the land comes in our country to a sudden halt at the sea-shore. None of the plants that flourish a few yards inland are able to establish themselves on the sandy beach. That strip of ground is practically a desert, or at best bears occasional, isolated plants, such as sand-binding grasses (*Ammophila arundinacea*), yellow horned poppies (*Glaucium luteum*), sea thistles (*Liringtonium maritimum*), and sea convolvulus (*Convolvulus Soldanella*). Nor is the dearth of plants along the sea-shore confined to the sandy tracks. In the wet, clayey marshes which mark the estuaries of small streams, the vegetation, consisting of sedges, succulent *Suaedas* and *Salicornias* (Marsh Samphire, Saltwort), is but poor. Over these marshes, the sea spreads at periods of highest tides, and few plants can live in the soil impregnated with sea salts. Thus, within the space of a few hundred square yards three different types of vegetation may occur, the grassland or woodland, which comes almost to the sea's edge, the semi-desert of the sandy shore, and the salt-marsh. The reason for this grouping of plants according to soil characters is not far to seek. For sandy soils are leaky reservoirs, from them the water drains away almost as fast as it falls as rain upon them. In the salt-marsh also, the east-

ing vote which determines vegetation is given by the soil, for only plants (halophytes) whose roots are tolerant of large quantities of salt can live therein.

Hence a region of uniform rainfall may contain soils which, like those of the meadows or woodlands by the sea, are good water-reservoirs, others which, like the sandy shores, are poor reservoirs—physically dry—and others, like the salt-marshes, which, though they hold much water, withhold it from the majority of plants, and are therefore, though they may contain plenty of water, dry as far as plants are concerned, or, as we say, physiologically dry. Such brief considerations show that a study of the soil is a matter of first importance to the physiologist. To this study—of the physics, chemistry, and biology of the soil—we will now proceed.

Soil is "rotted rock." By the action of rain and running water, frost and other natural agents of destruction, the rocks exposed on the surface of the earth are in part dissolved and carried to the sea by rivers, and in part broken into the fragments of varying size which constitute the soil.

The soil thus formed may accumulate over the rock from which it is derived, or it may be carried by running water for some distance and then be deposited. In the former case, the soil is called a sedentary soil, and in the latter, a transported or drift soil.

Since different rocks are composed of different kinds of minerals, they give rise to soils which differ chemically from one another.

Thus, a limestone disintegrates to form a chalky soil. Rocks rich in felspar (silicate of alumina with potash, soda, or lime) give rise to a clay—a soil consisting mainly of fine particles of hydrated silicate of alumina. If much lime occurs mixed with clay, the soil is termed a marl. From rocks made up chiefly of quartz (silica) the coarser-grained soils, called sandy soils, are formed. Mixtures of sand and clay, in which the decaying remains of plants and animals (humus) have accumulated, are called loams, and constitute the most fertile soils.

The rarer minerals contained in the rocks scarcely affect the general character of the soil, though some are, as we

know, by no means unimportant to the plant. Of mineral substances, iron in combination is present in soils in sufficient quantities to give rise to characteristic colours—red, yellow, blue, or grey—which may be pronounced as to be recognizable even in soils whose owing to the humus they contain, tend to be of a bluish hue.

Fuller information as to the origin of soils may be obtained from text-books of Geology (Bibliography, 13, 17), and much may be learned concerning the origin and nature of the soils of a locality by studying them, the aid of the "drift maps" published by the Geological Survey.

113 For our first experiment in connection with study of the soil, we dig a large hole in a field or garden. As the earth is turned up and put on one side, note depth and also the change of colour as soil merges into subsoil.

Press a little of the soil between the fingers and whether it feels gritty or greasy. If gritty, we know it is of a sandy nature; if greasy, that it is clayey.

114 Heat a small quantity of the soil to redness in a porcelain crucible. When it has cooled, note that, as to the humus having been destroyed by burning, the incinerated soil has lost its blackish colour. Repeat observations on the subsoil, and thus determine that devoid of humus.

115 Half fill a test tube with a sample of the soil. Add a little hydrochloric acid—note whether effervescence occurs. If in doubt, hold the tube to the ear and listen for a crackling or bubbling sound—effervescence indicates that the soil contains carbonate of lime, which, on addition of the acid, disengages carbon dioxide.

116* Shake up samples of the soil with water in tall glass jars or test tubes. Pour off the liquid from one of the jars almost immediately, and from other jars at timed intervals, leaving, however, one undisturbed.

117.* By means of a long pipette, take out sample of the sediment from each jar and examine it microscopically. Observe that the larger soil-particles fall first. Mo-

drop of the turbid liquid contained in the last jar, look at it under the microscope and note the minuteness of the particles which it contains. Ascertain how long the water in the undisturbed vessel takes to become clear.

118 If several days elapse before this happens, pour half the contents into a clean vessel and add a little lime-water. Observe that the effect of lime is to precipitate the fine particles, and note that lime is used in agriculture for the purpose of making clay soils more workable.

119. Throw back the soil into the hole (Exp. 113) without treading it down. Note that the earth by no means completely fills the hole. Hence, even though put back with care, it is more tightly packed than before it was dug out, and hence, also, the particles, in their natural position in the soil, are separated from one another by spaces.

120 Plunge a flower-pot, containing a plant not watered recently, in a pail of water: note that, as the water enters, bubbles of air escape from the spaces between the soil-particles.

121 Take two similar plants in pots, e.g. geraniums, or chrysanthemums, submerge the pot containing one plant in water in a pail, grow the other plant in the usual way. Note that the plant in the water-logged soil fails to flourish. After some time, turn it out from its pot and observe that the roots appear unhealthy and, it may be, rotted. Excess of water means deficiency of air, and hence an unsatisfactory rooting medium (see also Exp. 67).

From the foregoing experiments and considerations, it is evident that a knowledge of the physical, as well as of the chemical, properties of soils is essential to an understanding of the relations which obtain between plant and soil.

We will, therefore, devote ourselves to a brief experimental study of some of the more important physical properties of soils.

122 For this purpose, obtain samples of fairly fine sand or of a sandy soil, and of a heavy clay.

Having dried the soils in an oven, pound and sift them to remove stones, etc. Without pressing down the particles, fill two pots of equal size and of known weight, e.g. marmalade jars or pint pots,—one with the sand, the other with the clay. Determine the weights of

the equal volumes of sand and clay. Note that sand is heavier than clay, and observe, therefore, that the farmer, when he speaks of a clay soil as 'heavy,' means that it is heavy to work on account of its tenacious nature.

Next compare the water-capacities of the two samples of soil, that is, the amount of water which can be taken up by an equal volume of each.

123 To do this, fill a graduated burette with water, and run the water from the burette gradually into the vessels containing the sand and clay used in the last experiment. As soon as the soils can absorb no more water, weigh the vessels, and remembering that 1 cubic centimetre of water weighs 1 gram, calculate, from the increased weight, the volume of water absorbed by the sand and clay respectively.

Since both sand and clay consist of solid particles, the water added to the vessels has passed into the spaces between the soil particles, and thus the amounts of water taken up by the sand and clay give us a measure of the total volume of the spaces between the soil particles, or, in other words, of the pore-space of each soil.

124 Turn the wet soil out of the pots, clean, dry, and fill them again with samples of dry, sifted soil—sand and clay as before. Shake down the soil particles by tapping the vessels repeatedly against the table, observe that, as a result of such treatment of the dry soils, the clay particles lie closer to one another than do those of the sand, and that in consequence, the space now occupied by the clay is considerably less than that occupied by the sand.

125 * Mount a small quantity of the sand in a drop of water on a glass slide, and make a similar preparation of the clay. Examine microscopically, and observe that the sand particles are much larger than those of the clay (cf Exp 117*). Since a relatively large particle of sand is heavier than a minute particle of clay, sand particles lie more heavily on one another than smaller and lighter clay particles. We thus recognise that the much larger pore-space of the clay soil (Exp 123) is due, not to the *chemical* properties of clay, but to the fact that it consists of minute particles which lie lightly on one another, and so leave considerable spaces between

Since they are so small individually, the number of clay particles in a given volume of clay soil is considerably greater than that of sand particles in an equal volume of sandy soil.

Our experiments teach us further that, after a heavy rain-fall over fields, some of which have a clay and others a sandy soil, assuming that the layers of soil are of equal thickness, more water is contained in the former than in the latter.

But, as we know, the rain which falls on fields disappears gradually, and we must ask whether the "drying-up" proceeds more quickly in one kind of soil than in another.

If we have at any time observed the sand on the seashore, we must have noticed that in summer, soon after the tide has receded, its surface becomes quite dry, whereas the clayey ooze lower down the beach remains for hours glistening with the water which remains on its surface. Evidently, water does not drain away so quickly from a clayey as from a sandy soil.

How marked is the difference between different kinds of soil in this respect we may demonstrate as follows —

126 Procure two *glazed* flower-pots of equal size, each with a hole in the bottom. Close the holes by means of tight-fitting corks and determine the volume of each pot by measuring the amount of water required to fill it. Remove the corks, dry the pots and place a small filter-paper in each to prevent the soil from escaping through the hole. Fill loosely one pot with dry, sifted sand, the other with dry, sifted clay.

Add water gradually as in Exp. 123 until the soil is saturated and the pots can hold no more. Cover each pot with a glass plate, stand it in a saucer or bowl, weigh, and record the weights. Remove the pots from the saucers and stand them on blocks of wood or retort stands so that the water is free to drop from the holes. Compare the rates at which the water drains away. After some hours weigh the pots again and estimate the relative rates of percolation of the water.

We learn thus that clay holds more water, and holds it more tenaciously, than sand. Whence it follows that, since the water is held in the pore-spaces, a sandy soil

becomes dry more quickly than does a clay, and the latter tends to become water-logged—and, owing to it, sour—more readily than does a sandy soil.

The bearing of these facts on the drainage of agricultural land is evident.

127 When water has ceased to drip from the two of Exp. 126, weigh them. Knowing the total amount of water held by known volumes of similar samples of sand and clay (Exp. 123), and knowing the volumes of glazed pots, calculate the amount of water held by sandy and clayey soils respectively, after percolation ceased.

128 Now remove the glass covers from the pots, placing the latter where they are sheltered from rain, the soil exposed to the air. By weighing at intervals ascertain that the sandy soil loses water by evaporation more quickly than the clay.

In order to understand how this comes about, we determine how the water is held in the soil after percolation has ceased. We obtain an insight into the matter by the force by which the water is held on the surface of soil particles by the following experiment.

129 Take a series of capillary glass tubes, some wider, and some of narrower bore—the bore of the largest being about 1 mm., and that of the smallest about .25 mm. Stand the tubes vertically so that their lower ends dip into water, which may be coloured with methylene blue or any dye. Determine the height to which the water rises in the several tubes, and note that, the narrower the bore, the higher the capillary rise. Note also that the upper surface of the water in the capillaries is not flat but concave, rising higher at the sides where it is in contact with the glass than in the middle, and observe that the narrower the capillary, the greater is the concavity of the surface of the water contained in it. Since, to lift the water up the tube, work must be done, it follows that the inner surface of the tubes exerts a pull on the water, and that the narrower the tube the greater the pull. The force which holds up the water-columns is called surface-tension, and the water is said to rise in the tubes by capillarity. Tubes narrow as to exhibit the phenomenon are called cap-

tubes. Now we have already discovered (Exp 123) that the soil is penetrated in all directions by exceedingly fine spaces, and since water, added to the surface, drains away below, it follows that these spaces form a system of intercommunicating channels. With this conception of the soil, it is easy to understand how it is that after water has percolated from a soil the latter still contains a large amount of moisture. We conceive of the water held by the soil as forming films of greater or less thickness around each particle, and of being held thus with considerable force. If a soil consists of extremely fine particles, the number of particles per cubic inch is far larger than that in an equal volume of soil built up of coarse particles. Hence the total surface of the particles of this volume of the fine soil is far greater than that of the particles of the coarse soil.

In illustration, we may quote from Mr. A. D. Hall's book on the soil (Bibliography, 16), the extent of the surface of the particles of a clay soil, a loam and a sandy soil —

Pore space %.	Area of surface of soil particles in sq. ft. per cubic foot of soil
Fine clay soil - 48	110,500 square feet
Loam - - 44.1	46,000 " "
Sandy soil - 32.5	11,000 " "

From these numbers it is evident that the total surface presented by the particles of a clay soil is about ten times greater than that of the particles of an equal volume of sandy soil, and the number of water-films held tenaciously by surface-tension is, therefore, much greater in the case of the clay soil.

This picture of water in the soil held in the form of films or hollow shells on the surfaces of the soil particles helps us also to imagine what happens as the soil dries by evaporation. As the surface layer exposed to warm air or dry winds loses water in the form of vapour, the water lost is replaced by that attracted from the particles

below—for the thicker the film the less firmly it is held—and conversely, the thinner the film the greater the surface-tension exerted upon it by the soil particle. Thus, as evaporation proceeds, there is a redistribution of water in the soil—in other words, there is a *capillary rise*. The finer the system of capillary spaces, the more thorough-going is this upward movement of water from the surface of one particle to that of another. We investigate the rate of capillary rise in various soils, *e.g.* sand, clay, and also chalk and humus thus—

130 Take a long piece of fairly wide glass tubing and cut it into lengths of about 1 or 2 feet each. Fill flower-pots, one with dry, sifted sand, others with clay and powdered chalk respectively (another may also be filled with finely divided peat). Push a tube into each flower-pot so that it is held firmly, pour finely-sifted, air-dry sand into the tube standing in the sandy soil, clay into the tube in the clay soil, etc., tapping gently the sides of the tubes to pack the materials. Plunge the flower-pots in water to their rims and observe, by noting the change of colour of the soil, the rates of ascent of water in the tubes. Contrast the capillary rise in the sand, clay, chalk, etc. Draw or photograph the apparatus at the end of the experiment, and preserve the record in the museum.

The picture of the structure of a soil which arises in the mind as the result of our experiments is that of innumerable particles separated from one another by spaces. Around each particle is a film of water, and the remainder of each space is occupied by air. As the root absorbs water, the films around the soil-particles in the neighbourhood of the root-hairs become thinner. The surface-tension which the particles exert draws water from the films around the neighbouring soil-particles. This in turn is absorbed by the roots, which thus obtain water from a large area of the soil. In course of time, if no fresh supplies of water reach the soil, the osmotic pressure set up by the root-hairs is counterbalanced by the surface-tension exerted by the soil-particles, and the root ceases to absorb water.

The soil, however, loses water not only to the plant, but also by percolation and by evaporation. Hence the horti-

culturist, who cannot always supply water to the soil, has often to take steps to preserve the water in a drying soil.

This may be effected, for example, by spreading lightly over the surface of the soil a layer of fine particles, *e.g.* of finely-sifted, dry soil. By this means, the continuity of the network of capillary tube-like spaces, which extend from the depth of the soil to the surface, is interrupted at the surface, and, evaporation of water from the top layer being checked, the capillary rise of water from the deeper layer ceases. It is as though an infinite number of miniature bungs were put in the necks of as many capillary bottles. On the other hand, if the fine surface-layer is pressed down firmly, for example, by rolling, and its particles thus brought into close contact with those of the layer beneath, capillary continuity is re-established, the particles of the top layer are able now by surface-tension to obtain fresh supplies of water from that surrounding the particles with which they are in contact, and hence, as the water of their films evaporates, there is established an upward movement of water from particle to particle. Again, if, instead of finely-divided soil, a loose layer or mulch of material, such as straw, or grass cuttings, or leaves, is placed on the surface, evaporation is checked. This layer, not in capillary continuity with the soil-particles, may itself become dry, but cannot withdraw the water from the soil below. Thus, the capillary rise being stopped, the water in the soil is conserved.

It is of interest to observe how agricultural and horticultural practice have for one of their main objects the control of the water of the soil. Nor, when we realise the water-requirements of plants, is it surprising that this should be the case.²

The rainfall of this country, though spread over many days, is not great. In the neighbourhood of London, the annual rainfall is about 24 inches. Of the rain which falls on a field, some runs off the surface, much passes by percolation to the deeper layers of the soil and is carried away by drainage, much is lost by evaporation during those parts of the year when the fields are unoccupied by crops. Only that which remains is available for the plants.

That there is apt to arise a shortage of water, and the crop is likely to suffer, is evident when we consider how considerable are the water-requirements of various crops. Thus, the amount of water taken from the soil by a crop of wheat is equivalent to a rainfall of the whole wheat-field of no less than six inches. Other crops are yet more thirsty; mangolds, for instance, take from the soil the equivalent of 106 inches, or some little half the total, average annual rainfall in southern England.

It is in consequence of the annual risk of water-flood among crops that tillers of the land are at such pains to interfere with nature in the interest of the crops to cultivate. Thus, autumn ploughing, by breaking up surface layers of the soil, increases its water-holding capacity and, as the frosts of winter cause the debris to crumble, the water is conserved by the interrupted capillary continuity.

The practice of hoeing, resorted to by all good gardeners, has a doubly beneficial effect. In the first place, it kills down weeds, which compete with the cultivated plants for the water or other materials in the soil. In the second place, by breaking the upper layer of earth, a film is established and water is conserved in the soil.

How important are the effects of hoeing may be seen from the following figures, which give the loss of nitrogen from hoed and unhoed areas of soil, deduced from an experiment :—

Daily loss of water from cultivated soil, 14.5 tons of water per
 " " " uncultivated " 17.6 " "

In this experiment, soil cultivated to a depth of 3 was sampled at intervals to a depth of 6 feet. (samples were taken simultaneously from adjoining with a firm surface.

There was thus during the 30 days of the experiment a saving of water corresponding to 1.7 inches of rain.

We must not bring these studies of the soil to a close without considering the soil, albeit in the briefest manner, from another aspect, namely, the biological.

The soil is rich in many forms of life. Beside the flow

plants, it bears ferns and toadstools. Sometimes a green scum of minute algae may be seen on its surface, and by appropriate methods a rich flora of non-green organisms (fungi, bacteria, and also a fauna of minute, unicellular animals) may be discovered to have their homes in the upper layers of the earth.

Though, however, the parts played by the soil-flora and fauna in determining fertility are of the greatest importance, we can consider them here in the light of but one example.

131 Repeat Exp. 106, but mix with the sterile sand in one of the pots about an ounce of ordinary garden soil.

Water this pot, and those containing sterile sand only with culture solution lacking nitrogen (Appendix A). Observe that the peas growing in sand to which the earth was added, though they may remain puny for a while subsequently grow as vigorously as those planted in garden soil. Those in sterile sand remain small. After some weeks, turn the plants out of their pots and observe that whereas those which have grown vigorously have curious nodules or swellings on their roots, those in the sterile sand have none. Keep the specimens for the museum.

132 Germinate seeds of gorse (*Ulex europæus*), some in sand sterilized by heating in a hot oven, and others in ordinary soil. Observe the development of the seedling and the relation between their growth and the presence of nodules. Dig up any leguminous plant from the field or garden (vetch, lucerne, lupine, etc.), and observe its root nodules. From laborious investigation of these root nodules, it has been proved that they are the result of an infection of the root by a definite micro-organism, *Pseudomonas radiclecola*, and it has been demonstrated that the micro-organism, even when isolated in artificial cultures in the laboratory, has the power of obtaining its nitrogen from the free nitrogen of the air. Thus it does also when it lives in partnership (symbiosis) with the roots of leguminous plants. These plants, however, not only tolerate the micro-organism which multiplies in the cells of their roots, but eventually digest it, and so secure the nitrogen which the micro-organism has amassed. We thus learn that micro-organisms exist in the soil which have the power to "fix"

free nitrogen," that is to bring the nitrogen gas of atmosphere into such combinations as serve plants food-materials.

We may understand how all-important this fact is, if we remember that the amount of combined nitro (nitrates, etc.) in the soil at any given time is but small, that plants are constantly taking toll of it, that, in the decay of plants and animals, the complex proteins are broken down by various micro-organisms, in ordered stages, to form ammonia and ultimately nitrogen. The nitrogen so formed escapes into the air, as may also some of the ammonia. There is thus, as it were, a constant leakage of nitrogen from the living world. By the agency of the nitrogen fixing bacteria, some species of which occur in the nodules of leguminous plants and other kinds of which live exclusively in the soil, this escaped nitrogen is once more brought into combination and rendered available for nutrition of the higher plants.

It is more than probable that the general shortage of nitrogen in earth and sea has led to other relations between organisms as remarkable as that between leguminous plants and *Pseudomonas radicola*.

133 * Thus, when we examine with lens and microscope the young roots of forest trees, e.g. pine, oak, hornbeam, etc., and also those of heath plants, we discover that instead of root-hairs, the young roots are covered by mantles of fungous threads (mycorrhiza).

It is not unlikely that, as a result of this symbiosis, the flowering plants which take a part in it obtain increased supplies of nitrogen. The lichens which grow on rocks and tree stems and other situations represent symbiotic partnerships between fungi and algae. No few animals, of which the green hydra is an example, are associated always with green algae. It is possible that all these strange symbiotic unions have for their significance the solving of the problem of obtaining adequate supplies of nitrogen.

CHAPTER IX

THE absorption and loss of water by the plant. The water requirements of various types of plants —hygrophytes and xerophytes. The process of the transpiration of water by the leaves. The structure of the leaf in relation to this process. The part played by stomata. The opening and closing of stomata and the conditions under which these movements occur. Apparatus for measuring rate of transpiration—(Potometer)

Now that we have completed our studies of the properties of the soil and the bearing of these properties on plant-life, we have to discover what becomes of the water and mineral substances which are taken up by the root-system.

As a preliminary to this work we proceed to ascertain experimentally the amount of water absorbed by the roots of various plants. The most accurate method is to measure absorption directly, a simpler way is that given in Exp 137.

134 To employ the first method, germinate maize or other seeds used in water-cultures (Exp 109) so that the roots of the seedlings grow into a normal culture solution. Prepare two tall glass vessels, each provided with a graduated side tube (Fig 23). Fill the vessels with normal culture solution to within an inch or so of the top, transfer a well-developed seedling to one of the vessels so that its roots are in the solution. Fix its stem in a split cork which fits into the neck of the vessel. By means of paraffin or wax, (Appendix A) make air-tight the junctions between cork and glass, and cork and seedling. Plug the side tube with a wad of cotton wool to keep out dust. Wrap black cloth round the vessel in order to exclude light from the roots and thus to prevent the growth of

green algae in the culture solution, but leave the side uncovered. Fix a thermometer by rubber bands to side tube. Proceed in a similar way with the other vessel. Label the vessels A and B, place them in a good light, continue, for as long a period as possible, to take records, by reading the level of the fluid in the side

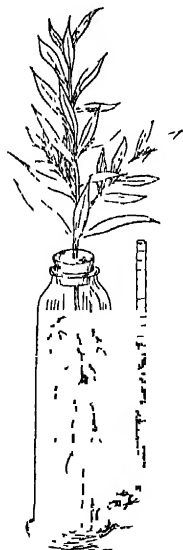


FIG. 23 — APPARATUS FOR MEASURING ABSORPTION OF WATER BY PLANTS

of the amount of water absorbed by the plants. Note day the temperature of the air and also the state of weather, fine and bright or wet and dull. Add culture solution whenever necessary.

Determine the total amount of water absorbed by A and B, plot the daily amounts on squared paper, and "curves" of the amounts of absorption. Observe relation which appears to exist between weather and amount of absorption. Since the volume of water absorbed by

plant during the course of the experiment is many times that of the plant itself, it follows that the water is either decomposed or else given off by the stem or leaves. The readiest method of demonstrating that water is given off from the plant consists in covering it with a bell-jar and observing that drops of water condense on the inner surface of the glass. Or we may use our plants A and B for the purpose thus —

135 Weigh the apparatus containing plant A. After two or more days weigh it again. Determine that the loss of weight of the apparatus corresponds approximately to the weight of water absorbed by the root during the period of the experiment.

Next demonstrate, in the following way, that the loss of weight recorded in the preceding experiment is due to loss of water by the plant.

136 After weighing apparatus B and recording the level of the liquid in the side tube, close its side tube with a cork and cover the plant in B with a large bell-jar, under which is placed a weighed vessel containing calcium chloride. The rim of the bell-jar must be sealed by wax to a flat block of wood or to the table, in order to prevent the calcium chloride, which is very hygroscopic, from absorbing water from air entering the bell-jar from outside. After 12-24 hours remove the bell-jar and also the cork from the side tube, weigh B and weigh the vessel containing the calcium chloride, by reading the level of the liquid in the side tube, determine the amount of water absorbed by the plant during the course of the experiment. Observe that the loss of weight of B is about equal to the gain in weight of the calcium chloride. Since the increase in weight of the latter is due to absorption of water, we infer that the loss of weight by the plant is due to loss of water. Hence it follows that the water absorbed by the root-system passes up the stem and is given off into the air in the form of water vapour, leaving behind in the tissues of the plant the mineral substances which it held in solution. Evidently therefore one important function of the water absorbed by the root is to act as a vehicle for the carriage of the mineral salts from the soil *via* the root to the shoot. The

fact that the loss in weight of plant B (Exp. approximately equal to the weight of water absorbed by the root during the same time does not mean that other processes involving change of weight are going on in the plant, but that the amount of change is due to such processes is, during the short time experiment, small in comparison with that due to water.

The second and simpler method which we use to determine absorption is based on the fact, which we have just stated, that the amount of water absorbed in a given time by a herbaceous plant is about equal to the amount lost as vapour. It consists in putting pot-plants in conditions that evaporation from soil and pot is prevented, and then weighing the plants at intervals. Loss of weight is due mainly to loss of water. Amount of loss of weight is approximately equal to amount of absorption. Loss of weight gives a rough measure of amount of absorption.

137 The apparatus is made as follows.—Choose leafy sunflower or dwarf bean plants grown in snail shells. Procure marmalade or other jars, each large enough to hold one of the pots. After watering the plants thoroughly, drain the superfluous water drain away and stand the jars on a sheet of American cloth cut out two circular pieces of tile placed at the bottom of the jars. Make a cut in the cloth to serve as covers for the jars. Make a cut in the cloth extending from the circumference to the centre. From the centre, cut out a small circular piece to allow of passing through. Place each cover in position and cut edges together by means of stitches, turn the covers down over the rim of the jar and tie it firmly by several turns of stout thread. Or the same end may be reached by preventing evaporation from the soil, etc.—may be reached by getting a tinman to cut out for each jar two circular pieces of tin, each notched to enclose the pot. Two such tin plates, when fixed by wax to one another to the edge of the jar, make a good cover. Use pots A and B. It will be necessary from time to time to remove the covers and add measured quantities of water. From the series of daily records of the loss of weight

the plants, calculate the total amount of water which they have given off during the time of the experiment (cf Exp 136).

We determine next that the water given off escapes mainly from the leaves.

138 After weighing the two vessels with their plants (using the apparatus either of Exp 134 or that of Exp 137), remove about half of the leaves from plant B. Weigh the apparatus again, replace it and take a daily or 48-hourly record of its loss in weight. Whilst the cut-off leaves are fresh, determine their area (Appendix B). Note that the rate of loss of water by the plant from which leaves have been removed is considerably less than it was before. Remove the remaining leaves and estimate their area. From the records obtained, calculate the rate of loss of water *per unit of area*, say per 10 sq cm of leaf surface. We thus demonstrate that the amount of loss of water, or, as we say, the amount of water transpired, is roughly proportional to extent of leaf surface, and infer therefore that the water escapes mainly from the leaves. Since a small quantity of water is also given off from the stem, even after the precaution of covering, e.g. with collodion, all the wounds formed by removal of the leaves, it may be necessary, in making this calculation, to deduct the amount of water given off by the leafless plant from the amount given off by the plant with half its leaves, and from that with all its leaves intact. In the foregoing experiment, plant A—with its leaves intact—serves as a control. We know the relative rates of transpiration of A and B over a period of many days. We know the rate of transpiration of A during the period of the experiment. Hence we can calculate approximately the amount of water which B would have lost if its leaves had not been removed, and we are able to compare the rate of loss of B with all its leaves with that of B with half, and with none of its leaves. By continuing to record the transpiration of plant B for some days, we may observe that though it is very much reduced, some loss of water occurs after all the leaves have been removed, and conclude that, though the leaves are special organs for allowing of the escape of water vapour, other parts, par-

ticularly the delicate green parts of the stems, may a give off water vapour.

Plant B is now useless for further experiment, but plant A should be grown for as long a period as possible, & regular records of the rate of transpiration made, tabulated and the record-chart kept in the museum.

Having gained precise information with respect to large amount of water taken up and given off by plants with thin delicate leaves like maize, sunflower or bean, we proceed to find out whether plants with other types of leaves are equally spendthrift of water. For this purpose, in any pot-plants the leaves (or stems) of which are thick & fleshy, e.g. Agave, india-rubber plant (*Ficus elastica*), a cactus with fleshy stems, a *Mesembryanthemum*, or, fail any of these, gorse (*Ulex europæus*) grown in pots.

139 Having fitted up a broad, thin-leaved plant (e.g. sunflower, in a maimalade pot A (Exp. 137), in apparatus B one of the just-mentioned plants. Determine, by weighing daily, the rates at which A and B lose water. After some days, determine the areas of the leaves and stems of the two plants and calculate their rates of transpiration per unit of area.

The following is a record of the rate of transpiration of two plants placed under similar conditions for twenty-four hours. It shows that the rate of transpiration per 100 sq. cm. of surface was 32.7 times as great in the case of *Hydrangea hortensis*, a plant with relatively large and thin leaves, as it was in that of *Opuntia cylindrica*, a cactus with a fleshy green stem and no leaves.

	Transpiration per 100 sq. cm. of surface
<i>Hydrangea hortensis</i> -	6.54 gm.
<i>Opuntia cylindrica</i> -	0.20 "

We learn from these experiments that different kinds of plants have very different water requirements. Some lose water and hence if they are not to wither, also take up water, much faster than others. Not only is the loss of water per unit of surface much less in succulent & leathery-leaved plants than in plants with delicate leaves, but also the total leaf- (or green stem-) surface of the former is much less than that of the latter. Since such succul-

nts live generally in dry situations, we may infer that reduction of surface, the succulence of the leaf or stem, thick cuticular covering of leathery leaves, the matted is common on such Alpine plants as the Edelweiss (*Leontodon alpinum*) are so many adaptations making water economy. Such plants find water either hard to (desert-plants, alpiners) or hard to keep (plants of hot ntries, agaves, cactuses, etc.) or both. Plants which nage to live in such situations do so by definite adaptation of their root-systems and shoot-systems. The root-tem of desert-plants is often of extraordinary length—roots, as it were, going in search of water. The shoot-tems display the most diverse adaptations serving to uce the amount of water given off from their surfaces. se water-economising plants are known as *xerophytes* (contradistinction to *hygrophytes*). The latter are as ravagant in the amount of water which they absorb and e off as the former are niggardly. Among British plants, note that those of sandy soils and er dry situations are xerophytes, e.g. the Scotch fir (*Pinus sylvestris*), gorse (*Ulex europæus*), broom (*Cytisus parvus*), heaths and various grasses which grow in n company.

40. Specimens of these xerophytes and also of the moner hygrophytes—broad-leaved trees, etc.—should collected, compared, and kept in the museum.

aving learned something of the water requirements of ous kinds of plants, we next enquire how the water and s absorbed by the roots pass to the leaves, and how the es get rid of the large quantities of water which reach m. We will first consider the latter question.

clothes hung upon a clothes line in the open air dry soon he weather is bright and warm. Since, in the course of s 134 and 137, it has surely been noted that plants lose er more quickly on a bright or warm day than on a dull old day, it might be supposed that the loss of water by leaf is, like that of clothes hung out to dry, due solely to poration. But, if this were the case, the plant would be gether at the mercy of its surroundings. In hot, dry ither, when evaporation is high and absorption, owing to nness of soil, is low, the plant would wither, wilt and die.

So it may, if the drought is prolonged, but not with inevitableness of the drying of clothes in bright weather. From their behaviour in spells of dry weather, it looks as though plants did not give way to drought without struggle, and that, though dry air causes evaporation of water from clothes and plants alike, the latter, unlike the former, have to some extent at all events, means of preventing excessive losses of water from their tissues. We can imitate the resistance of the plant to the last stages of desiccation in the following experiment —

141. Weigh two large saucers or basins, fill one with a known volume of water, the other with an equal volume of a fairly strong solution of sugar. Weigh the vessels containing the water and the sugar-solution. Expose them in a warm place, and determine, by weighing, the rates of loss due to evaporation. Note that, as the sugar-solution concentrates, its rate of loss of weight falls off very considerably. Consider now a vegetable cell. Within its protoplasm (plasmatic) membrane is a vacuole containing water, etc., in solution. Enclosing it is a cell-wall substance and protoplasm contain imbibed water. Imagine a tissue made up of a number of such cells exposed to a dry atmosphere. The cell-wall loses water by evaporation and withdraws water from the protoplasm, which in turn withdraws water from the cell-sap. The cell becomes more concentrated, and hence its osmotic pressure, *i.e.* its attraction for water, is increased. As evaporation proceeds, not only is the imbibed water remaining in the cell-wall and protoplasm held more strongly, but also the concentrated cell-sap, unless it is able to withdraw water from neighbouring cells and to become dilute again, offers more and more opposition to evaporation. Thus we understand one way in which the plant may resist extreme desiccation. Before deciding, however, that this is the only method, we must look at the leaf itself, for its structure may throw further light on the subject.

On inspecting various leaves, we observe that in many of them there are marked differences between their upper and lower surfaces.

142. Compare, for example, from this point of view

the following —beech, elm, holly, tulip, hyacinth, laurel, etc. Note that in leaves which, in their natural position, lie more or less horizontally, the under side is paler than the upper side. The surface exposed to the direct rays of the sun appears to be less delicate than the other, and looks as though it would allow water to evaporate from it less rapidly. We prove that this is the case thus —

143 Take two similar leaves of the india-rubber plant (*Ficus elastica*) or the laurel (*Prunus Laurocerasus*). Slip a piece of thin rubber tubing over the stalk of each leaf. Turn back the tubing, and, by means of wire, bind the loop thus made so as to block the hole in the tube and to make a convenient hook. Smear with vaseline the lower side of one leaf and the upper side of the other. Weigh each leaf. Hang up the leaves in a room, and determine, by weighing at daily intervals, the rates at which the leaves are losing water.

144 Not all leaves, however, show this marked difference between under and upper side, and if the above experiment is repeated on the leaves of beech or broad bean, using in these cases *similar branches* with equal numbers of leaves, no marked difference between under and upper sides with respect to resistance to evaporation will be found. This, however, does not affect the fact shown by Exp. 143, that plants which require to reduce evaporation, and yet, at the same time, are obliged, for other purposes, to expose a considerable leaf-surface to the light, produce leaves the upper surfaces of which hinder evaporation more than the lower. Microscopic examination will help us to learn more as to the nature of the differences between the upper and lower surfaces of leaves.

145 * Mount a small piece of a leaf of the frogbit (*Hydrocharis Moerhousiana*) or water starwort (*Callitriche verna*) in a drop of water, and examine it with low and high powers of the microscope. Observe that, in one or other or both surfaces, there are large numbers of extremely minute holes. These holes or *stomata* cannot be seen in thicker leaves when whole pieces are mounted. Therefore, in order to examine ordinary, thick leaves for stomata, strip off, by means of fine forceps, or cut with a razor, pieces from the upper and lower surfaces of the leaves of the

plants mentioned below, mount each piece in water on slide, and determine that, in the india-rubber plant and laurel, stomata occur only on the under side of the leaf whereas in the bean, beech and tulip they occur on both sides. The fact that the surfaces of leaves are not continuous but are interrupted by innumerable openings throws a new light on the results of Expts 143 and 144, and at once suggests the idea that water-vapour may escape not only from the general surface, but also through the stomata. This becomes the more likely when we discover, by the following experiment, that the stomata are in communication with spaces running between the groups of cells (which the soft tissues of the leaf are composed of).

146. Attach firmly by wire a small piece of rubber tubing to the stalk of a delicate leaf, e.g. marsh marigold (*Caltha palustris*) or wild arum (*Arum maculatum*). Immerse the blade in water and suck at the open end of the tube. Note that the leaf becomes darker green owing to the entrance of water through the stomata. If the surface of the leaf is first smeared with vaseline, the stomata are blocked and no water enters. Or, if an air pump is available, connect it with the side tube of an Erlenmeyer flask (Appendix B) containing water. Place a delicate leaf beneath the surface of the water. Having closed the opening of the flask with a rubber cork, exhaust the air by means of the pump. Note that, as the pressure in the flask is reduced, bubbles of air escape from the cut end of the petiole and water enters through the blade. Hence the stomata communicate with spaces—intercellular spaces—which run between the cells of the leaf. Hence also evaporation is possible not only from the surface tissues of the leaf, but also from the cells of deeper tissues abutting on the air spaces. For, if the air at the surface of the leaf is dry, water-vapour passes by diffusion from the air spaces through the stomata. Thus the air in the spaces becomes drier, and, in consequence, water evaporates from the cell-walls, in short, the internal cells neighbouring on the air-spaces lose water by the same process as that described for the cells on the surface of the leaf (p. 150). We are not, however, to suppose that the purpose of this system of spaces and of stomata is neces-

sarily to facilitate transpiration. Their main function, for aught we know at present, may be of quite another kind, e.g. for allowing air to circulate freely in the leaf. But whatever their main function may be (see p. 161), the facts remain, that they are there, and that, therefore, they cannot but serve as channels through which water-vapour, given off by the cells abutting on them, escapes into the air. That water-vapour does escape from the stomata we demonstrate thus —

147 Soak strips of filter- or blotting-paper in a 10% solution of cobalt chloride. Dry them in a desiccator, and when not in use, keep them there or in a closely stoppered bottle. Expose a strip to moist air, e.g. breathe on it, and note that it changes colour. Cut off a leaf, place on either side of it a strip of cobalt chloride paper covered by a sheet of thin glass or mica, bind the glass close to the leaf by means of rubber-rings, lay the leaf on a clean table and cover it with a bell-jar or glass dish. Observe the change of colour on one or both sides of the leaf and determine that stomata occur, in the one case, on one side, and, in the other, on both sides. Instead of being cut off, the leaf may be left attached to the plant during the course of the experiment.

148 * If a micrometer is available, determine (1) the number, (2) the size of the stomata either on strips from the surfaces of the leaves or on the whole leaves of Exp. 145. It may thus be shown that the number of stomata in a leaf is almost incredibly large, ranging generally from 100 to 700 per sq. mm. of leaf surface, i.e. from 50,000 to over 350,000 per square inch. The size of each stoma is as small as the number of stomata is large: on the average a stoma is about 0.006 mm. in diameter, i.e. 0.00024 inch.

Our discovery that the surface of a leaf is riddled with holes of microscopic size is a fact which has evidently to be taken into consideration in studying the mode by which water is given off by green plants. As we have proved, the water lost by the leaf escapes in large measure not from the general surface but through the stomata. Now, when we were examining these pores (Exp. 145), we could scarcely fail to notice that each of them is bounded on

either side by a cell, the shape of which is different from that of the other cells of the surface layer (epidermis). The cells which encompass the stomatal opening are called guard-cells. In surface view, each guard-cell appears more or less sausage-shaped. The guard-cells have the further peculiarity that, whilst they are connected with one another at their ends, they are separated in the middle line, the cleft between them constituting the stomatal pore.

Having obtained an idea of the nature of the stomatal apparatus, we return to our problem—how is water lost by the leaf merely a matter of evaporation, or is there some power of regulation? The only parts of the apparatus concerned in the regulation of water loss are the stomata. Now, stomata are holes—slits between cells. The plant could control the size of the stomata, and thus the amount of water-vapour escaping through them. It might perhaps be reduced in amount. But if the stomata change in size, they can do so only by change in the shape or size of the guard-cells. Therefore, we have to ask: are the guard-cells fixed in position or are they movable?

We know already that a cell, when it becomes turgid, increases in size, and that, when it is plasmolyzed, it shrinks. We proceed to ascertain whether any change involving change in the stomata, occurs when guard-cells are rendered turgid or when they are plasmolyzed.

149. Repeat Exp. 145 on surface-sections of *Tradescantia* or other convenient plants.

The leaves of the following plants serve well for observations on stomata, as the epidermis may be removed by the aid of forceps: *Primula sinensis*, *Prunella*, *Tropaeolum majus*, *Pelargonium zonale*, *Impatiens*, *Viola*, *Faba*.

Note that whereas, when they are in water, they are large or, as we may say, open, when a 10% salt solution is run in they become small, that is, they close. In that this change is effected by a change in shape of the guard-cells. A complete understanding of the mechanics of this remarkable movement of guard-cells involves a detailed study of their structure, a study which we cannot now embark upon (Bibliography, 5, 6).

Knowing that stomata are capable of being increased or decreased in size, we next enquire whether they actually change in size during the life of the plant, and if so, what are the conditions which determine the movements of the guard-cells by which the opening and shutting are effected? We therefore investigate the state of the stomata in plants which have been subjected to diverse conditions, e.g. bright light, darkness, dry air, moist air, lack of water at the root, etc.

To make the observations, we require a better method than that employed in Exp. 145, though the plants used in that experiment will serve admirably. The method we use is as follows —

150 * Expose growing pot-plants for an hour or so (1) to bright light, (2) to darkness, (3) to dry air (see Exp. 136), and also cut off leaves and allow them to wither. Throw the leaves treated in these ways into bottles containing absolute alcohol, stopper and label each bottle, expose the bottles to sunlight till the leaves are colourless. Transfer the leaves to another bottle of absolute alcohol. After an hour, take them out and plunge them into a bottle containing xylol. When the leaves are transparent, mount small pieces in cedar wood oil and examine under low and high powers of the microscope. Conclude from the experiment that the stomata of the leaves of many plants close in darkness, in dry air, when the soil is dry, and when the leaves wither, and that, on the contrary, they open in bright light and in moist air.

It is evident that certain of the conditions which make for the closing of stomata are conditions under which loss of water by evaporation tends to become excessive, e.g. dry air. On the other hand, it is *not* evident that change from light to darkness—apart from attendant temperature change—results in a reduction of the rate of evaporation. Let us ascertain, therefore, whether change from light to darkness produces any change in rate of loss of water by the plant. If it does, we shall be justified in concluding that the opening and closing of stomata serve, in some measure, to regulate transpiration. We shall not, however, be justified in concluding that this is the primary significance of these movements. For, as we have suggested already, the stomata are not merely concerned with

transpiration, but also with exchange of gases between plant and air, and it may well be that the opening stomata in the light and their closing in the dark are concerned primarily with the regulation of gaseous change, and only indirectly with transpiration.

In order to investigate the effect of external conditions, light, darkness, etc., on the rate of transpiration, we require a more delicate apparatus than that used in Exp. 134 or Exp. 137, though for certain experiments (Exp. 153 and 154) either of those may be employed. The principle on which our more delicate method is based is as follows:—

151. The rate of transpiration by an herbaceous plant is about equal to the rate of absorption (p. 145). Since water absorbed passes from the root to the leaves, it must pass through the stem. Therefore, if we connect the stem of a living plant with a narrow tube, we may be able to measure the water passing up the tube and to measure the rate at which it passes. Such an apparatus, called a potometer, may be of one of several forms. That shown in Fig. 24 and designed by Professor Farmer, consists of the following parts:—A wide-mouthed bottle which is fitted with a rubber cork pierced with three holes, the middle hole being just large enough to admit of the passage of the stem of a fan-sized leafy branch. A thistle funnel provided with a stop-cock is fitted into one of the other holes, and in the third, a bent tube of narrow bore is passed, so that its end is flush with the inner surface of the cork.

Another type (Fig. 25)—the original form devised by Francis Darwin—consists of a glass tube with a bulb at one end, to the free end of which a short piece of rubber tubing is attached securely by wire. The stem of a branch of the plant to be used in the experiment is passed through the rubber tube into the bulb of the vessel and fixed firmly by wire wound round the tubing. When in use, the upper end of the straight part of the tube is closed by a well-fitting cork, and the lower end is closed with a rubber cork with one hole, through which a glass capillary tube of about five or six inches in length is inserted. In setting up either apparatus, the following precautions are to be observed:—

(1) Cut off the branch under water, e.g. by bending it down beneath the water contained in a basin or pail

(2) Keep it for some hours with its cut end under water, e.g. if a laurel branch is used, cut it off in the evening and use it the following morning

(3) Before it is put into the potometer, remove a short length ($\frac{1}{4}$ inch) of the cut end by means of a sharp razor (making the cut under water)

(4) When inserting the branch, fill the potometer with water and submerge it in a sink or pail filled with water

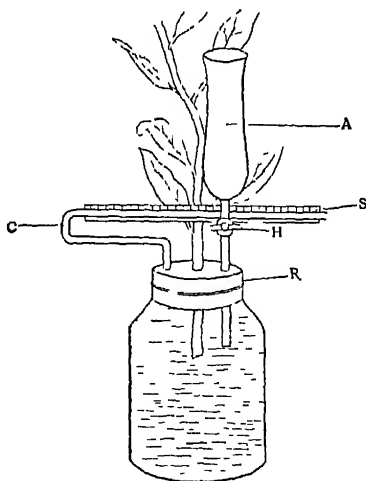


FIG. 21.—POTOMETER (AFTER FARMER)

A, funnel with stop-cock (H), C, glass tube of narrow bore, S, scale, R, rubber cork

(5) Fill a tumbler with water, insert the cut end of the branch in the tumbler, carry it to the sink, submerge the tumbler and transfer the branch to the potometer. If the bottle-potometer is used, submerge the rubber cork and pass the cut end of the branch through the middle hole so that, when the cork is inserted in the neck, the branch projects into the bottle. Press the cork into the bottle, close the stop-cock, pour water into the thistle funnel, and

lift the apparatus out of the water. See that fit between cork and branch is good; if not, dry it and make the joints air-tight with wax (Apt). Place the apparatus in a good light and fix a scale on the narrow tube, or make on it india-ink marks at a distance of three or more inches from one another.

If the tube-potometer is used, submerge it, and the cut end of the branch from contact with air. Push the branch through the rubber tubing, wire it by 10 turns of thin wire passed round the rubber tubing. When the joint has been made good, push in the cork, the capillary tube, cork the upper end of the limb, lift the apparatus out of the water, and fix it in the clamp of a retort stand. Adjust the apparatus so that the free end of the capillary tube just dips in a glass vessel containing water and standing on a block of wood. Make two india-ink marks a few inches apart on the capillary tube.

One advantage of the bottle apparatus is that air may be prevented from accumulating in the limb; for when air has passed some distance in the narrow tube it may be driven back by turning the limb and allowing water to enter the vessel from a funnel. When not in use, attach by means of rubber tubing, a bent glass tube to the free end of the narrow tube, and allow its end to dip in water to prevent the entrance of air.

To use the tube apparatus, slip away the wax, remove, by means of blotting-paper, a drop of water from the free end of the capillary tube so as to admit a bubble of air, replace the vessel and block so that the bubble enters. Time, by means of a stop-watch, the time for the air bubble from the lower to the upper mark. Admit another air bubble, repeat the time, and so on. If the branch is losing water rapidly the passage of the bubbles is very fast, and a few minutes suffice to give 10 or 12 readings.

When the bottle-potometer is used, the end of the air column in the narrow tube is followed and the time taken to pass from mark to mark recorded. It is back by opening the stop-cock and admitting

ottle. Thus a series of readings may be obtained. An average of the series gives the rate at which water is passing

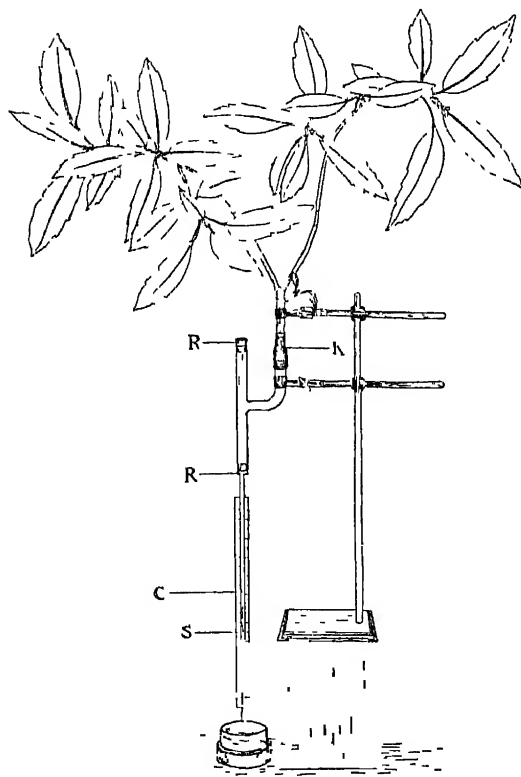


FIG. 25 — POTOMETER (See p. 156)

C, capillary tube, S, scale, R R, rubber corks, K, rubber tubing

the cut end of the stem, and since, as we have shown, the rate is, generally, proportional to the rate of transpiration, our potometer gives us a measure—albeit an indirect—of the rate of loss of water by the leaves. The chief

value of the apparatus, however, lies in its quick comparative records. In such comparisons assume that, if under one set of circumstances, pass, on the average, twice as quickly as under the rate of transpiration is twice as great in the latter circumstances. The records of transpiration now to be taken should be also temperature-records.

152 Expose the apparatus in the window, record the air temperature, take a series of readings to determine the average rate at which water is passing through the graduated tube. Transfer the apparatus to the open air and set it in a breezy, shady place, and take a series of readings. Expose it in the open sunlight, and, after an interval, make another series of readings. Replace it in its original position, and, after a quarter of an hour, take readings. Put it in a dark room, and leave it for about half an hour, and make a series of readings. The plant may be screened from the light (e.g., by a glass plate) used in reading the rate of passage of the bubble.

153 Put a small leafy branch in the tube. Having taken a series of readings in ordinary air, put the branch in a bell-jar, provided, if not old enough, with a "founce" of glazed calico. Make the jar thoroughly moist, e.g., by syringing water under the jar (avoiding wetting the leaves). Record the rate of transpiration in the bell-jar, and after an interval take a series of readings in ordinary air.

154 Having dried the bell-jar, replace it over the plant and put under the jar a large dish containing calcium chloride. Take readings, one series at once, at intervals of about half an hour.

The conclusions which we draw from the foregoing experiments are as follows. In the first place, the rate of transpiration is greater in the open air than in the dark room (Exps. 152 and 151). In the second place, it decreases in darkness and increases in light, a fact that the temperature may be considerably lower in the dark room than in the open air is in part due to the lower rate of transpiration in the former.

Bearing in mind that the stomata of many plants

in darkness and open in light, that they may close in dry air or when the plant is dry at the root, and that they open under circumstances in which evaporation is reduced, we may conclude that, though transpiration is set up as the result of evaporation of water-vapour from the air-spaces surrounding the cells of the leaf, though it is hastened by conditions which favour evaporation, and checked by conditions which hinder evaporation, the process of transpiration is not a mere evaporation. For, on the one hand, the living cells resist desiccation automatically, as they become drier, so the contents of their sap concentrate, the osmotic pressure increases, and water is held more strongly. On the other hand, by the opening and closing of its stomata, the plant regulates to some extent the rate of loss of water. As the guard-cells lose water, they decrease in size, change in shape, and come closer together, so that the stoma between them is made smaller. If water is supplied plentifully, *e.g.* by wetting a leaf, the guard-cells become turgid, the stomata open, and the rate of transpiration tends to increase. Whether the prime purpose of the movement of stomata is to secure this end, or whether this is a secondary matter, we cannot decide. It may be that, as the plant loses water, it becomes important for it to shut down *all* its activities, and that the closure of the stomata serves to secure this end by reducing the rate of gaseous exchange between the plant and the air.

CHAPTER X

THE passage of water from root to leaves the channels follow the transpiration current water conducting wood and skeletal w
The causes of the ascent of water Phenomena connected with absorption of water root pressure bleeding excretion of w
water-pores (hydathodes)

Our previous studies have taught us that of the w absorbed by the root hair-cells, some passes to neighboring living cells and, absorbed osmotically by them, contributes materially to the increase of weight manifested in the growing plant. The water used in this way is, however, but a small part of that taken up by the root. For the larger part of the water absorbed travels, as we know, from the root to the leaves, whence it passes into the air in the form of water-vapour. The stream of water which passes through the plant we may call the transpiration current, and we require to learn what are the channels along which this current travels in its passage from the root to the leaves. That the transpiration current passes along definite channels we demonstrate in the following way —

155 Dig up one or two well-grown seedling bean plants, wash their root-systems free from earth, and across the main roots, place the plants with their roots in water in which is dissolved some dye, *e.g.* eosin or ink, and leave them exposed to the light till the colour of the dye is visible in the "veins" of the leaves; and also, if the plants are in bloom, in those of the flower. Cut the root and shoot of one specimen transversely into a series of pieces, examine the cut surfaces by means of a lens, and determine that the dye has travelled

the stem along definite tracts. Note that the position of the stained areas is not the same in the root as in the shoot.

156 * Mount transverse sections of the stem in water or dilute glycerine, and examine them microscopically. Note that the tracts along which the dye has passed contain what appear to be largish spaces, each surrounded by a fairly thick and stained wall. Observe that the dye which stains the walls of the relatively wide water-channels is absent from the cellular tissues of the stem. By means of similar cross-sections, follow the course of the dye from the shoot, through the leaf-stalks to the "veins" of the leaf.

157 Cut the root and shoot of the second plant lengthwise, proceeding cautiously from below upward, and trace the longitudinal course of the dye. By the aid of the microscope, observe, in thin longitudinal sections, that, whereas the tissues of the part outside the stained areas are made up of minute cells, those in the stained areas contain long and wide tubes, the walls of which are curiously thickened in an annular, spiral or other regular manner. These dead elements are, as the dye indicates, the channels (vessels) along which the water passes to the leaves. On tracing them towards the leaves, the vessels are found to connect with similar, smaller groups which run in definite courses through the leaf-stalks, and are distributed in the "veins" throughout the leaf. Some idea of the length of individual vessels may be obtained in the following way —

158 To ordinary leaf-gelatine add so much water that it remains liquid at about 35°C , but sets to a solid mass at room temperature. Colour the fluid gelatine with eosin, and pour it into a tall jar, e.g. a measuring vessel, stand the latter in a large beaker or saucepan containing water which is kept at a temperature of about 40°C by means of a bunsen flame. Plunge the cut end of a leafy branch below the surface of the gelatine, and leave it for an hour. Remove the vessel containing the branch from the beaker of hot water, plunge it under a tap, and allow a stream of cold water to fall on it so as to cause the gelatine to set. When the branch is cooled, take it out of the gela-

time, cut it longitudinally, and determine the length of the stem along which the gelatine may be traced. Inasmuch as the gelatine is free to pass into the open ends of the vessels exposed on the cut surface, but cannot pass across the end walls of the vessels, we obtain a rough measure of the length of the vessels in the plant on which the experiment was made.

By more accurate methods it has been demonstrated that the length of vessels varies very considerably in different plants. Thus, in the oak, they have been estimated to be 2 metres or more long, in species of fig (*Ficus*) they ran from 10 to 66 centimetres, whereas the much shorter conducting elements (tracheids) of the Scotch Fir (*Pinus sylvestris*) are only about 4 mm in length.

159. Whilst the previous experiment is going on, cut off two similar branches, *e.g.* of laurel, stand one with its cut end in water, and plunge the cut end of the other beneath the surface of melted coco-butter. After about a quarter of an hour, remove the branch, cool its cut end, and make a fresh cut about a quarter of an inch from the exposed surface, stand the branch with its cut end in water in bright light, and observe that, owing to the vessel having been blocked by the coco-butter, the leaves wither far sooner than those of the branch which was placed at once in water.

The conclusion that water passes through the *cavities* of the vessels may be verified thus —

160. Set up a well-grown branch of sunflower (*Helianthus annuus*) or Jerusalem artichoke (*H. tuberosus*) in a potometer (Exp. 151), and determine its rate of transpiration. Fix the middle of the branch in a vice and secure the latter so tightly as to compress the stem and hence its vessels. Observe that the rate of transpiration falls off very markedly. Release the stem from the vice and ascertain that the rate of transpiration increases.

The microscopic examination of the cross-section of the bean stem of Exp. 155 showed us that the vessels are distributed in isolated groups.

161. That this is not their distribution in older stems we recognise by observing with the aid of a lens the cut ends of two-, three- and more year-old branches of woody plants, *e.g.* trees such as lime, lilac, etc. In these plants

the wood forms the greater part of the stem, and the vessels, which make up the larger part of the wood, are seen to be distributed uniformly throughout the woody tissues.

The process by which the wood is increased from isolated strands to a solid cylinder may be studied in text-books of plant-anatomy (Bibliography, 5, 6), but, by comparing the cut surfaces of one-, two-, and three-year-old branches with one another, we observe that each year a new ring of wood is formed on the outer side of the old ring. We note further, that there are fairly sharp lines of demarcation—the annual rings—between each year's wood and that of preceding and successive years. With a good lens, or by microscopic examination of sections, observe that the appearance of these concentric annual rings is due to the fact that the vessels which are the last to be formed in late summer of one year are thicker-walled and have smaller cavities than those which are the first to be formed in the spring of the ensuing year. We conclude that in our native perennial plants, which possess well-marked annual rings, it is possible to determine the age of the tree by inspection of the surface exposed by a cross cut. Though we cannot now study the question as to the cause of these marked differences between the wood formed in spring and that formed in late summer, we may remark that it is only in regions like our own, in which vegetation is subject to a distinct seasonal check, that the wood shows well-marked annual rings.

Our main object in referring to this matter is to ask ourselves the question, whether all the vessels of the wood are capable of conducting water, or whether this task is discharged by certain vessels only? The fact that there are to be seen hollow trees, such as willows, which have lost much of their older wood—that nearer the centre of the trunk—seems to indicate, that at all events in such plants, the young wood suffices for water-conduction. The first step towards an answer to this question we make as follows—

162 Obtain four, similar, leafy branches of some tree, *e.g.* oak, elm, etc., and stand them with their cut ends in water coloured with cosin or red ink. Having formed a clear idea of the extent of the wood by inspection of the cut

surfaces, remove a ring of bark from the stem of a specimen (No. 1), avoiding cutting into the young wood. Slip a rubber band around the stem of No. 2, and, using the edge of the rubber band as a guide, make a circular cut with a sharp knife into the bark and young wood. In No. 3 make a similar circular cut, but so deep that only the oldest, central heart-wood remains intact. Make no incision into No. 4. Ascertain, by the potometer method (Exp. 151) or by observing the rates of withering of the leaves, what parts of the wood are concerned specially with water-conduction.

By means of such experiments on different plants we demonstrate (1) that the parts of the stem external to the wood are not concerned with the transmission of the water current along the stem, (2) that the young wood is specially concerned with water-conduction, (3) that the older wood of such trees as oak is incapable of conducting water, though in others, e.g. lime, it aids the young wood in this work. In those trees, the old wood of which has ceased to serve the purpose of water-conduction, it may be noted that marked differences of colour, etc., exist between old and young wood, in these cases, the difference between young sapwood (alburnum) and older heart-wood (duramen) is made visible. It will be evident that though the heart-wood of a given tree has ceased to conduct water, it nevertheless may play an important part in acting as an internal skeleton to the trunk, and thus giving it the power of resisting the shearing action of wind and providing support for the ever-increasing crown of branches which the tree puts forth. Though we are not now making a thorough study of the structure of wood, we must satisfy ourselves that the woody tissue does not consist exclusively of vessels, but contains other elements, some of which may be shown by appropriate tests (Bibliography), to be unlike the wood, not only in size, but also in the fact that they contain living protoplasm and also reserve substances, such as starch.

We are now able to picture to ourselves the water-conducting system of the plant. Below, in the soil, are the root-hairs, the organs for the absorption of water. The inner walls of the root-hair cells adjoin those of the

cortical cells which absorb, according to their osmotic capacity, water from the root-hair cells and become turgid. The innermost cortical cells are in contact with others of the deeper-living tissues, and these, in turn, abut on the vessels. In its young stage, a vessel, which is formed from a longitudinal row of cells, is a living element, and, like all living cells, contains osmotic substances. Hence water passes by osmosis into the developing vessel. The constituent cells which are forming the vessel increase in length, their cross walls disappear, their longitudinal walls thicken, become lignified (woody), and are rendered rigid by the bars or rings or spiral bands of woody substance formed, as the last act of the protoplasm, on their inner surfaces. Thus, the completely developed vessel is full of water. Each such vessel of the root connects above with a similar older vessel, the cavities of the two vessels being separated only by a thin membrane which constitutes the end walls of the two vessels. Across this membrane, water and dissolved substances pass readily. If we think of the plant as it grows from the seedling stage, we can picture new vessels filled with water being added in root and stem by the development of rows of cells some distance behind the growing points, and hence we can imagine continuous columns of water in each of the longitudinal series of vessels which extend from the veins of the leaf, through the stem, to the root. Now, it has been shown that a continuous column of water, such as we have imagined to exist in the longitudinal series of vessels in the plant, has very remarkable properties. Even though it is subjected to a considerable pull, the column does not break. We know that, in the plant, water is evaporating from the inter-cellular spaces of the leaves into the air, and that this process sets up a chain of events, resulting in the withdrawal of water from the vacuoles of the green cells. Hence the osmotic pressure of these cells is increased, and they withdraw water from the colourless cells with which they are in contact. The colourless cells, abutting on the vascular elements in the veins of the leaf, absorb water from these elements. But as we have pictured the conditions, these vascular elements contain the tops of continuous water-columns, which extend through the stem to the

root. As water is lost above, one of two things must happen, either the water-column must break, or the column as a whole must be hauled up. It can be shown experimentally that such a column does not break, even when subjected to a very considerable pull. Hence we may infer that it is hauled up bodily. Thus it may be in this manner that the ascent of water is effected even in great trees, such as the Eucalyptus of Australia, or the Sequoia of Western America, which reach a height of about 300 feet.

It should be noted that we do not offer proof that this is actually the mode by which the ascent of water is effected, we may, however, adopt it as an hypothesis to account for the fact—which is one of the most remarkable in the whole range of plant-physiology—that a tree is able to haul up large quantities of water to a very great height. Although it must not be regarded as a model illustrating the mechanism of the lift of water which takes place in transpiration, it is none the less instructive to make an apparatus which shows that, when water is evaporating from a surface connected with a water-column, a considerable “pull” is exerted. For this purpose we proceed as follows —

163 Soak a piece of parchment membrane in water and tie it tightly over the wide end of a thistle funnel. Fill the funnel *completely* with boiled water, and, closing the narrow end with the finger, stand it with the wide end uppermost in a dish containing mercury. Place a scale behind the apparatus and record from day to day the height to which the mercury is lifted in the tube as a result of evaporation of water from the surface of the parchment.

164 A better apparatus, but one that requires a little skill to construct, is made by causing plaster of Paris to set so as to form a layer closing the mouth of the funnel. The apparatus is filled with water, and in this, as in the previous experiment, it is important that the water used should be rendered by boiling as free from air as possible, since the air contained in unboiled water, being liberated as evaporation proceeds, accumulates beneath the surface of the plaster and tends to break the continuity between

the water in the plaster and that in the tube. That the transpiring plant exerts a similar pull may be demonstrated in a similar way as follows —

165 Cut a branch from an actively-transpiring plant and stand it for some hours in water. Fit it, by means of a split cork, into a vertical tube filled with water. This may be done by plunging the tube and the cut end of the branch in a pail of water and placing the cork in position. Raise the tube, fix it in the clamp of a retort-stand, wipe the cork dry, and make the joints air-tight by means of a layer of wax. Whilst the lower end of the tube is closed with the finger place the apparatus so that it just dips into a dish containing mercury. The rise of mercury in the tube is then recorded as in the previous experiment.

Another experiment may now be performed to demonstrate the fact that the activity of the roots in absorbing water may, especially at certain seasons of the year, set up a considerable pressure in the plant, which is described as *root-pressure*. Not all plants exhibit this phenomenon, and hence root-pressure is not to be regarded as an agent effecting the ascent of water. The experiment may be tried on any convenient vigorous plant. *Sparmannia africana*, or species of *Phaseolus* or *Fuchsia* are suitable for the purpose.

166 Give the pot-plant selected for the experiment a thorough watering. Cut back the stem to within a few inches of the soil. Slip a piece of rubber tubing over the cut end and wire it securely, pour in a little water and fix, by means of wire, a long glass tube of fairly narrow bore in the other end of the rubber tubing. Place a scale behind the tube and record at intervals the height of the water in the tube. In the case of trees and shrubs in the open, experiments to demonstrate root-pressure succeed best in spring before the leaves are fully expanded. The phenomenon of root-pressure may be connected with others occurring in nature. For instance, under certain circumstances, vines, especially when pruned late and with the wounds imperfectly healed, exhibit a "bleeding" process. In the spring, sap containing various substances in solution exudes from the cut ends of the shoots.

The quantity of fluid which may escape is often considerable, e.g. from birch, sugar palms, etc. The liquid which escapes from the flower-stalks of certain palms is so rich in sugar used in Ceylon and Java for the manufacture of liquors. In some cases, this exudation or 'blee' be due to root-pressure, but, in others, it is the pressure set up locally, often as the result of an the stem or other parts. Thus, in Java, the having beaten the inflorescences of palms with hammers, cut off the flower-stalks, and fix vessels the liquid which exudes from the wounds.

Connected with phenomena of bleeding exhibited by certain plants, e.g. oat, lucerne, etc., which show frequently in the morning of water, often mistaken for dew, glistening on or surfaces of the leaves (p. 2, Chapter I) tropical plants, e.g. species of *Colocasia*, this phenomenon of *guttation* is so marked that at intervals of a few drops of water drip like rain from the tips of leaves. In these cases, it is found that the leaves possess organs, often formed by the modification of stomata as hydathodes¹. In various plants, notably in savillage, the water exuding from the hydathodes contains carbonate of lime in solution, which, on the evaporation of the water, is deposited as a white fringe on the edges of the leaves. Certain water-plants, e.g. *Ranunculus*, *Callitriche*, etc., possess very conspicuous hydathodes at the tips of the leaves, which either open to the exterior or are covered over only by the cuticle. It is probable that, in water-plants, these organs serve the purpose of increasing the rate at which water is discharged from the leaves, and so of augmenting the supply of

¹ The phenomenon of "guttation" is most easily observed in plants growing under conditions which, whilst favouring absorption at the roots, hinder rapid transpiration, e.g. in the early morning in a moist greenhouse, or, if the plant is kept under a bell glass, in the dark. The excretion of liquid water may be observed in any of the following plants: — *Primula sinensis*, *Tropaeolum majus*, *Aconitum napellus* sp., *Fuchsia* sp., *Helicoborus niger*, *Vicia sepium*, *Thymus*, *Avena sativa*, *Zea Mays*. (See list of plants, Appendix.)

mineral salts. Owing to the fact that many of these plants have submerged leaves, the transpiration of water-vapour does not occur, indeed, in some cases, stomata are not present, and hence the movement through the plant of water containing dissolved salts is likely to be very slow. The explanation of the significance of hydathodes in the case of terrestrial plants may be as follows.—During the night, the roots continue to absorb water, but transpiration, as we have already indicated, is considerably reduced. The water-courses, and the tissues of the plant generally, become filled with water, and the increasing pressure is relieved by the exit of water through the “water-pores” or hydathodes, which may thus be likened to safety-valves.

In concluding our study of the relation of plants to water, we note that, beside effecting the transport of dissolved mineral substances to the leaves, transpiration serves also to prevent the undue heating of leaves exposed to bright sunshine. Just as perspiration, evaporating from our bodies, lowers the surface-temperature, so transpiration lowers that of the leaf. Finally, the fact that, during bleeding, sugars and other organic substances may be exuded from the vessels, may be taken to indicate that, not only are the vessels the conduits along which water and salts (crude sap) are transported, but also that they may serve, upon occasion, for the transport of “elaborated sap,” that is, of plastic food-materials. In spring, at all events, when the unfolding leaves are making a heavy demand on other parts of the plant for food-substances, it appears that the demand is met by the discharge of sugar and other plastic food-substances in solution into the vessels through which these substances are transported rapidly, in the transpiration-stream, to the leaves.

Before leaving this subject, we demonstrate that transpiration—that is, the regulated process of evaporation of water—is not confined to the leaves. As the results of Exp. 138 indicate, the shoot-axis may also transpire.

167* Microscopic examination of sections across a young shoot shows that stomata are present in the epidermis, and we demonstrate, by means of cobalt chloride paper applied to a portion of a young stem, that the

stomata serve as channels for the escape of vapour.

168 In older stems, e.g. two- or more-branched trees or woody shrubs (elder), stomata are replaced by less regular openings, lenticels, which, as we may show by the cobalt method, serve for the exit of water-vapour. The in the elder, etc., may be recognised by the naked small elongated areas, differing in colour from the surface of the stem. That they serve as channels of communication between the external air and the intercellular spaces of the plant we demonstrate.

169 In summer, cut off a two- or three-year-old elder or horse-chestnut. Remove from it a piece of leaves and several inches in length. Attach to a short length of rubber tubing, turn the tubing and bind it tightly to that on the stem so as to be completely. To the other end of the piece of stem a thick india-rubber tube (pressure tubing) also long. Insert into the free end of the pressure stout glass tube several feet in length. Make good, lay the stem in a vessel of water, add water to the glass tube, and, by squeezing the rubber, drive air contained in the pressure tubing. Clamp the tube vertically in a retort stand. Pour mercury into the glass tube by means of a funnel. Note that, as the height of the mercury column increases, streams of water escape from the lenticels. Compare with Exp. 168 and apply a similar explanation.

170 * Microscopic examination of a cross-section of the superficial tissues of the old stem of elder, and the explanation of the behaviour of the leaf allowing the escape of gases. Whereas the openings of the shoot are made up generally of brick-shaped cells fitted closely together, the lenticels consist of four cells between which are definite intercellular spaces. The brick-shaped cells are empty, dead, and have corky walls. A section through a small piece of cork shows similar layers of brick-shaped cells. Cork is used domestically to prevent the escape of water and gases. Man, in fact, puts cork to a use

to that to which it is put by the plant. The layers of cork which occur in the older shoots of woody plants serve to prevent loss of water by evaporation from the thin-walled, delicate, underlying tissues. But a continuous jacket of cork about a branch would not only mean prevention or reduction of loss of water by evaporation, it would also mean prevention or reduction of all gaseous exchange. The tissues of the stem would be deprived of oxygen, and, consequently, their growth and activity would be checked. The lenticels, like the stomatal openings of the leaf and of the young stem, serve as aerating channels, and hence also as channels for the escape of water-vapour. We thus discover that the plant, by developing a "water-proof" layer of cuticle in the outer walls of its epidermal cells in the young parts (leaves, young shoot), and by forming sheets of cork in or beneath the epidermis of its older stems, contrives to meet two needs, (1) the prevention or reduction of indiscriminate evaporation, and hence the regulation of transpiration, and (2) the provision of aerating channels whereby gaseous exchange between the air and the tissues of the plant is maintained. If oxygen is being used by any of these tissues, it is derived from that in the air of the intercellular spaces. The oxygen pressure of that air falls. When it falls below that of the external air, diffusion of oxygen molecules from the external air to that of the intercellular spaces is set up, and thus the cells abutting on the intercellular spaces receive a constant supply of oxygen. Similarly, any gases produced by the cells of the tissues of the plant may escape by diffusion into the outer air.

We have a means of verifying the accuracy of the twofold significance of lenticels in relation with transpiration and with gaseous exchange. In winter, the plant is, compared with its summer state, at rest. The leaves fall, and the buds remain till spring enclosed within hard scale-leaves. In such circumstances, the plant has need for but little oxygen, and hence is under no necessity of keeping its aerating channels open. Moreover, since, in winter, the soil-temperature falls so low that the root-system is able to absorb but little water, the existence of open passages through which water-vapour may readily escape would be a source

of danger to the plant, leading to its death by frost. Therefore, if our interpretation of the twofold significance of stomata and lenticels is correct, we should expect to find that the plant takes steps to put them out of action during winter. With respect to the leaves, we know in most plants this is so—they are, in deciduous trees, out of action altogether by being cast off from the branches. With respect to herbaceous perennials, the prevention of loss of water is effected by the most drastic measure: the whole plant dies down to the level of the ground in winter, and grows underground.

171 That the lenticels also are put out of action in winter we demonstrate by repeating, at that time, Exp. 169. We find then that, in spite of considerable pressure, no air escapes from the lenticels. They are

172 * By means of sections we prove that, when a plant is preparing for winter, it produces, in its older parts, new sheets of cork below those already in existence. Instead of forming lenticels of loose masses of cork cells in spring, it produces a continuous cork layer beneath the lenticels as well as elsewhere. Thus transpiration is slowed down and gaseous exchange is reduced.

It is indeed remarkable that the fall of the leaves is engineered in many plants in precisely the same way. Thus, in the horse-chestnut in late autumn, cells at the base of the leaf-stalk begin to divide, forming a layer of brick-shaped cells. These cells divide to form tiers of cells, which become corky. The cork cells form a layer around the base of the leaf-stalk; moreover the water-channels running through it to the blade become blocked. The leaf, thus cut off from fresh supplies of water, gradually withers and, at a breath of wind, comes rustling to the ground. This is facilitated by the formation of an "abscission zone" on the leaf-stalk side of the corky band. It consists of a few layers of delicate cells formed by rapid division. A row of cells shortly before the leaf falls. This layer serves the purpose of healing the wound caused by the venting loss of water, ingress of spores of fungi, etc. Each leaf that falls is cut off by a surgical operation performed by the plant, and

wound produced is healed in advance. Similarly, as a consequence of the formation of cork in the stem, all the tissues external to the cork are cut off from supplies of water, and hence dry, shrivel and crack. Trees which produce cork in the surface layers of their stems have thin, smooth bark, those which produce it in their deeper layers have thick, corrugated bark. According to the exact mode of development of the cork, the bark is seen to be in sheets as in birch and beech, in lozenge-shaped masses as in elm and oak, and so on.

When growth resumes in spring, not only are new leaves formed, but, in many trees, by the growth of the deeper tissues of the stem, the bark is split and torn, and may be cast off. Many plants utilise the tissues which constitute the bark for excretory purposes, that is, for getting rid of waste-substances. Thus it is that many and varied kinds of materials, some of which are of great value in commerce and medicine, are obtained from the bark of trees (quinine, tannin, hazeline, etc.).

Using the nomenclature of p. 149, we may say that, in winter, a deciduous tree, leafless and girt in cork, is a xerophyte, in summer, crowned with delicate leaves and with open lenticels, it is a hygrophyte. Such plants, which change periodically from the xerophytic to the hygrophytic state are called tropophytes, and to the fact that, as soil-temperature falls, absorption of water decreases below the amount necessary to make good the loss due to transpiration is to be ascribed the wonderful series of changes which mark the preparation of the plant for passing the winter. The subject is an alluring one, but we cannot follow it further now.

173 The changes of some common tree should be followed throughout the year and records of the dates of bursting into leaf, of closing of the lenticels, of cork formation, etc., should be made, and the records compared with meteorological records, particularly with those of soil temperatures.

CHAPTER XI

THE origin of the carbon compounds contained in plant materials from which the plant constructs these compounds played by chlorophyll groups (chloroplasts) in the process, the energy by which the process is carried on, of carbohydrates, from the leaves to other parts of the synthesis of organic nitrogen compounds by the plant.

Before we embark on the last stage of our study of the mode of nutrition of plants, we will review the conclusions to which our experiments have led.

Sand- and water-cultures have served to demonstrate that certain mineral substances are essential to plants, also, that all the elements required by the plant for nutrition are, with the exception of carbon, derived from the soil in the form of water and mineral salts. The water and mineral salts absorbed by the plant, a certain amount passes by osmosis from cell to cell, thus satisfies local needs, but the great bulk of the water is transpired and the mineral substance is left behind. Thus the green cells of the leaves are constant supplies of mineral substances. Since the mineral salts do not accumulate indefinitely in the leaves, since also the elements which they contain become constituents of complex organic bodies, such as proteins, it is evident that the mineral salts, in serving for nutrition, undergo chemical change. Therefore, they are to be regarded, not as food-substances, but as raw materials used by the plant in the manufacture of food-substances. Similar considerations convince us that the carbon compounds occurring in the plant are

result of processes of manufacture carried on by the plant, and that the carbon which finds a place in these organic compounds is derived from external sources.

It is our object in this chapter to discover what we can of the nature of this manufacturing process, and we commence by enquiring into the source whence is derived the raw carbon-containing material used in the manufacture of organic compounds.

The problem is simple. We know that a plant may grow and that the amount of its carbon-compounds may increase, although there may be no compounds of carbon present in the soil. Therefore, the air is the source whence the plant derives its carbon. It must, however, be admitted that this does not appear, at first sight, to be very probable. For, on the one hand, the amount of combined carbon contained in a plant—a tree, for example—is very great, on the other hand, the only carbon-containing constituent of pure air, carbon dioxide, makes up but a minute fraction (0.3%) of the total volume of the atmosphere. It is doubtless these facts which made botanists, down to comparatively recent times, slow to accept the result of the earlier experiments which pointed to the air as the source of carbon to the plant. They preferred to adhere to the old "humus theory," which taught—seemingly on good grounds—that the organic compounds of carbon in the soil are the sources whence the green plant obtains its supplies of this element. They knew that plants flourish when provided with a plentiful supply of organic carbon-compounds, when, for instance, the land is richly manured, they noticed that soils deficient in these substances support but poor crops, and it seemed to them a breach of common-sense to be asked to believe that the air, with only some three parts in 10,000 of carbon-dioxide, could provide the large quantities of carbon which are contained in plants. A more thorough consideration of the facts shows us that, though the *percentage* of carbon dioxide in the atmosphere is but small, the actual quantity contained in that vast ocean of air is enormous. Moreover, this quantity is augmented constantly. Every fire that burns, every plant or animal that respire, and every organic substance which undergoes decay,

produce carbon dioxide, and discharge it into the atmosphere.

But, after all, experiment is better than argument, and we proceed, therefore, to put the matter to experimental proof.

If the carbon dioxide contained in the air is the raw material used by the plant in the manufacture of organic compounds, then we shall expect to find that an adult plant supplied with air, from which the carbon dioxide has been removed, ceases to manufacture organic carbon compounds. Since, as our studies of germination have shown, organic carbon-compounds, such as sugar, serve for the nutrition of the plant, it follows that, if there are no stores of organic carbon-compounds present by preventing the plant from manufacturing these compounds, we reduce it to a state of starvation. To carry out the experiment, we proceed as follows:—

174 Take two approximately equal, actively-growing young sunflower or dwarf bean plants raised in small pots. Label them A and B, cut off a piece about half an inch square of one of the older leaves of each plant. Having watered the plants, place them in the dark. At daily intervals, cut off similar samples. Keep the samples separate, and immediately after they are cut off, plunge them for a minute into boiling water to kill them. Place the pieces in wide specimen tubes containing methylated spirit. Label each tube with the letter A or B and the date on which the piece of leaf was cut off. Cork the tubes and expose them to bright sunshine. After a piece has been in a tube for 24-48 hours, take it out, observe that it is brittle and colourless, and rinse it with water. Place each sample separately in a white saucer, and pour over it a solution of potassium iodide-iodine, to which has been added a strong solution of chloral hydrate (Appendix A). Observe that, as indicated by the iodine reaction, when the plants are maintained in the dark, the starch which is contained in their leaves gradually decreases in amount until, after one or more days, it disappears altogether. When this has happened, the plants are ready for use. They should, however, be kept in darkness till the apparatus now to be described has been prepared.

175 Having obtained two large bell-jars—(the cloche

(Appendix A) used in "French Gardening" sieve admirably), take one of them to a carpenter and get him to make the following—A circular, wooden base-board about an inch thick and of a diameter about two inches larger than that of the rim of the bell-jar. The base-board must be of one piece, and should be varnished. Three, 3-inch cubes of wood to serve to raise the base-board above the level of the table or ground. Instruct the carpenter to cut on the base-board a circular groove, about $\frac{3}{4}$ inch in depth and $\frac{1}{2}$ inch wide, of such diameter that the rim of the bell-jar fits comfortably into it, and to bore a round hole 1-1 $\frac{1}{2}$ inches diameter, at a distance from the groove about equal to one-third of the diameter of the circle described by it. The edge should be circular and smooth since a rubber cork is to be inserted in the hole.

Bring the apparatus into the laboratory, and select a glass tube of such a size and shape that, when it is in position under the bell-jar, its wide end is an inch or more from the top of the latter, and its narrow end projects below the bottom of the base-board, but clears the table by about 2 inches. A cylindrical separating funnel with a narrow stem serves the purpose (Appendix A). Push a small plug of cotton wool to the bottom of the wide end of the tube, and fill the latter with soda-lime—a substance which absorbs carbon dioxide. Insert the stem of the absorption tube into a rubber cork which is large enough to fit tightly into the hole of the base-board. Fix the rubber cork in position, make a good joint, attach to the end of the stem which projects beneath the board a piece of rubber tubing, and close the latter with a clamp. Set up the board in a well-lit place—preferably in a sheltered position in the open air but screened from direct sunlight. Bring quickly one of the plants (A) from the dark room. Make *rapid* measurements of the height of its stem and the size of its smallest leaves (this need only take a minute or so). Stand the plant on the board, place the bell-jar in position, and cover the bell-jar with a double fold of black Italian cloth in order to protect the plant temporarily from the light. Having melted enough wax-mixture (Appendix A) for the purpose, pour it into the groove so as to seal the bell-jar (or soft putty may be used). Bring the second plant from the dark, measure

its stem and youngest leaves. Cover it loosely bell-jar. This second apparatus serves as a control. Remove the black cloth from the first bell-jar, and connect the rubber tube so that air may now enter the bell-jar through the absorption tube. We thus have the two plants under fairly similar conditions, except that one is deprived of carbon dioxide, the other ordinary air. The plant which is exposed to ordinary air will show a definite increase both in height of stem and in the younger leaves, estimate and record the increase, noting also the number and size of any leaves it has unfolded since it was put under the bell-jar. The wax from the edge of the other bell-jar, take the control plant, cut off one of its older leaves, and, having left it in a dark place, determine that the plant has but little or not at all, and that it has developed new leaves. Make these observations as quickly as possible. As soon as they are made, cut off one of the leaves from the control plant. Before examining the cut place the plants in the dark. Make a deep incision on one of the two isolated leaves in order that it may be distinguished from the other, and test the leaf for starch by the iodine method. Observe that the leaf taken from the plant which was exposed to ordinary air gives a well-marked, blue starch reaction, and the leaf of the plant which was deprived of carbon dioxide gives no starch reaction. Bring the plants from the room into bright sunlight. At intervals of five minutes take samples of the leaves of the plant which was under the bell-jar, and, by means of the iodine test, demonstrate within ten minutes or so of the time at which plants exposed to the light, starch makes its appearance in the leaves, and that the amount of starch—as judged by the depth of blue colour—increases as time goes on. Furthermore, that the plant, the development of which was checked completely when it was in air devoid of carbon dioxide, may begin again, in the course of a day or two, to increase in size, and to unfold new leaves.

We learn from this series of experiments that

(1) The leaves of a green plant, grown under ordinary conditions, contain starch.

(2) Starch disappears from the leaves of a plant kept in darkness

(3) Starch reappears very quickly and in increasing quantities when the plant is brought again into the light and exposed to normal conditions

(4) If a plant is deprived of carbon dioxide, no starch appears in the leaves, even though it is exposed to a good light

(5) In the absence of carbon dioxide, the growth and development of the plant are checked

From the results of our experiments on the germination of seedlings, we know that reserve starch stored in the seed serves to provide the seedling with carbohydrate food-material. Hence it is probable that the starch which makes its appearance in the leaves of plants grown in the light in ordinary air serves also to provide for the nutrition of the plant. Since, moreover, starch does not appear in the leaves of plants grown in an deprived of carbon-dioxide, it is probable also that carbon-dioxide serves as the raw material from which carbohydrate, which appears in the leaf as starch, is manufactured. If this is so, the leaves—and other green parts of a plant—are the seat of a manufacturing process which is of fundamental importance not only to the plant itself, but also indirectly to animals, all of which obtain their food ultimately from plants. But since the leaves of plants kept in darkness do not form starch—indeed, as we have seen, this substance actually disappears from them under these conditions—it follows that, if starch is manufactured by the leaves of plants, light plays an essential part in the process. Now, it is easy to demonstrate that the starch which appears in the leaves of plants exposed to the light is actually manufactured in the leaves, and is not carried thither in the form of sugar from other parts of the plant.

176 For this purpose, take a well-grown pot-plant (bean, clover, etc.) and keep it in the dark till its leaves—as determined by tested samples—contain no starch. Then detach one or more leaves from the plant, place their petioles in distilled water and expose the leaves to the light. After an hour or so, on applying the iodine test (cf. Exp. 174), we discover that starch has made its appear-

ance in the isolated leaves. Inasmuch as the leaves are isolated from the plant, the starch which a leaf contains cannot be derived from substances translocated from other parts. Therefore, either the starch has been manufactured from raw materials, or some substance, such as sugar, already present in the isolated leaves. Since, however, as Expt. 174 has shown, starch does not appear in leaves which, after they have been depleted of this substance by a sojourn in the dark, are brought into the light and confined in a vessel in which carbon dioxide is excluded, we conclude that starch which appears in the illuminated, isolated leaves is the result of a process of manufacture from raw material, of which carbon dioxide is one of the ingredients. That this is the only conclusion open to us is demonstrated further by showing that the dark leaves contain little or no sugar. Thus:

177 Take two lots of leaves, one from the plant of Expt. 175, and one from a similar plant which has been exposed to a good light for some hours. Cut pieces, under water, pound each lot separately in a mortar with a little water, transfer each mass to a beaker, and filter. Test the filtrate for sugar. Observe that the dark leaf filtrate gives no, or at most a slight, reaction, and that the light-exposed leaves give marked reactions. We thus learn that no starch is produced when the green parts of leaves are exposed to light, but that sugar also is formed. Thus, sugar and starch can only have been formed from raw materials, carbon dioxide, and we may speak of their formation as a synthesis of sugar, and since light is essential for their formation, describe the process as one of *photosynthesis*, *hydrocarbonate synthesis*. Inasmuch as it is by this process that plants obtain the supplies of organic carbon-compounds, we may call the process a *photo-synthesis* sometimes if we mean *assimilation of carbon*, or of carbohydrates.

The question now presents itself: Have the green parts the power of photosynthesis, or is it possessed by the whole plant? The answer we obtain from the following observations:

178 Cut transverse sections through a leaf (beech, privet, sunflower, etc.) gathered during the day-time. Mount the sections in water. Observe that the green colour of the leaf is due to the presence of vast numbers of green granules (chlorophyll grains or chloroplasts), lying in many of the cells. Note that the ordinary cells of the outer layer (epidermis) of both upper and lower sides of the leaf contain no chlorophyll grains, but that chlorophyll grains are present in the guard-cells of the stomata and in the vertical rows of cells (palisade parenchyma) beneath the upper epidermis as well as in the less regular groups of cells (spongy parenchyma) which lie beneath the palisade parenchyma and extend to the lower epidermis. Observe also the large intercellular spaces between the cells of the spongy parenchyma. Run in potassium-iodide, iodine, chloral hydrate solution, and observe that starch grains are present in the chlorophyll-containing cells (including the guard-cells of the stomata) but are absent from the colourless epidermal cells. Note the stained starch grains attached to the chloroplasts.

179 Then obtain one or two leaves of a variegated plant the foliage of which is thin and delicate (variegated maple, Abutilon, etc.). Either photograph the leaf or make an accurate drawing, *e.g.* a tracing of the outlines of its green and colourless parts. Test the leaf for starch by the iodine method and, by comparing the leaf with the original photograph or drawing, determine that the original green areas are stained blue, and hence contain starch, whereas the colourless areas of the leaf are not stained blue—that is, contain no starch. We thus learn that starch occurs only in those cells of the leaf which contain chlorophyll. Hence we infer that the chloroplasts play an essential part in the photosynthesis of carbohydrates.

But, as we learned in Exp. 16, starch may occur in colourless cells, *e.g.* in cotyledons of seeds, in cells of woody tissues, in tubers such as the potato, and so on. Moreover, since organs such as potato tubers are formed and remain underground, the starch which occurs therein does not, like that in leaves, require light for its formation. The well-known fact that green plants cannot live and in-

crease except when growing in the light, combined with results of these observations, leads us to conclude that reserve starch of underground organs and other colourless plants is not synthesised from inorganic materials in the storage organs, but is derived from the carbohydrates manufactured by the leaves and other green parts of plant. On this view, a green leaf not only manufactures carbohydrates to satisfy its own needs, but also supplies carbohydrates to the colourless cells of the plant.

This conception of the leaf as a factory of carbohydrates for the plant as a whole, helps us understand the comparative rapidity with which starch may disappear from the leaves of plants kept in the dark. We can imagine that the starch, accumulated in leaves, undergoes a change similar to that which starch, accumulated in a bean cotyledon, undergoes during the germination of the bean, namely, a conversion by agency of diastase into sugar, and just as the sugar formed in the cotyledon passes away to nourish, or accumulate in, the embryo, so that formed from the starch the green leaf travels *via* the petiole to the shoot, where it is distributed in all directions, to serve either for nutrition of growing cells or to form the reserve starch stored in the various tissues.

In order to confirm the truth of this conjecture, require, firstly, to prove that the green plant possesses power of converting sugar into starch, and secondly obtain evidence that the starch which is found in the green leaf as the result of the photosynthetic process may pass away in some soluble form, such as sugar, from the leaf to other parts of the plant.

180 To demonstrate that a plant is capable of converting sugar into starch we keep a plant, *Elodea*, in the dark till all starch has disappeared from its leaves. Then, having prepared a 3% solution of glucose, cut off several of the starch-free leaves, float them on the surface of the sugar solution in saucers and keep them in the dark. After a day or two, apply iodine test, and observe that starch has made its appearance in the leaves.

181.* By treating sections of the leaves with iodine

and examining microscopically, we demonstrate that the starch grains occur in close association with the chloroplasts

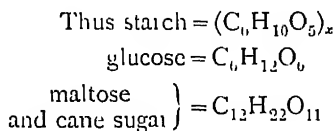
Thus we learn (1) that a plant has the power of converting sugar into starch, and (2) that, in the green leaf, this conversion is effected by the chloroplasts. A chloroplast, therefore, appears to have the power of manufacturing starch in two ways—either by photosynthesis or by acting on sugar. The former mode of manufacture occurs only in light, the latter may occur in darkness. The evidence in favour of the view that the starch which accumulates in the leaves of plants exposed to sunlight may be translocated in a soluble form, we obtain thus—

182 Cut off three similar leaves from a plant, *e.g.* *Spartmannia*, which has been exposed for some hours to a good light. Put two of the leaves with their leaf-stalks in water in the dark, and, at the same time, place the plant itself in the dark. Test the remaining detached leaf for starch. After twenty-four hours, cut off one or more of the leaves of the dark-kept plant, test them and also one of the detached, dark-kept leaves for starch. Repeat the experiment after forty-eight hours, using the remaining, dark-kept, detached leaf and one or more leaves which have remained attached to the dark-kept plant. Observe that the starch disappears more rapidly from the leaves which remained attached to the plant during its exposure to darkness than from the detached, dark-kept leaves. Similarly, demonstrate by using starch-free leaves of a dark-kept plant that starch may accumulate in a detached leaf (with its petiole in water) brought into the light just as in a similar leaf which is attached to the plant and exposed to light. The latter part of the experiment demonstrates that the activities of the detached leaf have not been impaired by its removal from the plant, and hence the only explanation of the persistence of starch in the detached leaf in the former part of the experiment is that carbohydrate cannot escape from the detached leaf. We conclude, therefore, that, in the normal plant, the carbohydrate produced photosynthetically in the leaf, passes away in the form of sugar to supply the needs of the whole plant.

Now, when we consider the extraordinary rapidity with

which the leaf photosynthesises carbohydrate when exposed to light (Exp. 175), and also the narrowness of the stalk, some of the tissues of which, e.g. vessels and tracheids of the wood, are concerned with other work, we can understand that the rate at which a leaf manufactures carbohydrate may be greater than the rate at which the product of this manufacture can pass from the leaf to the other parts of the plant. This being the case, there is a glut of carbohydrate in the leaf. Such an accumulation would undoubtedly—unless it could be got out of the way—pede the action of the machinery involved in the manufacturing work. Thus the idea suggests itself that the machinery must either cease working, or that the excess photosynthesised carbohydrate must be stored temporarily in the leaf in some form in which it does not interfere with the working of the machinery. Now, an insoluble, infusible, solid substance is less in the way than a soluble, diffusible substance. Starch belongs to the former and sugar to the latter type. Hence we may suppose that, if sugar is the first product of photosynthesis, and if it is produced by the green leaf faster than it can be conducted away from the leaf, the excess may be stored in the form of starch. The chlorophyll machinery must perforce lie idle during the night, so that whereas the manufacturing process can go on only during the hours of daylight, the work of distribution can be carried on night and day. The hours of daylight are too brief, as it is, for the green leaf to manufacture enough carbohydrate to satisfy the needs of the growing plant; supply it with food-material and to allow it to accumulate large stores of starch in reserve organs, seeds, etc. It would be bad business, therefore, for the photosynthetic process to be shut down merely because the product cannot be distributed fast enough; it would be good business if the leaf could store its excess of manufactured material temporarily, so that the chlorophyll machinery might run full time. Such considerations suggest to us that the starch which appears in the leaf is not the direct product of photosynthesis, but is—as it is in seeds, etc.—a reserve form, in which the excess of photosynthesised carbohydrate is stored temporarily.

Let us ascertain if the facts support this hypothesis. In the first place, we have proved already that sugar as well as starch makes its appearance in the leaves when a dark-kept plant is exposed to light (Exp 181). Hence sugar has at least as much claim as starch to be regarded as the immediate product of photosynthetic activity. In the second place, starch is, as we know, a more complex body than are sugars —



and it would seem more likely that the less complex body is produced first.

183 Test for starch and sugar the leaves of such plants as iris, onion, *Scilla*, which, after having been kept for a few days in the dark, e. g. by covering with an inverted flower-pot, have been exposed to the light for some hours. Observe that, whereas sugar is present, the leaves contain little or no starch. We thus learn that, even in leaves in which starch is formed as a result of photosynthesis, sugars are also present, and that, in other plants, the products of photosynthetic activity are sugars and not starch.

We may conclude, therefore, that some form or other of sugar is the immediate, carbohydrate product of photosynthesis. Of this sugar, some may be used for local nutritive purposes, some may be used locally for respiration, but the bulk is translocated from the leaves to other parts of the plant. Of the sugar thus translocated, part is used immediately for nutritive or respiratory purposes, and part serves one or other of these purposes ultimately, but undergoes, in the meantime, a change into the reserve form, starch. So, too, the excess of sugar produced over that translocated accumulates in the leaf during the day, and is converted into the more convenient storage form of starch, which, however, is changed by diastase back again into sugar as

soon as the road is clear for the passage of this substance. Thus the green leaf is, during the day, the seat of carbohydrate synthesis, and thus, night and day, there passes from the green cells to all parts of the plant a constant osmotic stream of sugar. Just as the excess of production over local consumption and over translocation leads to an accumulation of starch in the leaves, so an excess of supply over consumption in any part leads to like storage in the form of starch, and thus it is possible for reserve-organ to assemble large quantities of this substance.

But one point still awaits explanation, viz. —How is the conversion of sugar into starch effected?

We know that the starch in the leaf makes its appearance in the form of grains attached to, or imbedded in, the chloroplasts. We know, on the other hand, that reserve organs, underground tubers, etc., contain no chlorophyll grains. How then do the cells of such organs contrive to convert sugar into starch? Now, we have learned already that most plants grown in the dark produce leaves which contain no green colouring matter, and we may ask ourselves the question whether such dark-grown plants are able to store starch in their leaves? This question is, of course, capable of being solved experimentally.

184 * In order to do this we raise seedlings of *Phaseolus* or potato in darkness. Without stopping now to consider the peculiar features presented by the dark-grown etiolated plants, we proceed at once with our experiment by testing leaves for starch. Having shown that the leaves contain no starch, cut off several leaves, float them on a 10-20% glucose-solution in the dark. At daily intervals, apply the starch-test to one of each of these leaves, and also to others which have remained attached to the plant in the dark. Having demonstrated that starch is present in the leaves which have floated on sugar-solution and not in the others, cut sections of the former, mount in water, run in iodine solution, and note that the starch-grains are associated each with a yellow brown stained granule. Bring the plants from darkness into a good light. Soon after a distinct greenish color is visible in the leaves, (a matter of an hour or so) cut sections of a leaf of each plant, observe the chloroplast

and run in iodine; note that the starch-grains are attached to the chloroplasts, and hence conclude that the chloroplasts of the green, light-grown plant are represented in the colourless, dark-grown plant by colourless granules, which we may call *leucoplasts*. Evidently the leucoplast is a starch-former, and is able to manufacture starch from sugar, and the chloroplast is a more efficient starch-former, for not only is it able to manufacture starch from sugar, but also to manufacture sugar from inorganic materials.

185¹ By microscopic examination of young tubers of potato, etc., demonstrate that the starch-storing cells contain leucoplasts, and that the starch-grains are each formed by a leucoplast.

186 Expose potato tubers to a good light, and observe that, in course of time, they become green, that is, their leucoplasts develop, like those of an etiolated leaf exposed to light, the green colouring matter, chlorophyll, which permeates the chloroplast as water permeates a sponge.

We conclude, therefore, that specialised protoplasmic structures, leucoplasts, may occur in the colourless cells of a plant, that they have the function of manufacturing reserve carbohydrate (starch) from plastic carbohydrate (sugar), that the chloroplast is a leucoplast in which green colouring matter, chlorophyll, is developed, and that the chlorophyll confers on the chloroplast the power of manufacturing sugar—in the presence of light—from raw inorganic materials.

187⁴ It is interesting to note also that the green chloroplasts may themselves undergo a change in colour, as, for example, in the petals and fruits of certain plants, which become brightly-coloured as they ripen. Microscopic examination of various stages shows that such colours are due to small irregular bodies, chromoplasts (or chromatophores), which are derived from chloroplasts. In other cases, however, bright colours, e.g. the purple of the copper beech, the colours of the petals of the sweet pea, violet, etc., are, as we determine by the microscopic examination of sections of these objects, and as we have seen in the case of the beetroot, due to pigments dissolved in the cell-sap.

We may express our conclusions as to chloroplasts and chromoplasts thus —

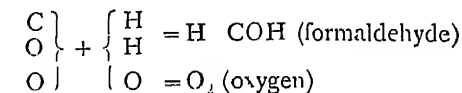
plastids	leucoplasts, chloroplasts, chromoplasts,
leucoplast and chlorophyll	— chloroplast,
chloroplast in which chlorophyll has undergone chemical change and is replaced by a yellow or red pigment	chromoplast (chromoplast)

We must return now to consider more the chemistry of the photosynthesis of sugar, and which light plays in this process. So far, all that is that, during photosynthesis, carbon dioxide is and sugar formed. When we consider the form simple sugar (e.g. glucose) $C_6H_{12}O_6$, it becomes that, in the process of photosynthesis, some substance which serves as the source of hydrogen must, together with carbon dioxide, take part in the reaction. The source of the hydrogen which enters into the sugar molecule is water. But if we assume that the hydrogen of the photosynthesised sugar is derived from water, carbon dioxide and water together to produce sugar must first be a deoxidation or reduction of one of these substances. Assuming this, we may symbolise the reaction thus: $-6CO_2 + 16H_2O \rightarrow C_6H_{12}O_6 + 16O_2$.

If, therefore, our assumptions are correct, we expect to find that, whenever photosynthesis goes on, oxygen is liberated. In order to ascertain whether this is the case, we proceed as follows:

188 Take several green shoots of some common plant, e.g. *Elodea canadensis*, tie them loosely together, with their cut ends upwards, inside an inverted funnel, the mouth of which just fits a large beaker full of water. Fill a test tube with water, and closing it with the thumb, invert it under the stem of the funnel, which projects upward near the level of the water in the beaker. Thus, any gas which is evolved by the submerged shoots passes up through the funnel and collects in the test tube. Bring the apparatus into light, and observe that, at once, bubbles of gas are

from the shoots and rise in the test tube. Allow the experiment to go on until the test tube is nearly full of gas. This may take several days. Having prepared a glowing splinter, *e.g.* by lighting a match and blowing it out, lift the tube from the water, plunge into it the glowing splinter and observe that the latter bursts into a bright flame, whence we infer that the gas in the tube consists largely of oxygen. By more exact methods it may be demonstrated that, as is required by our formula, the volume of carbon dioxide absorbed during photosynthesis is equal to the volume of oxygen evolved. We thus conclude that a step in this process consists in the reduction of carbon dioxide and water, and we may picture this step thus —



On various grounds it is believed that this substance $\text{H} \cdot \text{COH}$ (formaldehyde) is actually the first product of photosynthesis, and that its molecules condense to form sugar thus $6\text{H} \cdot \text{COH} = \text{C}_6\text{H}_{12}\text{O}_6$. In any case, we recognise that in photosynthesis the reaction takes place in several stages, and that the first stage consists in a reduction process, which is made evident by the evolution of oxygen.

Water-plants such as *Elodea* serve also for investigations into the conditions under which photosynthesis takes place.

189 Select a healthy shoot of *Elodea*, make a clean cut across its stem, wipe the cut end gently with blotting-paper, and brush a layer of collodion over it. When the collodion has set, make a small hole in the film by means of a needle, tie the shoot, cut end upward, to a glass rod, and submerge it in a large beaker of water. Using the rate at which bubbles escape from the cut end as an index of photosynthetic activity, determine the influence of light-intensity on the rate at which this process goes on. This may be done by counting and comparing the rates at which bubbles are given off when the apparatus is placed at varying distances from a well-lit window. A better method consists in the use of an incandescent light